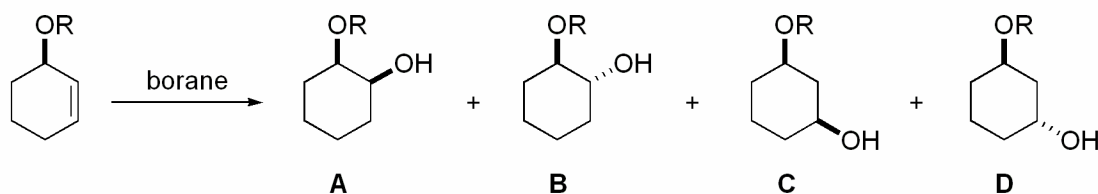
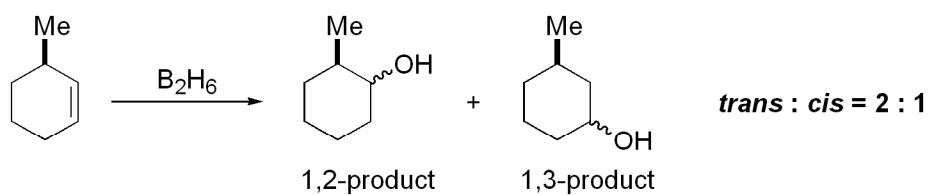


The Hydroboration Reaction

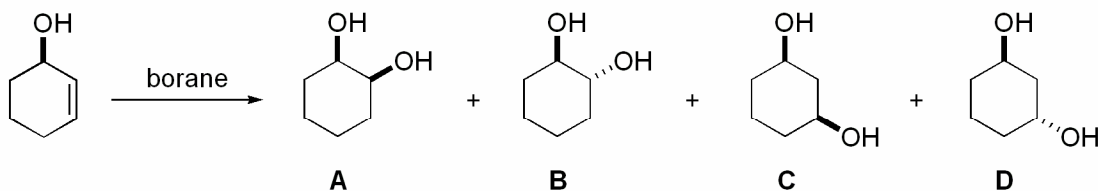
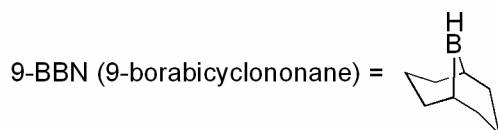
1. Regioselectivity



R	borane	ratio			
		A	B	C	D
Me	B_2H_6	10	81	9 (combined)	
Bn	9-BBN	0	68	19	13
TBS	9-BBN	0	74	13	13

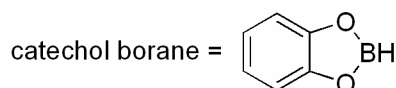
regioselectivity due to inductive effect

Both larger protecting groups and larger boranes give higher facial selectivity, but lead to reduced regioselectivity.



borane	ratio			
	A	B	C	D
B_2H_6	6	86	8 (combined)	
catechol borane, $(Ph_3P)_3RhCl$	1	18	9	72

inductive effect



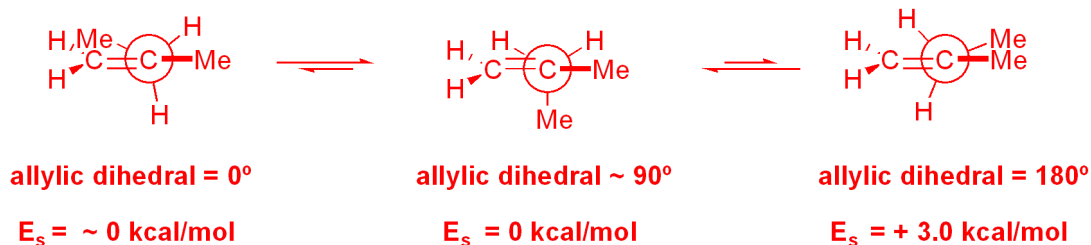
J. Am. Chem. Soc. **1988**, *110*, 6917.
J. Am. Chem. Soc. **1968**, *90*, 4445.
Tetrahedron **1972**, *28*, 1009.

Metal mediated and metal-free processes have complementary regioselectivities.

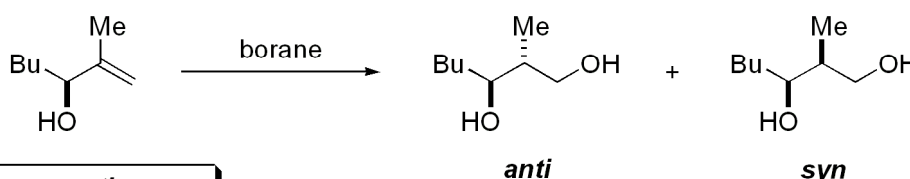
2. Substrate Stereocontrol

2.1. Allylic Stereocontrol

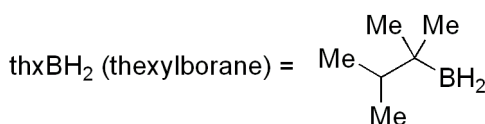
1,2-allylic stereocontrol:



hydroboration: a classic in acyclic stereocontrol



borane	<i>anti</i> : <i>syn</i>
B_2H_6	1 : 1.4
thxBH ₂	3.5 : 1
Cy ₂ BH	8.5 : 1
9-BBN	11 : 1

