## Enantio- and Chemoselective Intramolecular Iridium-Catalyzed O-Allylation of Oximes

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**ABSTRACT:** A method for the enantio- and chemoselective iridium-catalyzed *O*-allylation of oximes is described. Kinetic resolution in an intramolecular setting provides enantioenriched oxime ethers and aliphatic allylic alcohols. The synthetic potential of the products generated with this method is showcased by their elaboration into a series of heterocyclic compounds and the formal synthesis of glycoprotein GP IIb-IIIa receptor antagonist (–)-roxifiban. Preliminary mechanistic experiments and computational data shed light on the remarkable chemoselectivity of the reaction.

Transition metal-catalyzed allylic substitution reactions have emerged as powerful tools for enantioselective carbon-carbon and carbon-heteroatom bond formation.<sup>1</sup> In particular, iridium-catalyzed reactions have proven useful for the asymmetric synthesis of chiral building blocks due to a strong preference for branched allylation products.<sup>2</sup> We have recently documented the iridium and Brønsted acid-catalyzed enantio- and chemoselective intramolecular *N*-allylation of oximes (Scheme 1).<sup>3</sup> The method provides 5-, 6-, and 7-membered cyclic nitrones **2** *via* kinetic resolution of secondary allylic alcohols and proceeds with remarkable selectivity for *N*-allylation with the *O*-allylation adducts merely as minor side products. Because *O*-allylated cyclic oxime ethers are important structural subunits in drug discovery and crop protection,<sup>4</sup> procedures for their enantioselective synthesis are warranted. Herein, we report the iridium-catalyzed, chemoselective, intramolecular *O*-allylation of oximes. Preliminary mechanistic studies were conducted to investigate this remarkable switch in chemoselectivity.

Cyclic oxime ethers, in particular 5-membered isoxazolines, have been highlighted as privileged structural motives in pharmaceuticals and pesticides.<sup>4</sup> However, methodology for the enantioselective synthesis of cyclic oxime ethers relying on transition metal-catalyzed allylic substitution remain undeveloped.<sup>5,6,7</sup> The groups of Du<sup>8</sup> and Takemoto<sup>9</sup> have reported enantioselective intermolecular *O*-allylation reactions to access acyclic oxime ethers using palladium or iridium catalysts, respectively. In 2019, You and co-workers documented a single example of intramolecular iridium-catalyzed *O*-allylation of an oxime furnishing a 7-membered 1,2,5-oxadiazepane product.<sup>10</sup> Yet, a general enantioselective method that provides access to cyclic oximes ethers of various ring sizes remains elusive.





At the onset of the methodological studies described herein, we envisioned that chemo- and enantioselective *O*-allylation of oximes in an intramolecular setting could provide convenient approaches for the preparation of cyclic oxime ethers of various ring sizes. In our previous report,<sup>3a</sup> the chiral catalyst derived from  $[Ir(cod)Cl]_2$  and phosphoramidite-olefin ligand (*S*)-L<sub>1</sub> combined with Brønsted acid cleanly converted oxime **1a** (Scheme 1, R = H, n = 2) to the corresponding 6-membered nitrone **2a** with complete chemoselectivity for *N*-allylation. Thus, a series of experiments aimed at identifying conditions for selective *O*-allylation were conducted (see supporting information). Gratifyingly, we found that in the presence of base

 $(Cs_2CO_3)$  and a Lewis acid  $(Sc(OTf)_3)$ , the same iridium(I) catalyst  $([Ir/(S)-L_1])$  effected the conversion of hydroxy oxime **1a** to 7-membered oxime ether **3a** via kinetic resolution (47% yield, theoretical yield: 50%, 96% ee, Table 1). Notably, N-allylated nitrone **2a** was not detected using these reaction conditions.

Having established optimal parameters for iridium-catalyzed, chemoselective O-allylation of oximes, we next focused on the substrate scope (Table 1). The optimized reaction protocol allowed facile preparation of 6-membered dihydroxazine **3b** with complete O-selectivity. Similarly, 2-thienyl dihydroxazine **3c** was accessed in 44% yield and 95% ee as a crystalline solid suitable for X-ray crystallographic analysis. Enantiopure allylic alcohol (S)-1c (R = 2-thienyl, n = 1) was reisolated in 41% yield. In addition to 7-membered oxazepane (3a) and 6-membered oxazine (3b and 3c) scaffolds, we examined the formation of 5-membered isoxazolines. The optimized protocol for O-allylation furnished **3d** in 45% yield and 93% ee. Analogously, isoxazoline **3e**, was accessed in good yield and high enantiomeric purity (41% yield, 92% ee). Notably, the kinetic resolution described is highly efficient with calculated selectivity factors s > 50 for all substrates.<sup>11</sup>



aReactions conducted on 0.3 mmol scale. Numbers in parentheses refer to recovered starting material 1. Yields refer to isolated products after flash column chromatography. Enantiomeric excess values (ee) were determined by HPLC, SFC or GC analysis on a chiral stationary phase. Selectivity factors (s) calculated using Kagan's method.<sup>11 b</sup> reaction run at rt. <sup>c</sup>Thermal ellipsoids displayed at 50% probability level.

Next, we aimed to highlight the synthetic utility of the established protocol for chemodivergent kinetic resolution of oximes that would give rise to optically active cyclic oxime ethers, nitrones,<sup>3a</sup> and aliphatic allylic alcohols. To this end γ-hydroxy thienyl oxime **1c** was chosen as a starting point for the preparation of diverse enantioenriched building blocks. (Scheme 3A).



