

Modeling of surfaces: the third dimension in XPS analysis of multilayer structures

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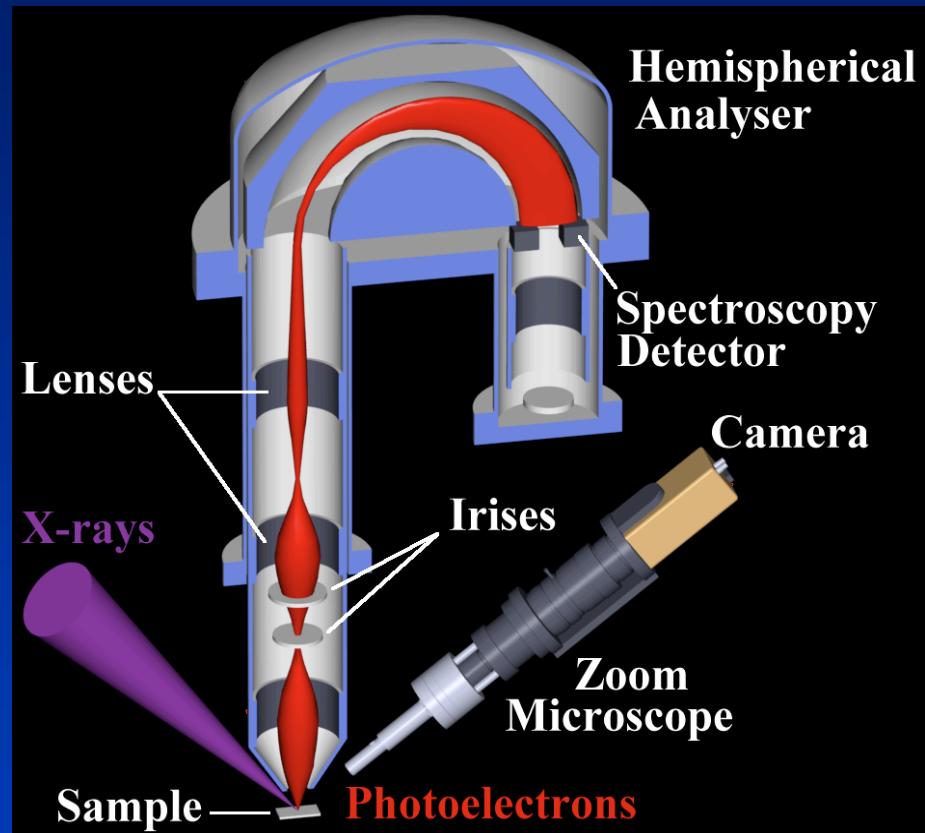


Materials Day: Zürich, January 24, 2003



XPS Analysis

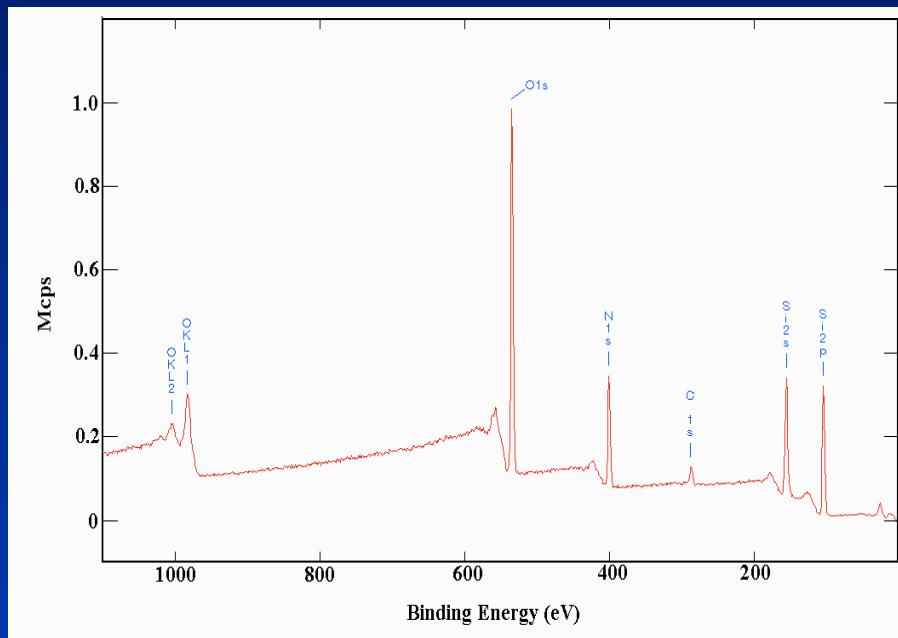
- The material to be analyzed is in UHV
- The surface is irradiated with photons in X-ray range (Alk \square 1486.6 eV)
- The emitted e^- are separated according to their energy by the analyzer and counted



