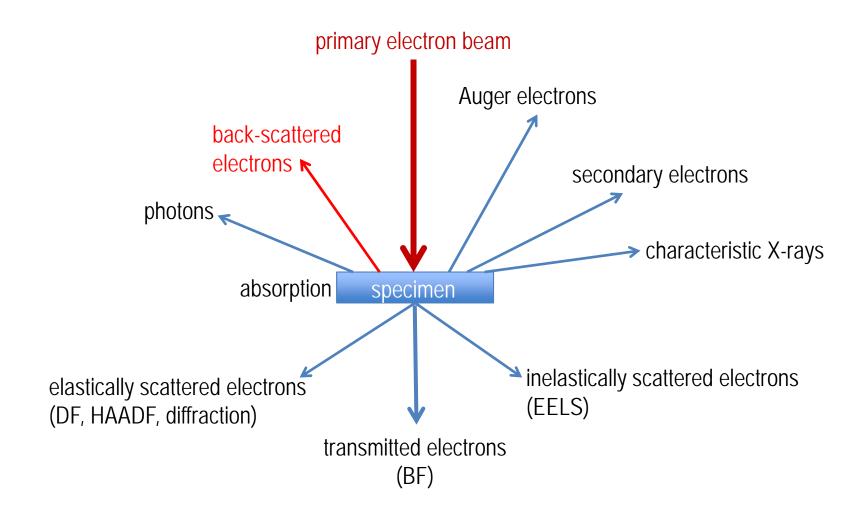
UHV Techniques

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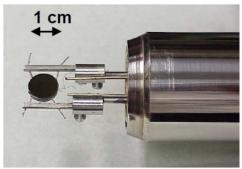
Electrons-matter interaction

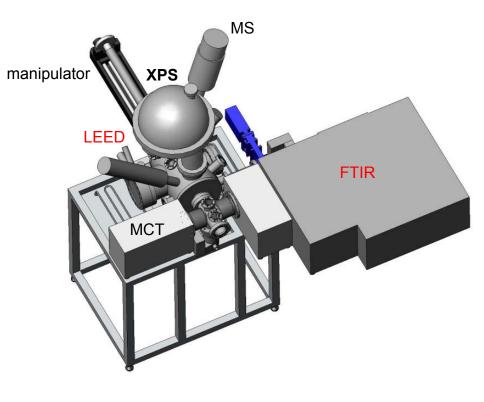


The UHV setup

- Stainless steel UHV setup with flanges, pumps, pressure gauges, ...
- Typically, 10⁻¹⁰ to 10⁻¹¹ mbar base pressure
- Tool and components:
 - preparation
 - characterization
 - sample manipulation
 - resistive heating