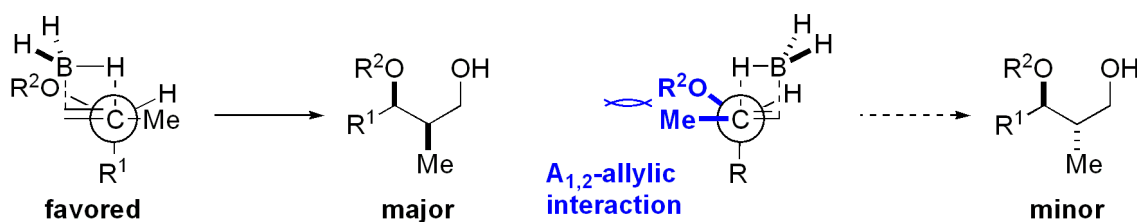


Cy₂BH = dicyclohexyl borane

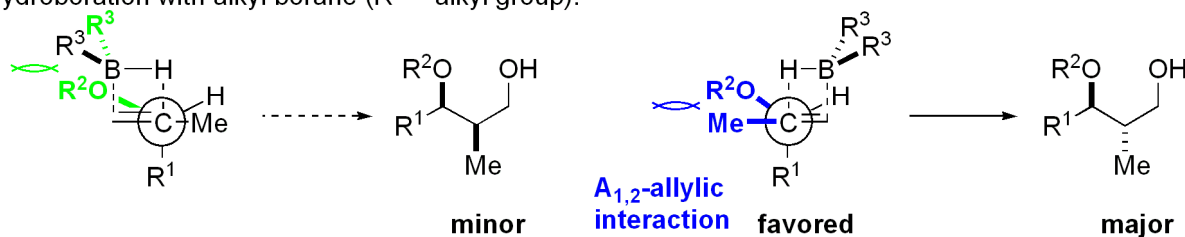
Still *J. Am. Chem. Soc.* **1983**, *105*, 2487.

explanation for difference between BH₃ and alkyl boranes:

for hydroboration with BH₃:



for hydroboration with alkyl borane (R^3 = alkyl group):

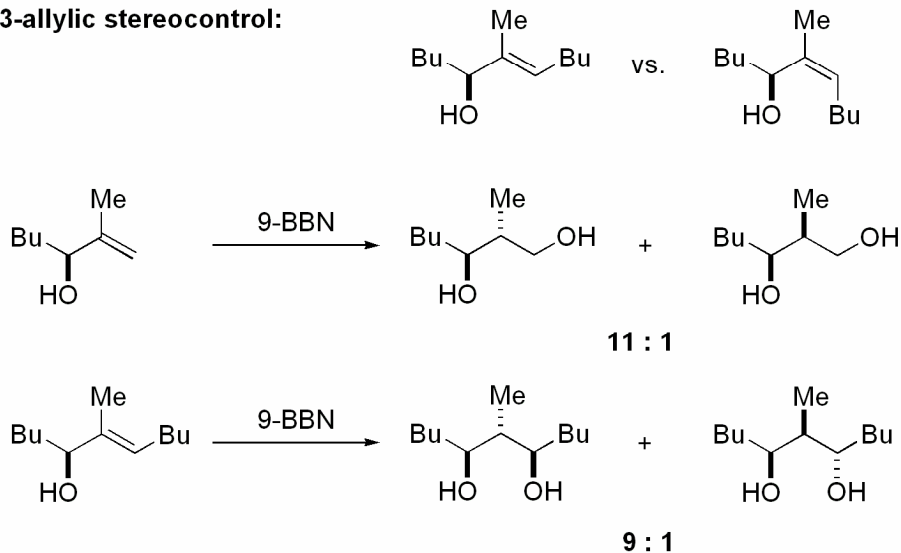


Interaction between R^3 and R^2O overrides $A_{1,2}$ -allylic interaction between R^2O and Me. Consistent with this interpretation is the increase in selectivity with increasing size of R^3 on borane.

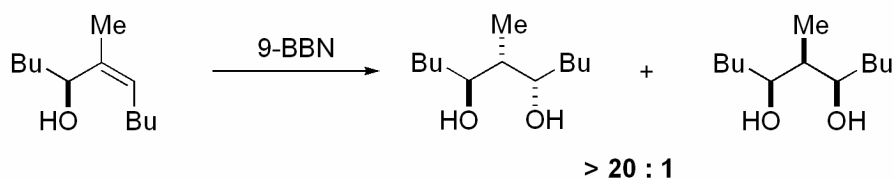
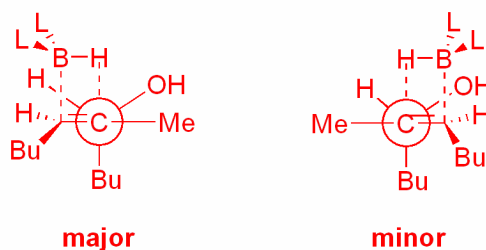
key point is interaction between substrate and reagent, see: Still *J. Am. Chem. Soc.* **1983**, *105*, 2487.