## Scheme 3. Functionalization of Products Derived from Oxime 1c<sup>a</sup>

<sup>a</sup>Reactions conducted on 0.3 mmol scale. Yields refer to isolated products after flash column chromatography. Reagents and conditions: (a) NaBH<sub>4</sub>, MeOH. (b) SmI<sub>2</sub>, THF/MeOH (4:1). <sup>b</sup>Thermal ellipsoids displayed at 50% probability level.

Rhodium-catalyzed, oxime directed C–H activation<sup>12</sup> of enantioenriched (*S*)-**1c** furnished thienopyridine **4**, a promising motif in drug discovery,<sup>13</sup> without erosion of optical purity. Lithiation of dihydrooxazine **3c** and subsequent trapping with allyl bromide allowed the selective installation of an additional stereocenter (**5**),<sup>14</sup> showcasing the applicability of the method for the synthesis of highly substituted heterocycles. Next, nitrone **2c** was treated with EtMgBr at –78 °C to afford trisubstituted pyrrolidine **6** in 95% ee and 12:1 dr. These experiments showcase that starting from a single racemic compound (**1c**), a series of structurally diverse enantioenriched products can be accessed rapidly.

In addition, we sought to showcase the method in the context of target-oriented synthesis (Scheme 3B). We aimed at the formal synthesis of (–)-roxifiban (**8**), a glycoprotein GP IIb-IIIa receptor antagonist investigated in clinical trials by DuPont for the treatment of various cardiovascular ailments including platelet adhesion (Scheme 3B).<sup>15</sup> Enantioenriched isoxazoline **3e** was hydroborated and oxidized to the corresponding primary alcohol using 9-BBN and H<sub>2</sub>O<sub>2</sub>. Further oxidation with pyridinium dichromate (PDC) in anhydrous DMF cleanly afforded carboxylic acid **7** in 71% yield over two steps. Compound **7** has been used previously by Olson and co-workers in their synthesis of (–)-roxifiban thus completing the formal synthesis of **8**.<sup>15a</sup>

Over the course of this study we became interested in the origin of the remarkable chemoselectivity observed for both *N*and *O*-alkylation reactions (Scheme 4). In particular, we wondered if the formation the cyclic nitrones or oxime ethers is governed by thermodynamic biases and whether the size of the newly formed ring factors into this. The relative energies of two pairs of constitutionally isomeric nitrones and oxime ethers were calculated (**2a/3a** and **2c/3c**, Scheme 4). Interestingly, density functional theory calculations at the B3PW91/6-311++G(d,p) level of theory revealed the nitrone products to be thermodynamically favored over the oxime ethers by  $\Delta\Delta G^0$ = 5.0 kcalmol<sup>-1</sup> and  $\Delta\Delta G^0$ = 5.7 kcalmol<sup>-1</sup>, respectively, regardless of ring size as shown in Scheme 4.<sup>16,17</sup>





 $a\Delta\Delta G_{rel}$  values calculated using DFT at the B3PW91/6-311++G(d,p) level of theory, see supporting information for details.

To probe whether the chemoselective formation of nitrone **2a** may occur *via* equilibration from **3a**, we conducted a series of control experiments (Scheme 5). Subjecting racemic oxime ether **3a** to the optimized reaction conditions for *N*-allylation afforded the corresponding nitrone **2a**. This result demonstrates that C–O bond cleavage occurs for **3a** in the presence of the iridium catalyst under acidic conditions. Control experiments indicated that both  $Cl_2CHCO_2H$  and  $[Ir/(S)-L_1]$  were necessary for conversion of **3a** to give **2a**. Next, we investigated whether nitrone **2a** undergoes reversible C-N bond cleavage. Subjecting **2a** to the conditions optimized for *O*-alkylation did not afford any detectable formation of oxime ether **3a**. Treating **2a** with  $Cl_2CHCO_2H$  and the iridium catalysts derived from either (*R*)-L<sub>1</sub>, the enantiomeric ligand, or achiral L<sub>2</sub> did not lead to any erosion of optical purity, even after prolonged reaction times (48 hrs). Collectively, these experiments indicate that C-N bond cleavage does not occur under basic nor acidic conditions. We postulate that while *O*-alkylated product **3a** may be formed under acidic conditions as the kinetic product, it can undergo C-O bond cleavage to afford thermodynamically favored nitrone **2a**. The formation of oxime ether **3a** under basic conditions is kinetically favored, as a consequence of deprotonation of the oxime-OH (pKa of benzophenone oxime in water ~ 11).<sup>19</sup> However, our observations with **2a/3a** were not generalizable, as subjecting six membered oxime ether **3c** to  $Cl_2CHCO_2H/[Ir]$  did not lead to formation of nitrone **2c**. Hence, 6-membered oxime ether **3c** does not undergo equilibration which explains the lower chemoselectivity observed in the formation of nitrone **2c** despite a strong thermodynamic bias for *N*-allylation



<sup>a</sup>Reactions were set up according to general procedures, see Table1 and supporting information.

In conclusion, we have developed conditions for the highly enantio- and chemoselective intramolecular *O*-allylation of oximes. The method furnishes cyclic 5-, 6-, and 7-membered oxime ethers in high enantiomeric purity. Combined with our earlier report on chemoselective *N*-allylation of oximes, we have established conditions chemodivergent synthesis of cyclic oxime ethers and nitrones from the same oxime starting materials. Experimental and computational data suggest that for some substrates *N*-allylation occurs via equilibration whereas *O*-allylation proceeds under kinetic control. The synthetic utility of the method was shown by functionalization of the cyclization products to give a diverse set of chiral building blocks and the entioselective formal synthesis of glycoprotein GP IIb-IIIa receptor antagonist (–)-roxifiban.

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