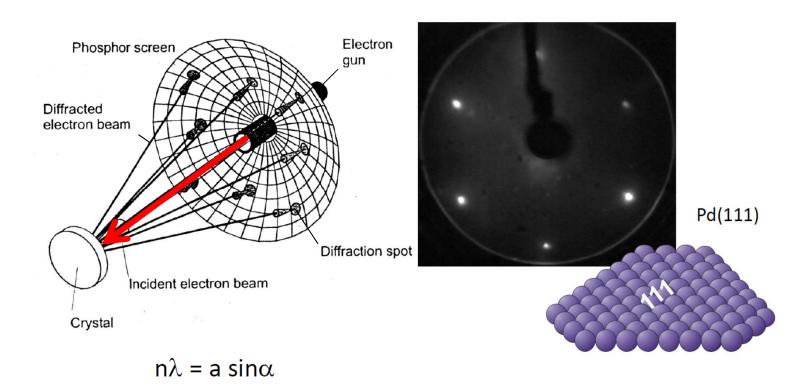






LEED

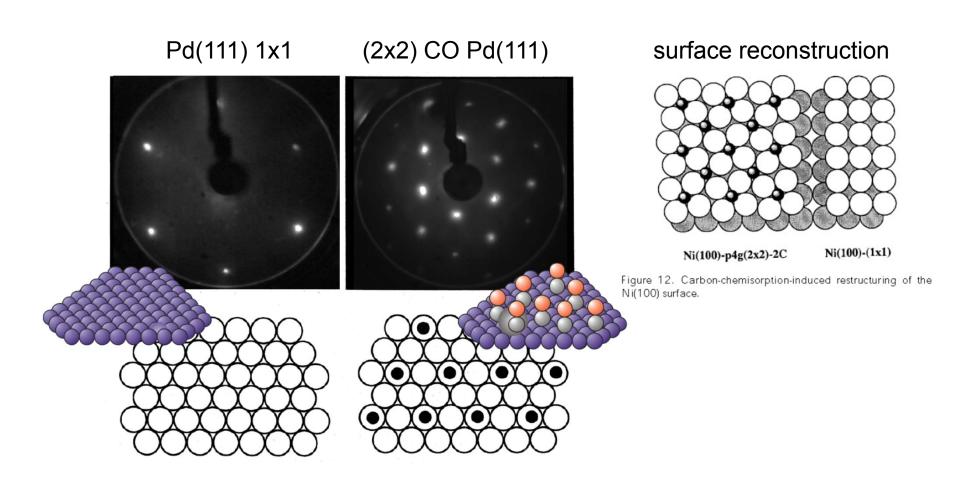
Low-energy electron diffraction



Collimated beam of low energy electrons (20-200 eV) Wavelength (λ) of 100 V electrons: ca. 1 Å, < interatomics distances

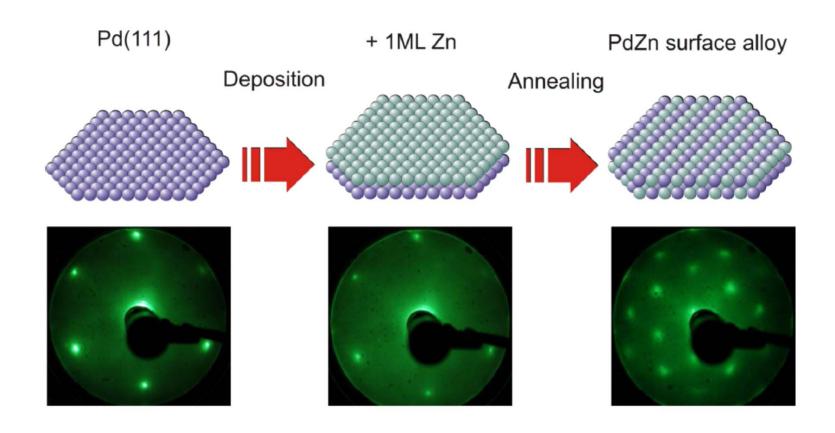
LEED

■ The structure of surfaces



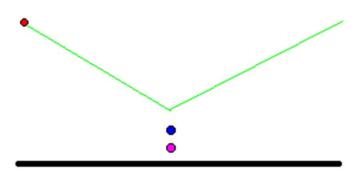
LEED

■ Preparation of PdZn surface alloy for model studies



- High resolution electron energy loss spectroscopy
- Interaction of low energy electrons with surfaces
- Backscattered electrons are detected
- Possible interactions are
 - the excitations of vibrations of adsorbed molecules
 - vibrations of the top layers of the substrate (surface phonons)
 - vibrations of the electrons in the substrate or in films or islands on the substrate surface (plasmons)
- UHV is required
- Surface sensitive technique
- Technique of choice to study adsorbates at single crystal surfaces

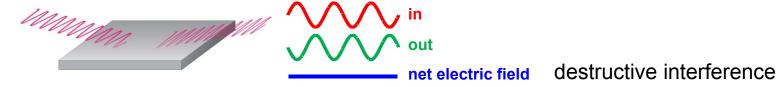
■ How it works



- Some of the electron beam's energy is converted into vibrational motion of the adsorbed molecule
- The result is a characteristic loss peak in the HREEL spectrum

- How it works Interaction between electrons and molecule
- Two scattering modes
- Dipole scattering
 - incident electron is like a electromagnetic wave interacting with oscillating dipoles (vibration of species at surface)
 - long-range effect mediated by the Coulomb field. The incoming electron is influenced by a vibrating dipole at the surface.
 - electron is scattered specularly with an energy loss characteristic of the energy it delivered to the vibrational mode.