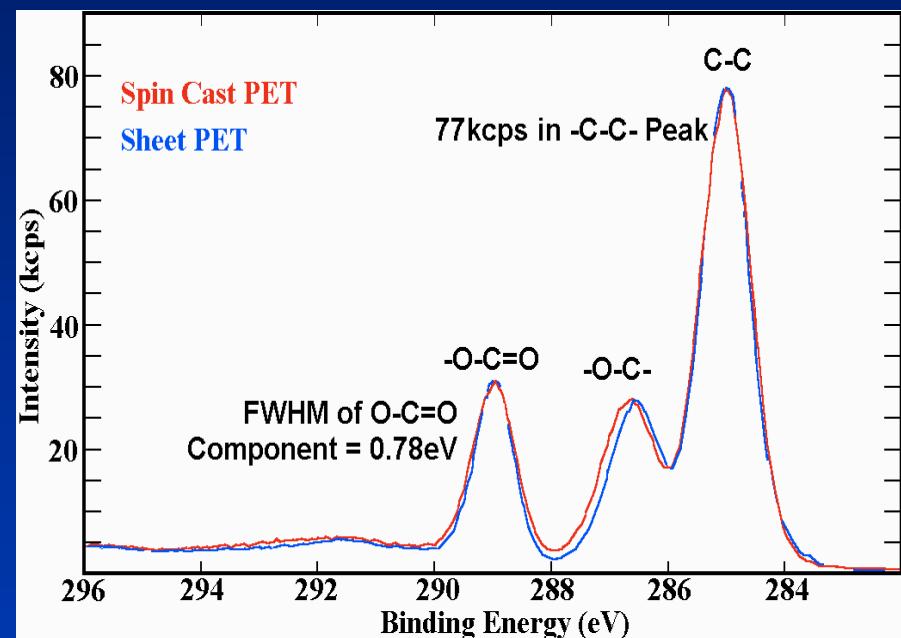
Courtesy of Thermo VG Scientific

X-ray Photoelectron Spectroscopy (XPS)