In addition to a careful conformational analysis of the substrates, interactions between reagent and substrate have to be taken into consideration.

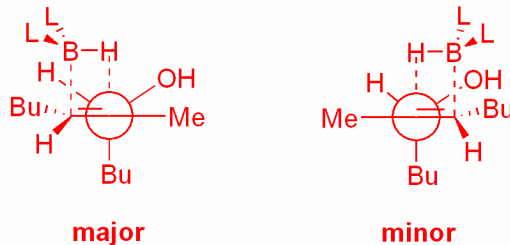
1,3-allylic stereocontrol:



transition states:



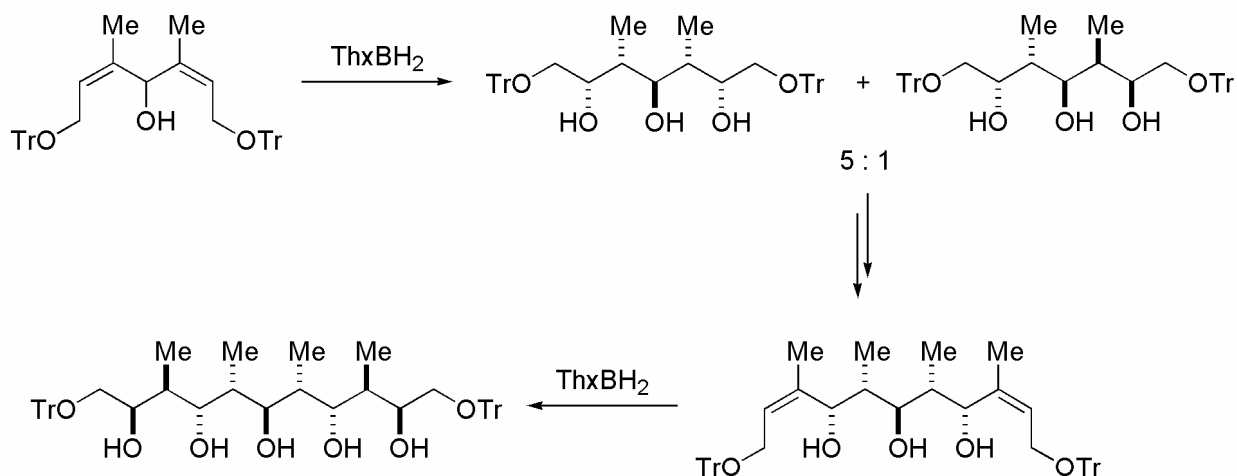
transition states:



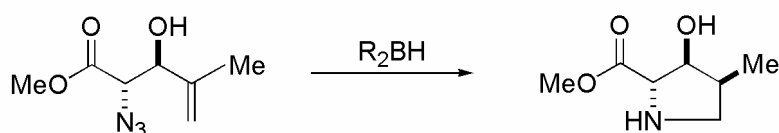
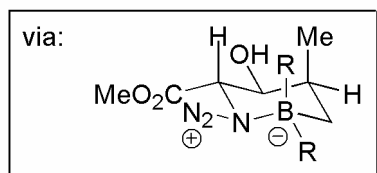
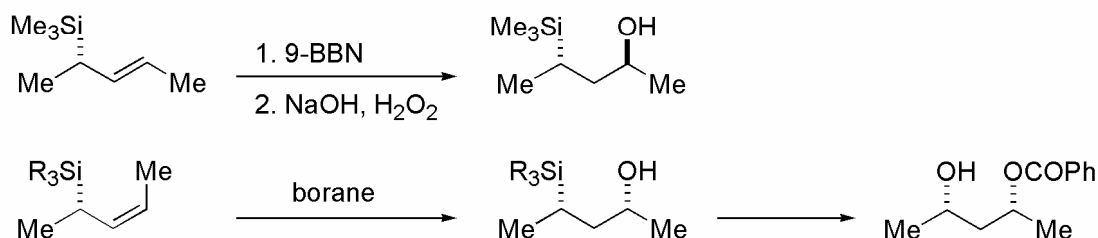
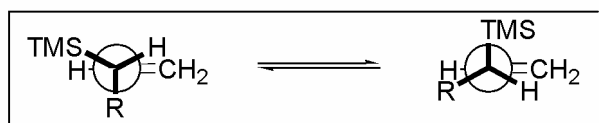
Still *J. Am. Chem. Soc.* **1983**, *105*, 2487.

applications:

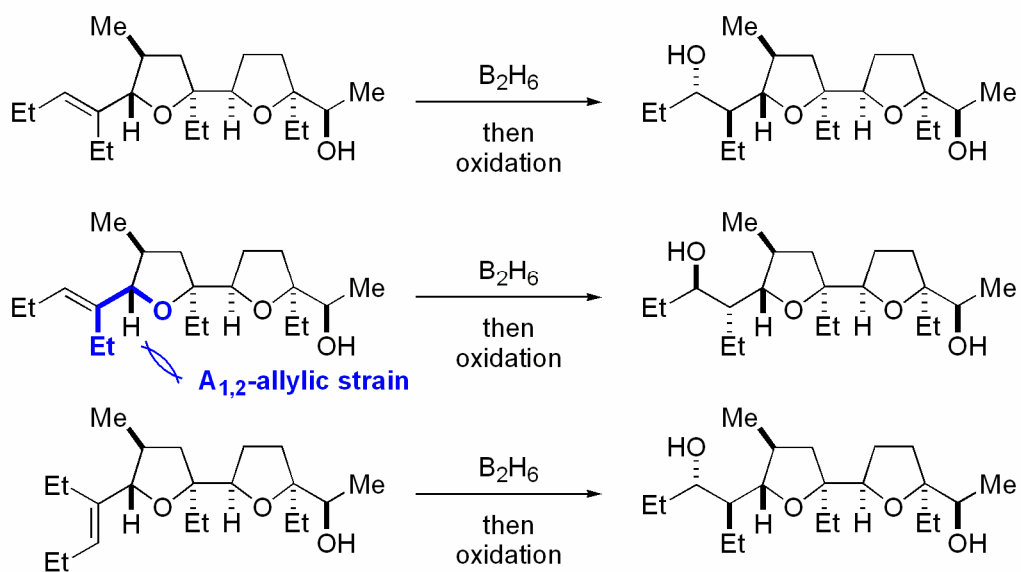
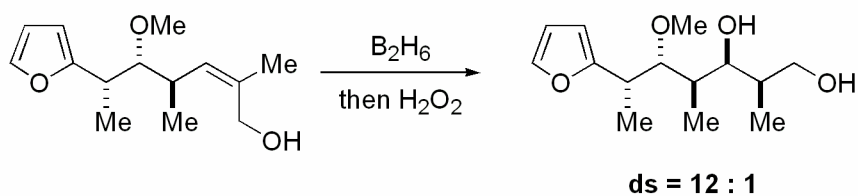
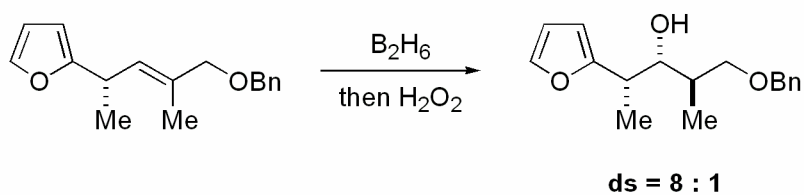
rifamycin synthesis:

*J. Am. Chem. Soc.* **1983**, *105*, 2487.

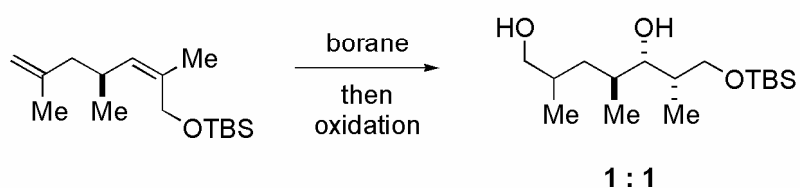
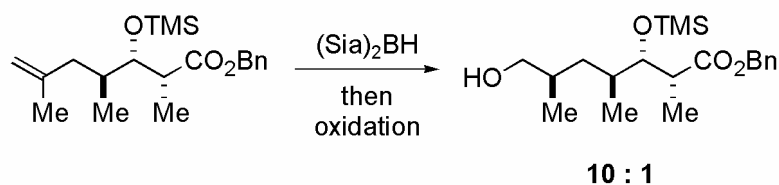
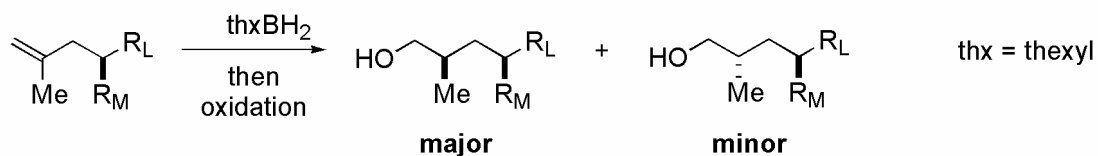
echinocandin synthesis:

**single isomer***Evans J. Am. Chem. Soc.* **1987**, *109*, 7151.**stereoelectronic effects:***Fleming Tetrahedron Lett.* **1988**, *29*, 2077.