- How it works Interaction between electrons and molecule
- Two scattering modes
- Dipole scattering
 - identical information to a IR spectrum
 - same rules as IR spectroscopy
 - only fundamental transitions allowed
 - only vibrations accompanied by change in dipole moment allowed
 - s-polarized light undergoes 180° phase change upon reflection



- surface (normal) selection rule: only dipoles perpendicular (normal) to the surface are active
- intensity is at maximum for specular reflection

The surface selection rule

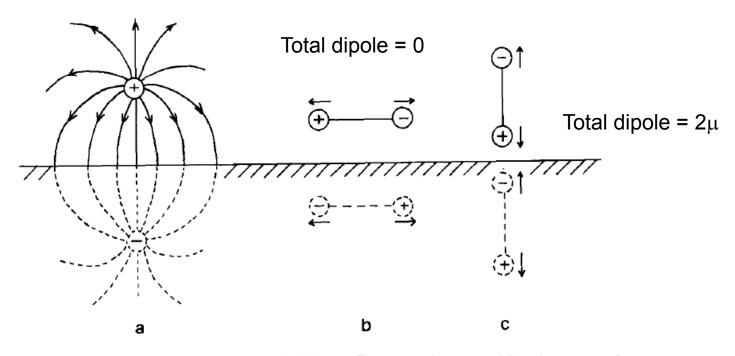
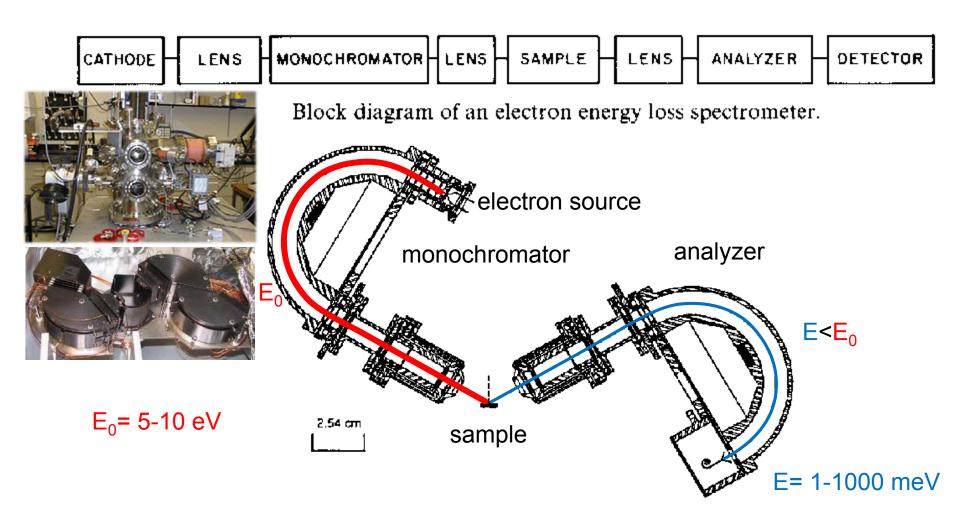


Fig. 1. (a) The lines of force and the electrical "image" resulting from a positive charge over the surface of a conductor (the metal surface is the upper line above the hatched area). (b) The changes during the vibration of a dipole parallel to the surface of the metal; the "image" dipole change is in the opposite direction to the original. (c) The changes during the vibration of a dipole perpendicular to the surface; the "image" dipole change is in the same direction as the original.

- How it works Interaction between electrons and molecule
- Two scattering modes
- Impact Scattering
 - transfer of energy between electron and molecule while the electron is in the molecule
 - short range scattering process from the ion core
 - quantum mechanical formalism required for the theory
 - vanishes in specular direction
 - isotropic (not in the specular direction, but everywhere) but the energy losses still reflect vibrational excitations in the adsorbate
 - dominant at high vibrational energy
 - strong dipole scatterers (CO) are weak impact scatterers

