### An enquiry into theoretical bioinorganic chemistry:

## How heuristic is the character of present-day quantum chemical methods?

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### An enquiry into theoretical bioinorganic chemistry:

# How heuristic is the character of present-day quantum chemical methods?

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Laboratory of Physical Chemistry, ETH Zurich, Switzerland Paper Intended to be provocative in order to stimulate discussion on

Swiss Federal Institute of Technology Zurich developments of theoretical approaches

### Predictive power of quantum chemical calculations on active sites?

**DFT**: functionals parametrised on data of test molecule set; statistical analysis of energy errors known for test set; not (never?) systematically improvable

Can one expect that energies of crucial steps in metalloenzyme catalysis are reliably reproduced?

Or properties (=energy derivatives)?

WFT: if accurate —> expensive (often unfeasible); hence, structures often taken from DFT; in principle, systematically improvable total energies,

but does this consistently improve the relative energies?

What to do if not much is know about a new system?

# Are quantum chemical calculations on active sites most useful only *a posteriori* to rationalize experiment?

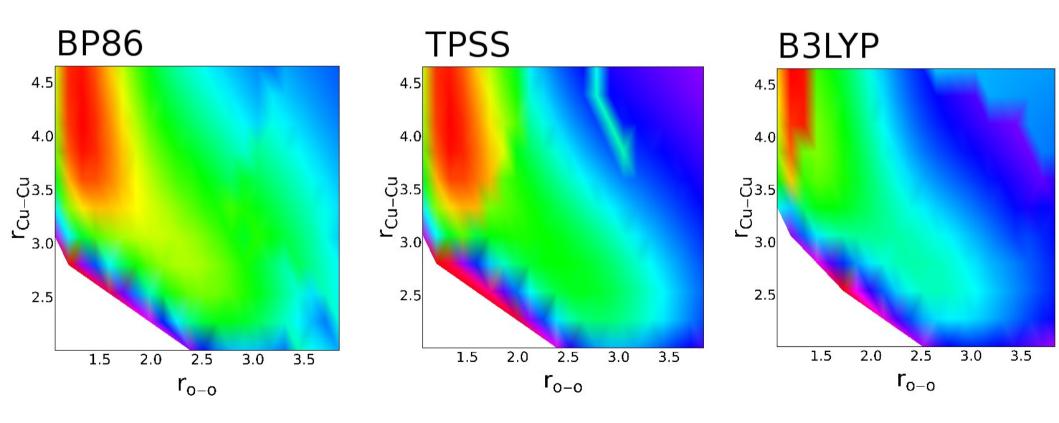
Danger to pick the most consistent result though this might be:

"the right answer for the wrong reason"

Example: DFT spin state splitting energies seem OK, but spin density distributions not the same

## Thesis 1: Method-inherent approximations difficult to control

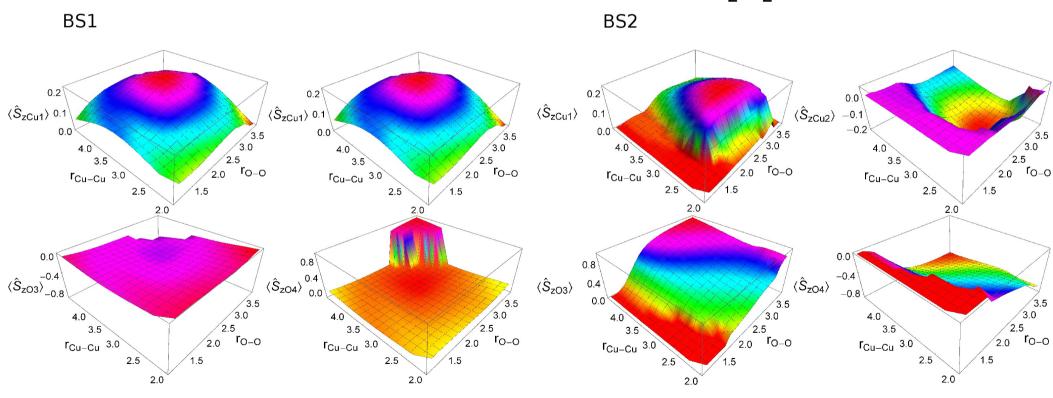
Correlation diagrams for a better "look and feel" of results; **example:**  $Cu_2O_2^{2+}$  bis- $\mu$ -oxo vs. side-on peroxo; triplet state



### Correlation diagrams provide insight

- ... into flexibility of a structure (potential energy well depth/width),
- ... into emergence of properties etc.

Example: two broken-symmetry solutions for Cu<sub>2</sub>O<sub>2</sub><sup>2+</sup>



### Thesis 2: Structural models of active sites can be difficult to construct

... because of: system size (active center+protein environment), solvation, dynamics

"convergence" depends on system under study

many examples known from the literature

### Example: De novo protein design

Pre-requisite: In silico design and re-engineering of enzymes require reliable prediction of energy barrier heights!

Kinetics depend exponentially on activation energy

-> Small energy changes have dramatic effects on catalysis.

Apart from method-inherent errors,

structural models can produce these energy modulations.

## Transition state flexibility in [Fe] hydrogenase

Distort transition-state model along normal coordinates (protein environment and water neglected)

Allowed increase in energy: 10 kJ/mol

Mode 2 Mode 3 68 cm<sup>-1</sup> 79 cm<sup>-1</sup>

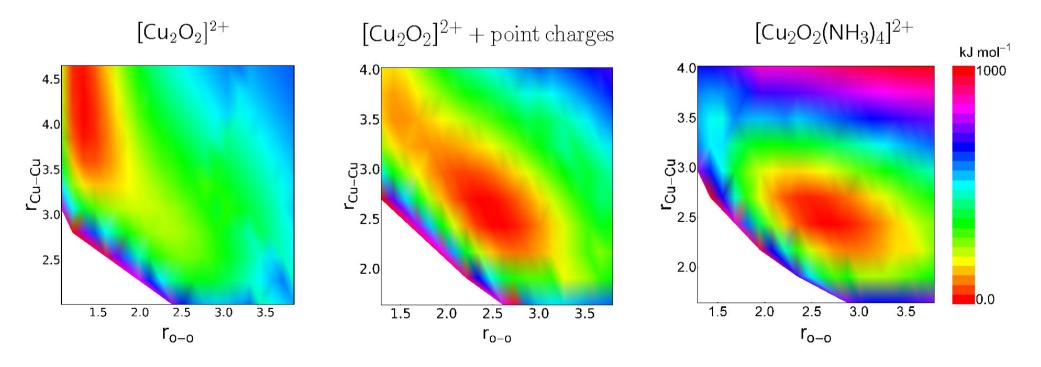
blue: transition state, red: distorted transition state

#### Thesis 3: Transferable results?

How to compare many individual results obtained for

- (a) different model structures
  - (even result for a small model that does not fit to exp. may yield insight into chemistry of active site)
- (b) different/modulated active sites performing similar chemistry

Study correlation diagrams? — like in the "old days" but quantitative triplet



#### Conclusions

Can we expect major breakthrough yielding a universal quantum chemical method with reliable error estimates?

If not: How to cope with the method insufficiencies?

Standard protocols to uncover these insufficiencies?

How to relate the data obtained?

Use/invent concepts that relate calculated data like

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correlation diagrams (cuts through potential energy hypersurface, property surfaces, ...)
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concept of transition state flexibility

(for de novo design/re-engineering of active sites, design of biomimetic catalysts, ...)