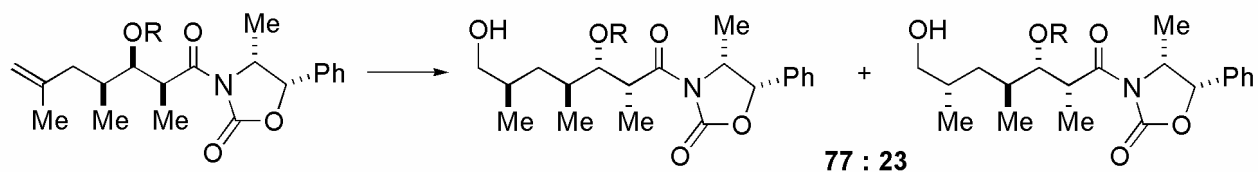
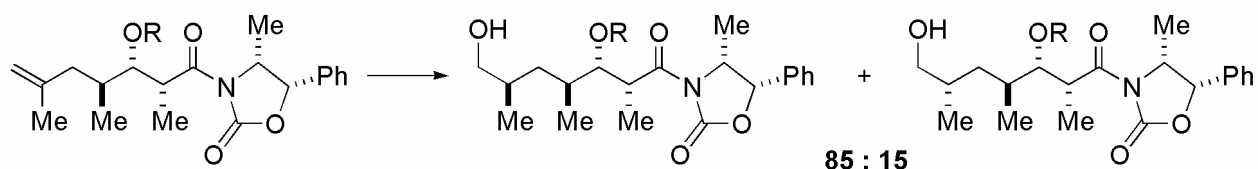
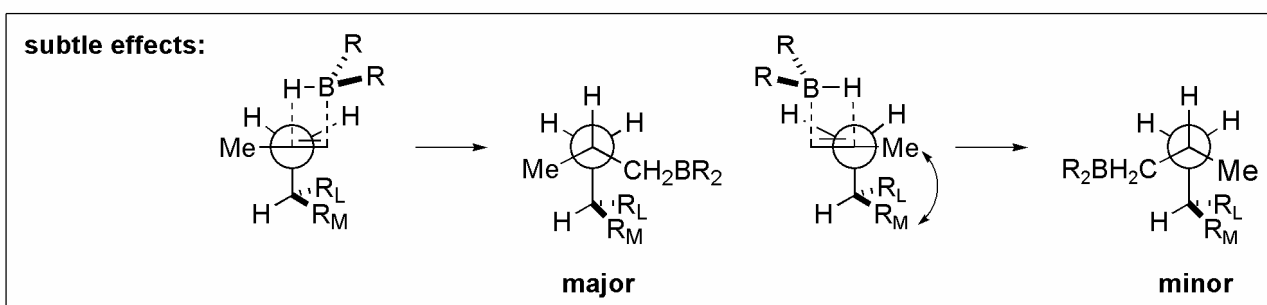
monensin synthesis:

*J. Am. Chem. Soc.* **1978**, *100*, 2933.*Kishi J. Am. Chem. Soc.* **1979**, *101*, 259.

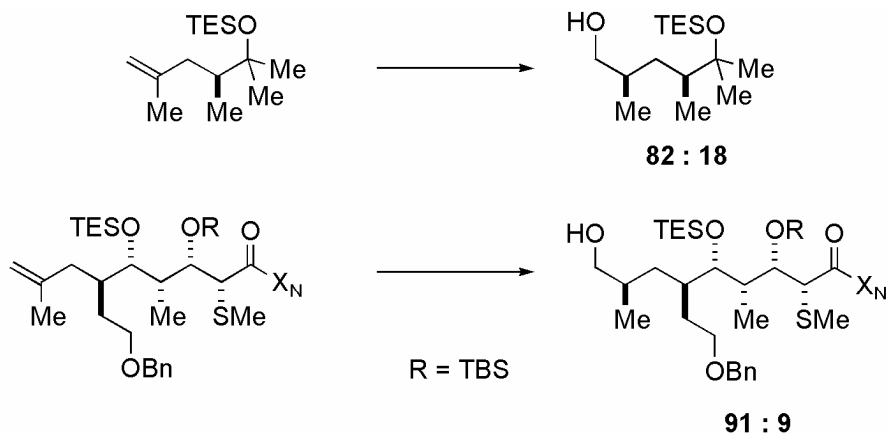
2.2. Homoallylic Stereocontrol



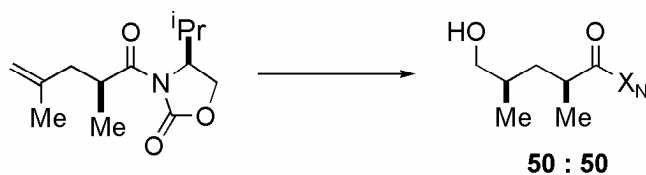
Evans *Tetrahedron Lett.* **1982**, *23*, 4577.



Same sense of induction observed, as reaction is directed by proximal chiral center.