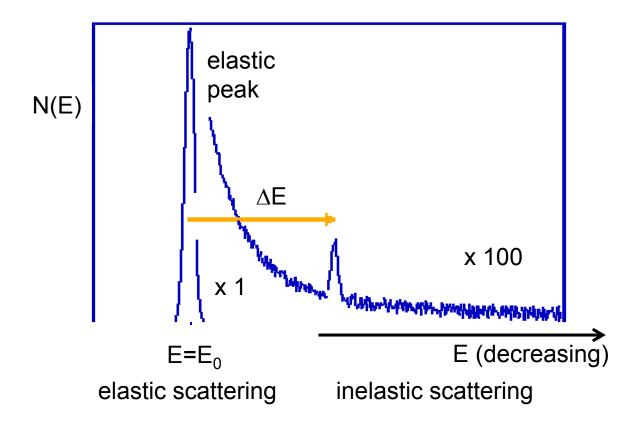
The angular distribution of peaks around the specular direction can distinguish between peaks which result from different scattering modes

Setup



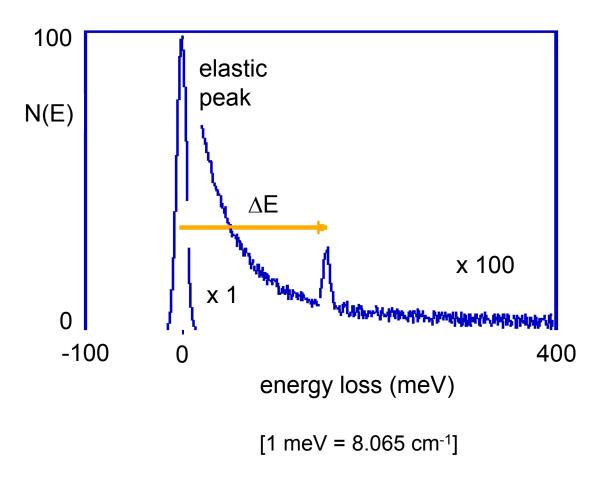
Electrons from a cathode pass through a monochromator, strike the sample, and the energy spectrum of the scattered electrons is probed by a second monochromator.

The HREELS spectrum



The HREELS spectrum

 ΔE = energy of vibrational mode of excited adsorbate upon inelastic scattering



HREELS vs. RAIRS

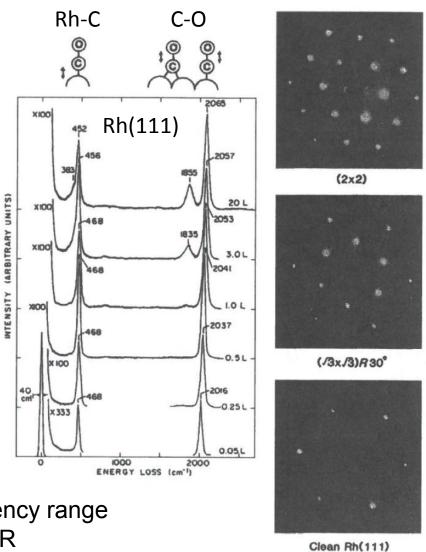
HREELS

- access to low vibrational energy, e.g. of O (few meV)
- low energy resolution (≈ 4 meV;
 <30 cm⁻¹)
- 0.0001 monolayer of CO
- specialised setup (UHV)

RAIRS

- detectors limited to 400 cm⁻¹
- high energy resolution (4 cm⁻¹)
- 0.01 monolayer of CO
- also ambient pressure

Examples

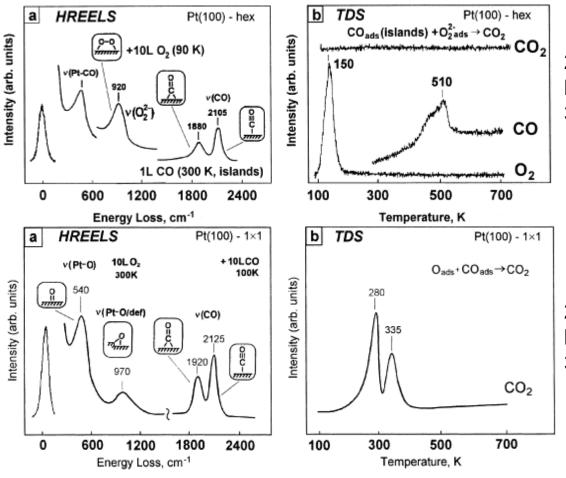


surface structure is known at all points of the experiment (LEED)