Type of information

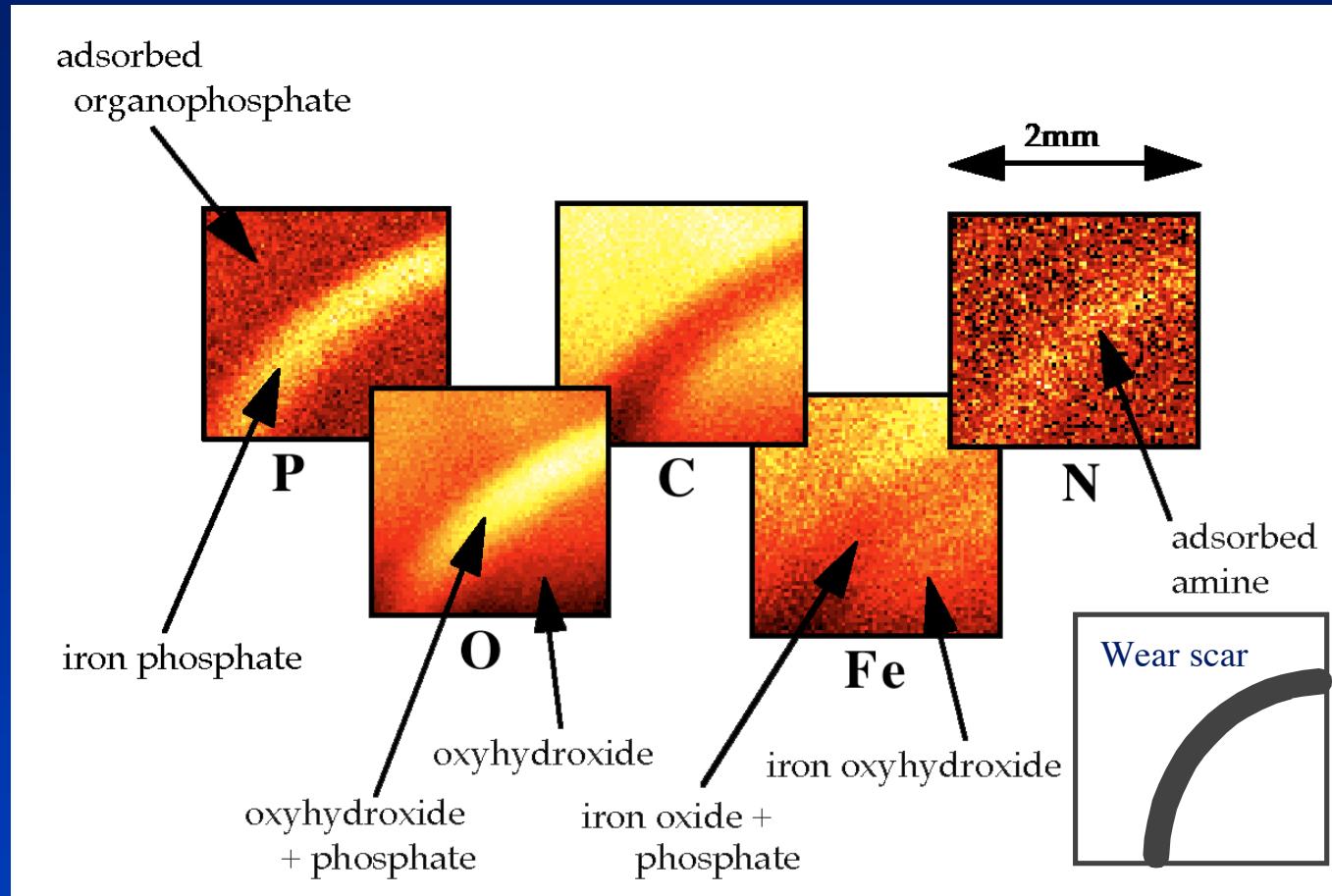


elemental identification



chemical state identification

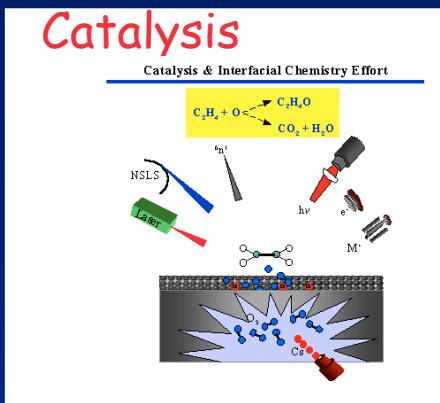
Imaging XPS Study of Additive-Surface Interactions



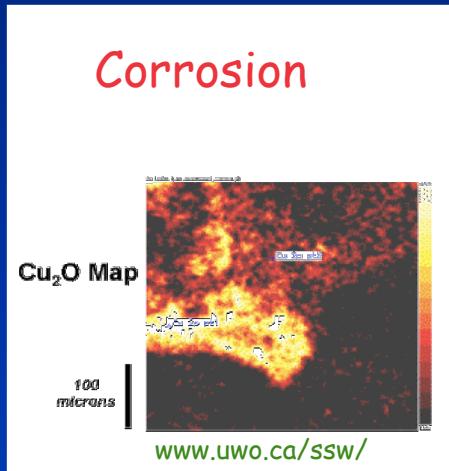
Irgalube® 349/PAO/100Cr6 following tribotesting



Catalysis

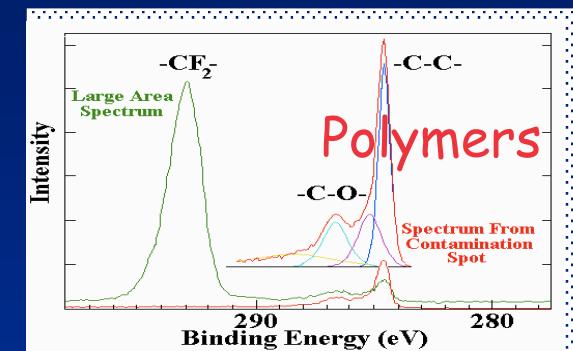
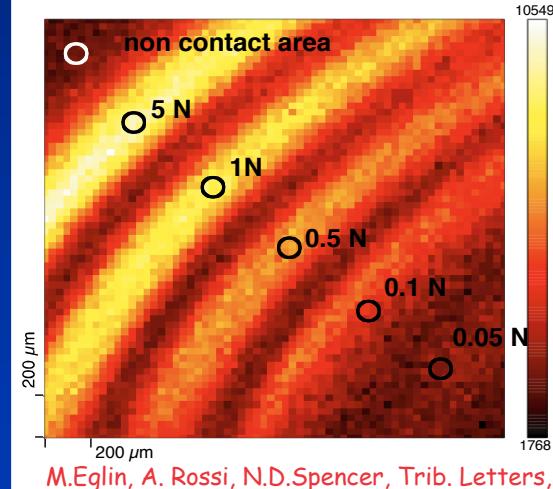


Corrosion

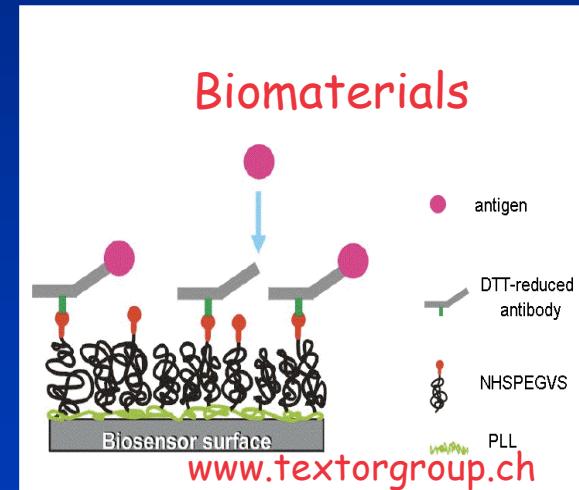


Quantitative analysis is mainly performed under the assumption of the homogeneity of a material.