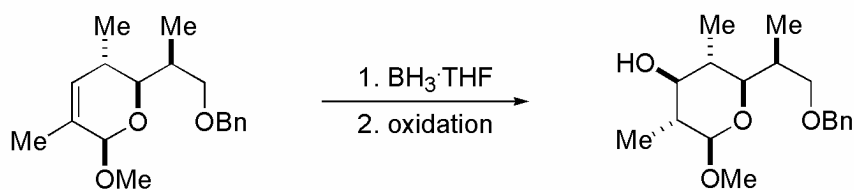


consider however:

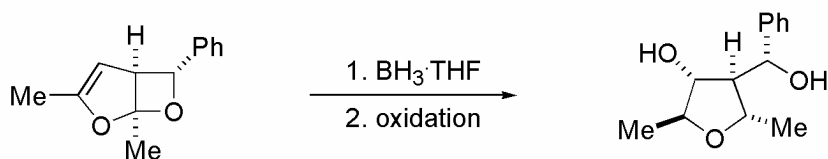


2.3. Cyclic Stereocontrol

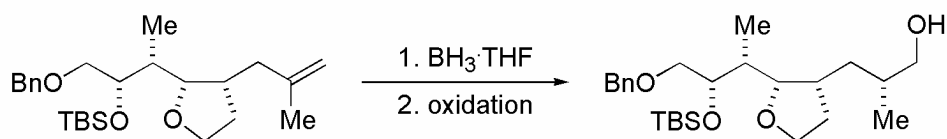
examples:



Danishefsky *J. Am. Chem. Soc.* **1987**, *109*, 862.

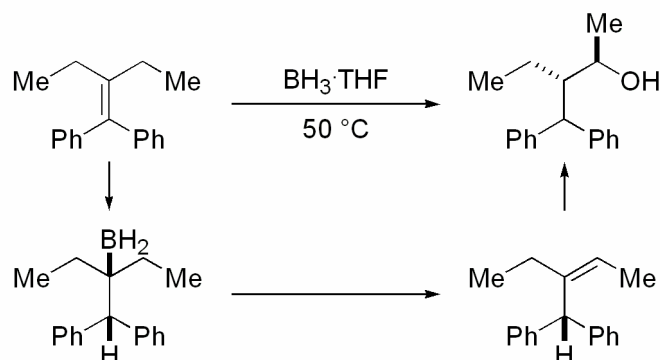


Schreiber *J. Am. Chem. Soc.* **1983** *105*, 660.



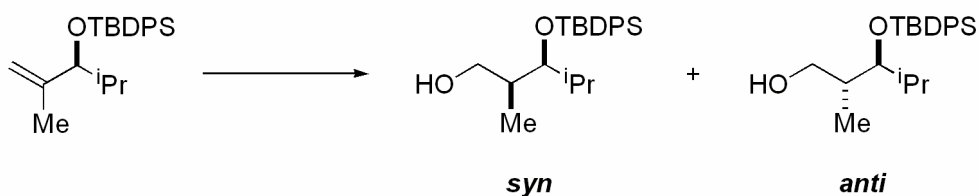
Nicolaou *J. Am. Chem. Soc.* **1982**, *104*, 2028.

2.4. Reversibility of the Hydroboration Reaction



J. Am. Chem. Soc. **2000**, *122*, 10218.

2.5. Rhodium-Catalyzed Hydroboration



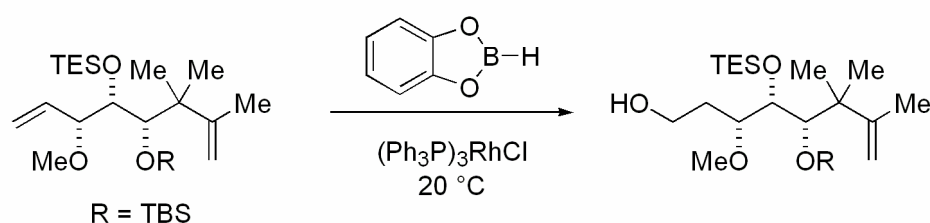
reagent	<i>syn</i> : <i>anti</i>
9-BBN	5 : 95
catechol borane, Wilkinson's catalyst	97 : 3

Evans *J. Am. Chem. Soc.* **1991**, *113*, 4042.
J. Am. Chem. Soc. **1992**, *114*, 6679.
J. Org. Chem. **1992**, *57*, 504.
Chem. Rev. **1991**, *91*, 1179.
Tetrahedron **1997**, *53*, 4957.

important features:

1. opposite sense of induction compared to uncatalyzed hydroboration
2. high selectivity for monosubstituted olefins

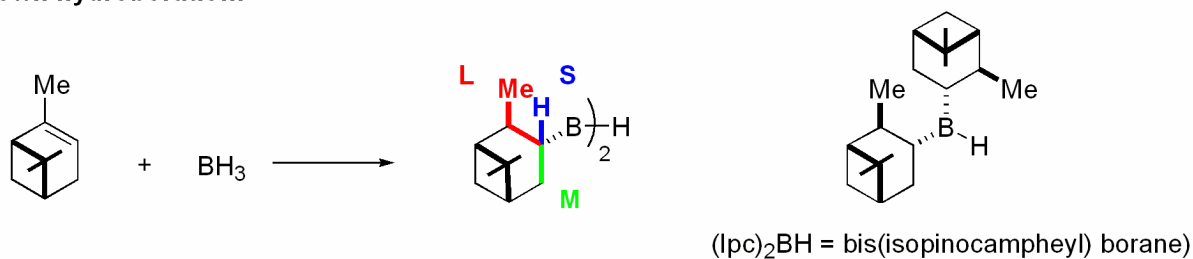
calyculin synthesis:



Evans *J. Am. Chem. Soc.* **1992**, *114*, 9434.
J. Org. Chem. **1992**, *57*, 1961.
J. Org. Chem. **1992**, *57*, 1964.