enhanced frequency range compared to FTIR

Examples

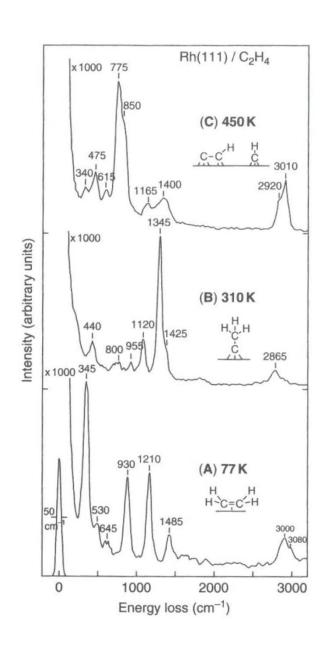
Access to low vibrational energy



- 1. CO adsorption at 300 K
- 2. Pt(100)-(hex) surface covered by CO/(1x1) islands
- 3. O₂ adsorption at 90 K

- 1. O₂ adsorption at 300 K
- 2. Pt(100)-(1x1) surface covered by atomic O layer
- 3. CO adsorption at 100 K

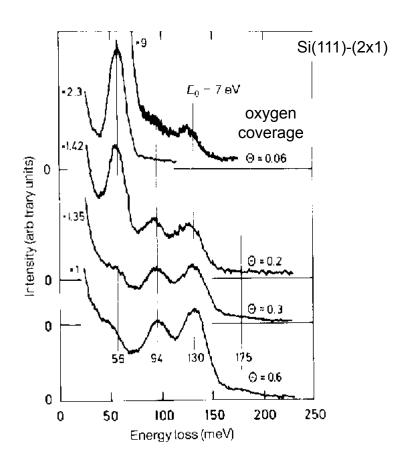
HREELS Examples



Examples

HREELS of surfaces

■ Phonons: elementary vibrational motion in which a lattice of atoms or molecules uniformly oscillates at a single frequency

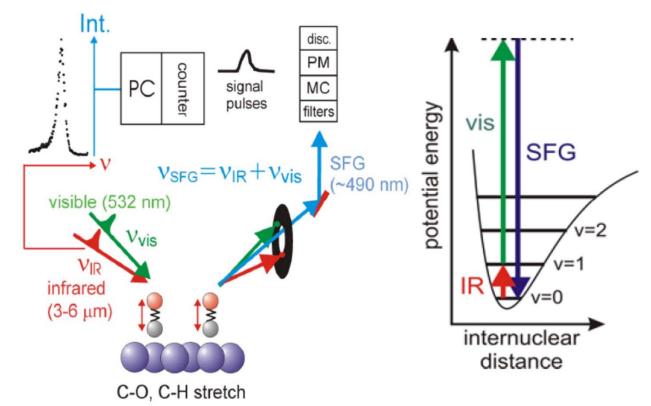


Sum frequency generation

- Two laser beams mix at a surface and generate an output beam with a frequency equal to the sum of the two input frequencies
- Advantages
 - ability to be monolayer surface sensitive
 - ability to be performed in situ (for example aqueous surfaces and in gases)
 - does not cause much sample damage
- SFG gives complementary information to FTIR and Raman spectroscopy

■ IR-vis SFG laser spectroscopy

- no signal from isotropic gas phase
- no signal from centrosymmetric solid
- only adsorbates! (UHV to 1 bar)