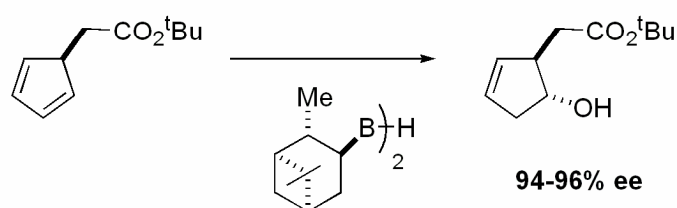
3. Enantioselective Hydroborations

Brown hydroboration:

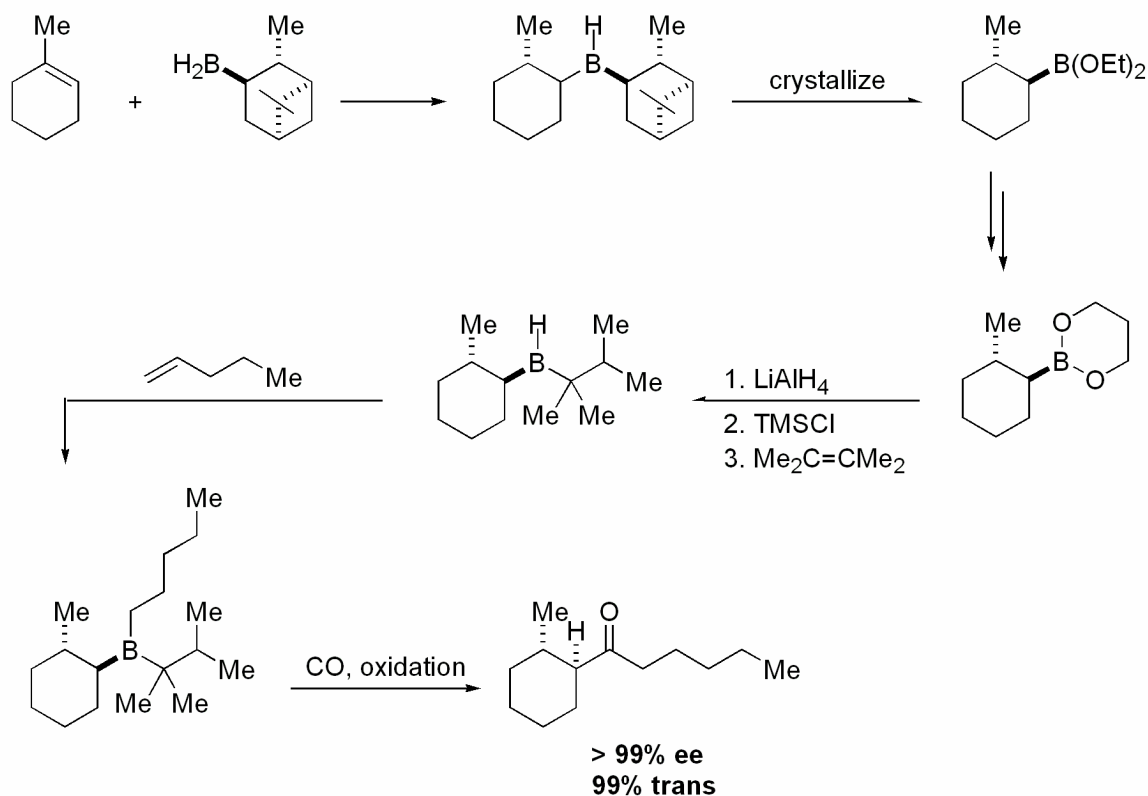


Tetrahedron **1981**, 37, 2547.