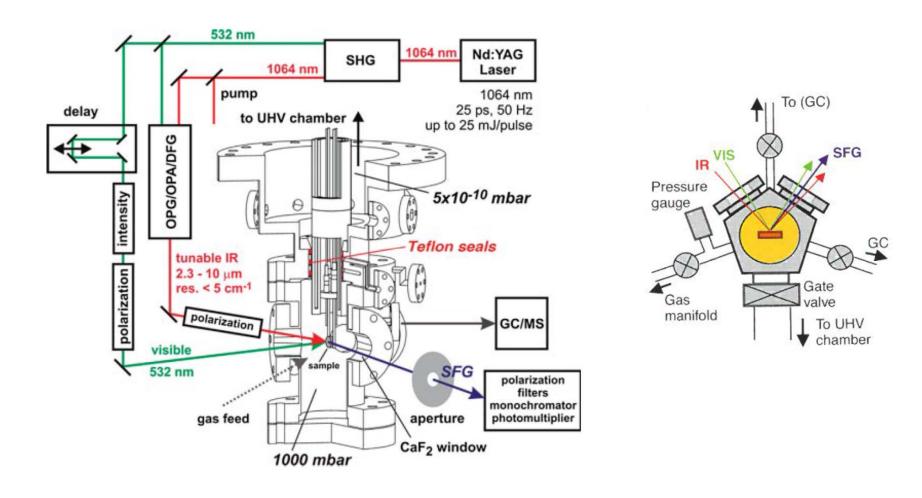


■ laser required: non-linear process is weak

Information

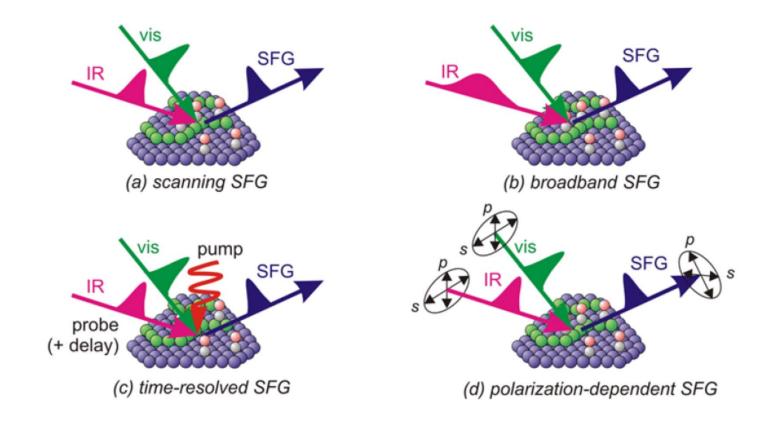
- composition
- orientation distributions
- structural information of molecules (vibrational spectroscopy)
- gas-solid, gas-liquid, liquid-solid
- selection rule: in order to generate a sum frequency emission, the excited vibrational mode must be both IR and Raman active

Setup

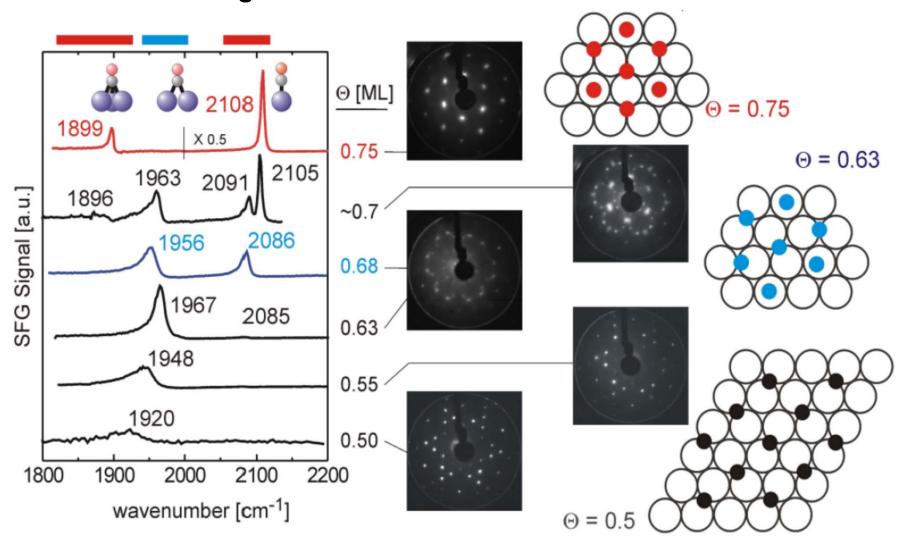


Föttinger et al., Characterization of Solid Materials and Heterogeneous Catalysts, 2012, Wiley-VCH Verlag