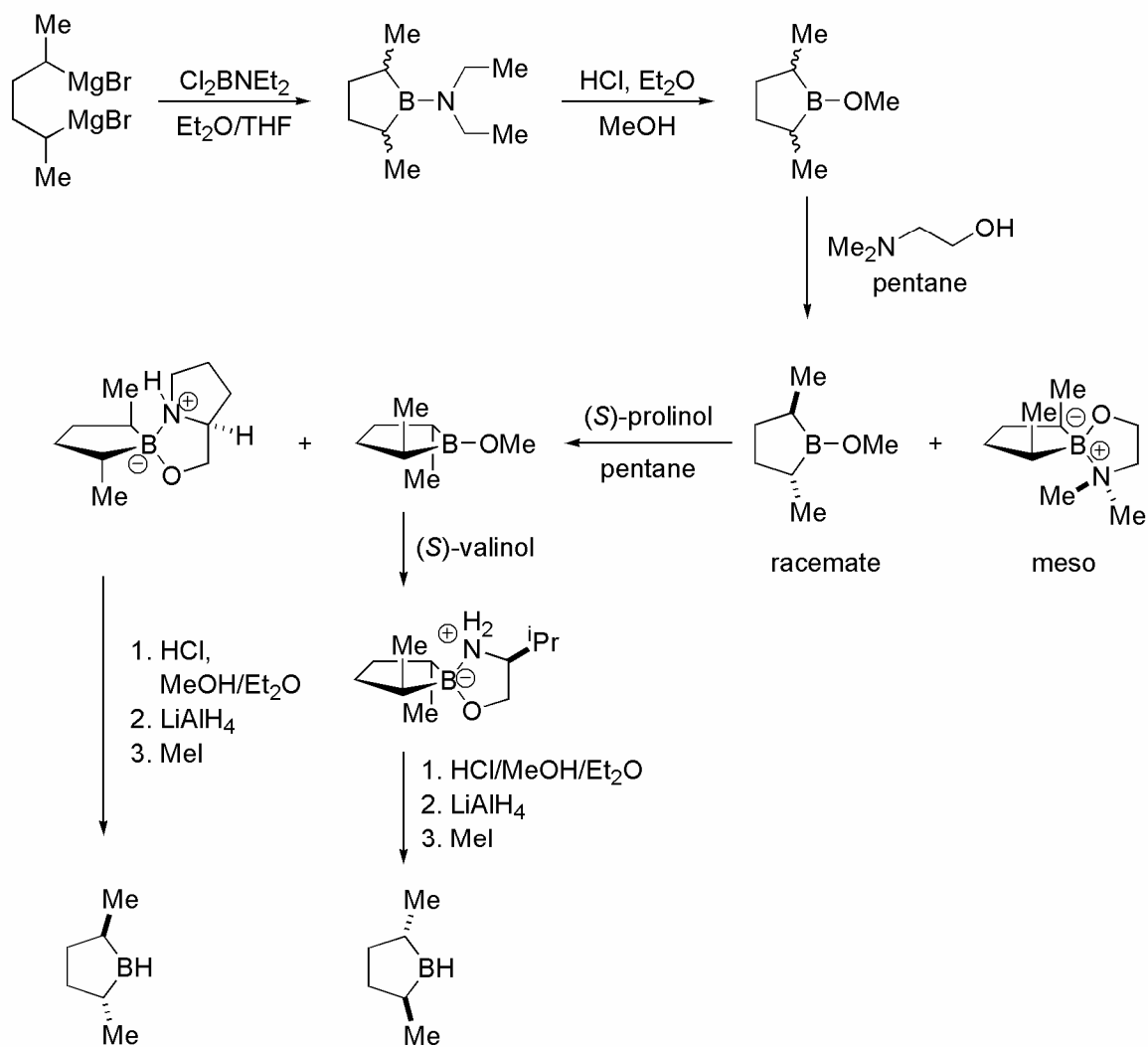
(Z)-alkenes react with better selectivity (75-90% ee) than (E)-alkenes (5-30% ee).



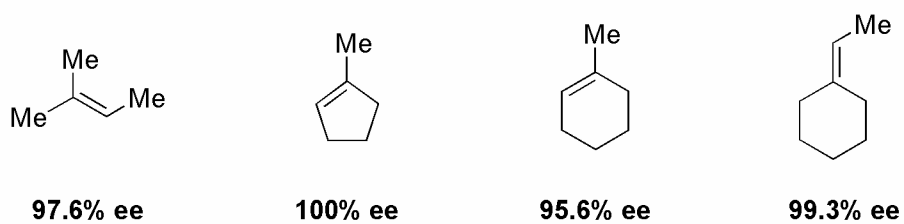
Org. Syn. **1985**, 63, 44.
J. Am. Chem. Soc. **1985**, 107, 4549.



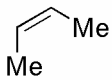
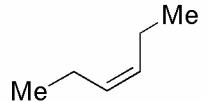
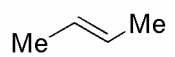
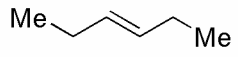
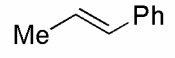
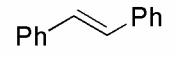
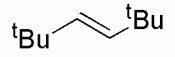
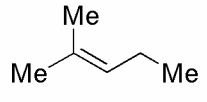
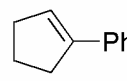
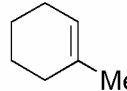
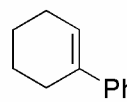

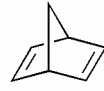


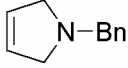

J. Am. Chem. Soc. **1988**, 110, 1529.

Masamune hydroboration:

Masamune *J. Am. Chem. Soc.* **1985**, *107*, 4549.

selected substrates:

substrate scope of the asymmetric hydroboration:

alkene	ee with 1 [%]	ee with <i>lpc</i> ₂ BH [%]
	97.6	98
	99.9	93
	99.5	73
	99.5	75
		76
		65
		92
		58
		100
		72
		88
		83
		80
		93
		100
		100
		83