■ Modes of operation

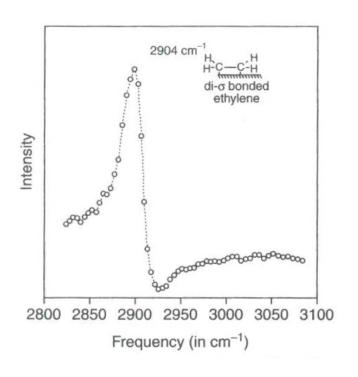


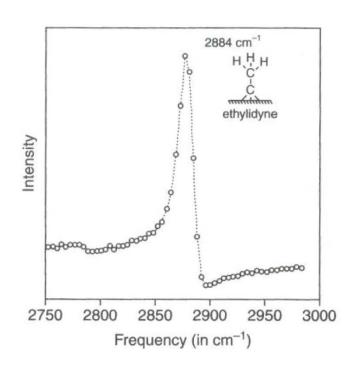
Structure assignment in combination with LEED



Rupprechter, PCCP 3 (2001) 4621; Unterhalt et al., J. Phys. Chem. B 106 (2002) 356

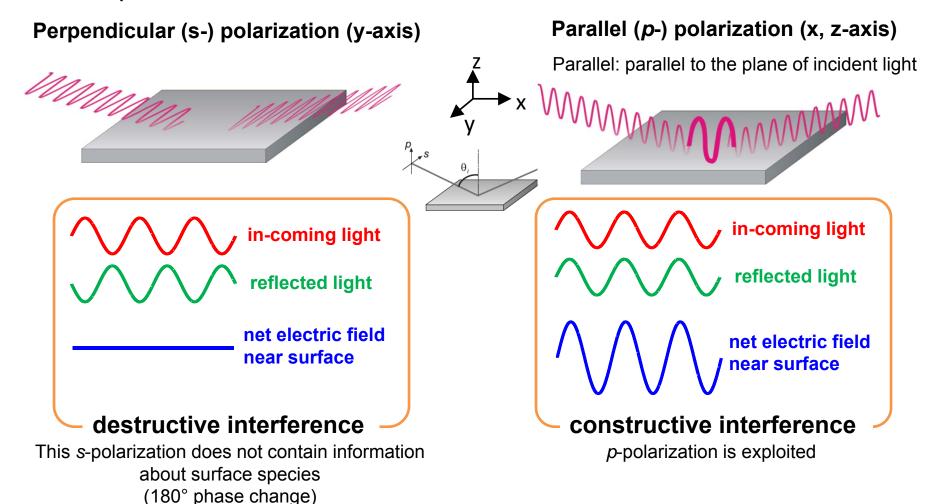
Hydrocarbon fragments on Pt(111)





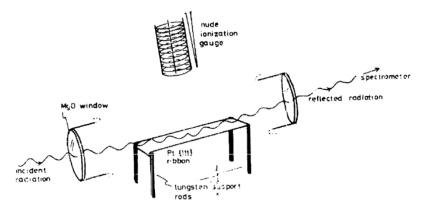
RAIRS

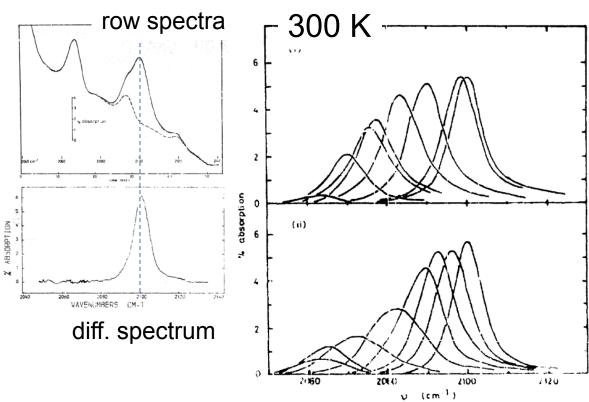
- Reflection absorption IR spectroscopy
 - Specular/external reflection method

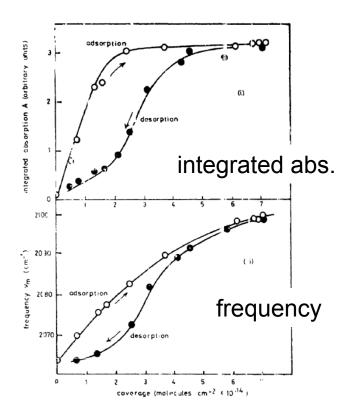


CO adsorption on Pt(111)

- v(CO) blue-shifts with increasing coverage
 dipole-dipole interactions
- linear increase of intensity up to 1/3 of full surface coverage, then only shift
- ads/des hysteresis



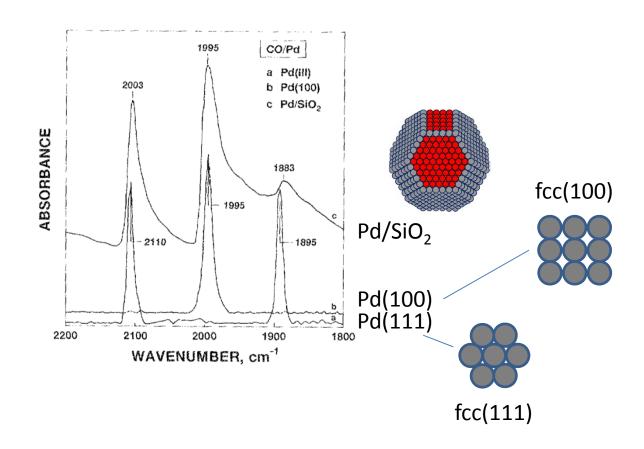




Shigeishi et al., Surf. Sci. 58 (1976) 379

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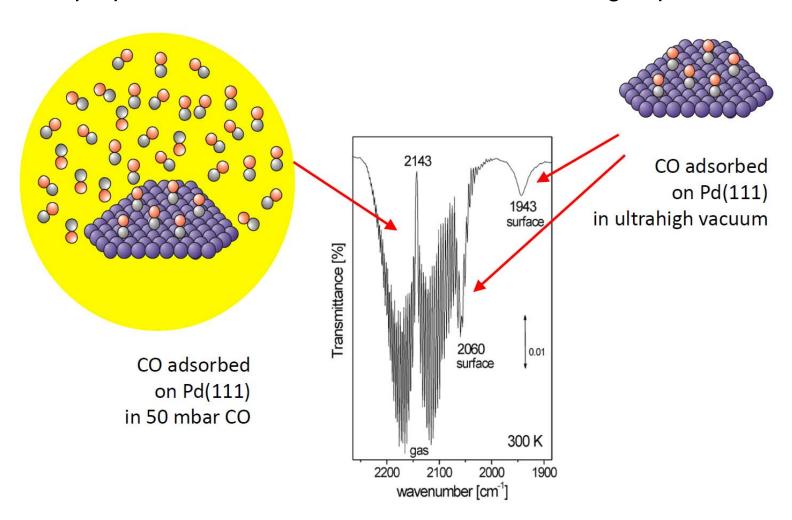
- Single crystal vs. Powder
 - reference for assignment of signals on technical catalysts!



RAIRS

■ Non-UHV conditions

Superposition of contributions from surface and gas phase

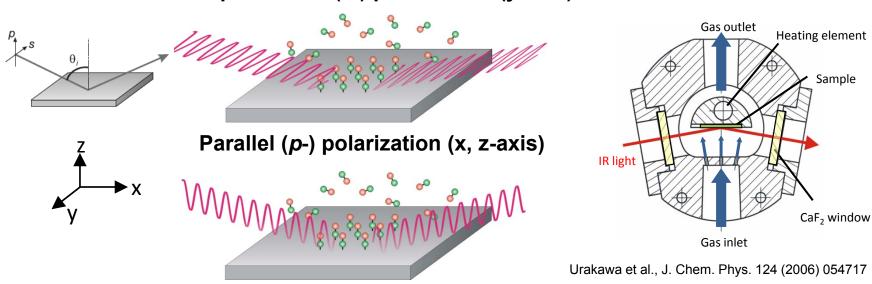


Phase-modulation IRRAS (PM-IRRAS)

Non-UHV experiments

- continuous generation of 2 polarizations (photoelastic modulator)
- excellent gas-phase compensation
- highly sensitive, time-resolved

Perpendicular (s-) polarization (y-axis)



 R_p - R_s = ΔR Parallel polarization Perpendicular polarization Surface + gas gas surface

The surface spectra are often shown in $\Delta R/R$ (R=R_s+R_D)

RAIRS

Isolation of surface contribution