Tribology



Biomaterials



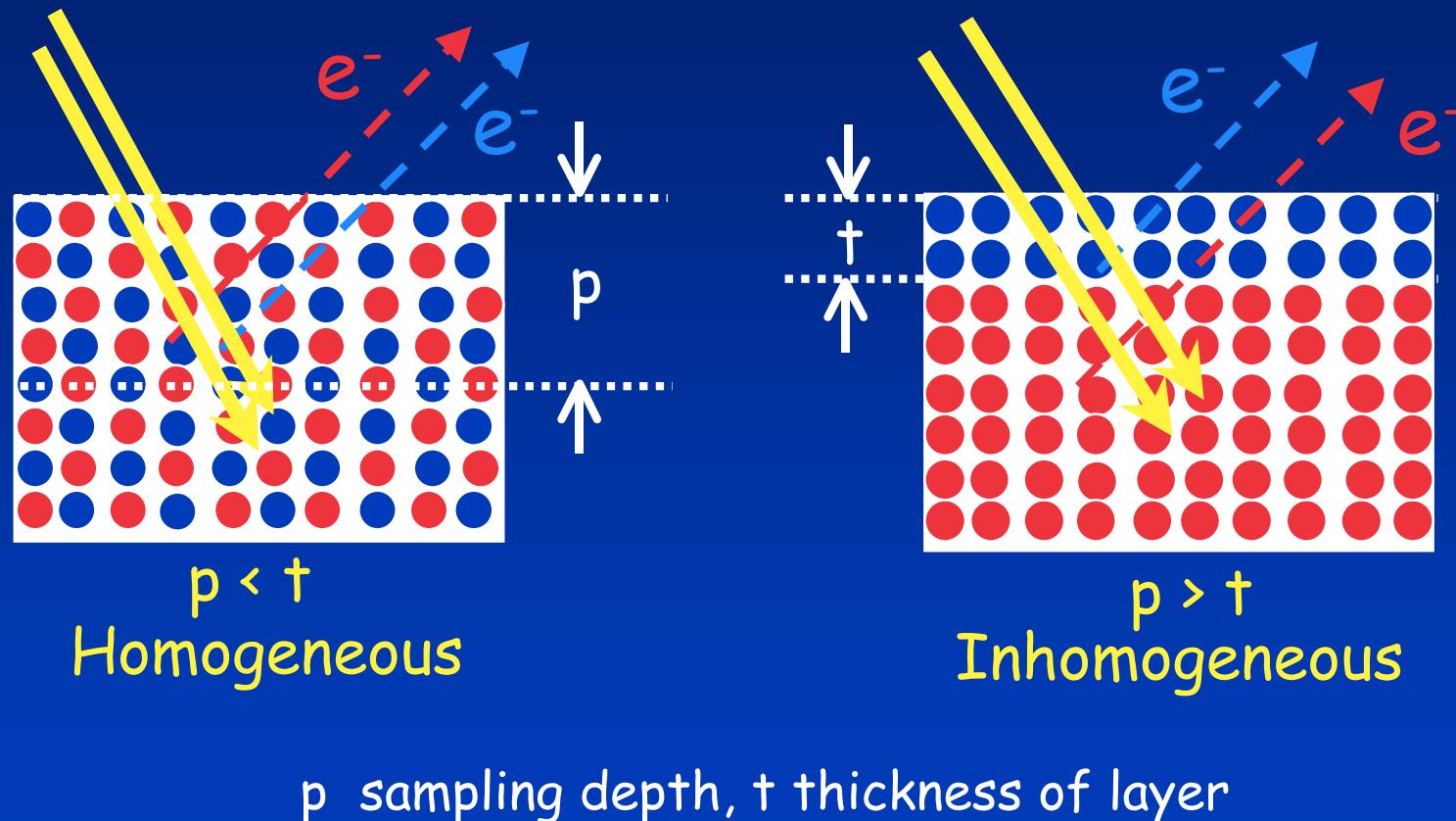
The need of the study of very thin films with a depth resolution on the nanometers range stimulates the proposal of new models for XPS quantification.



Outline

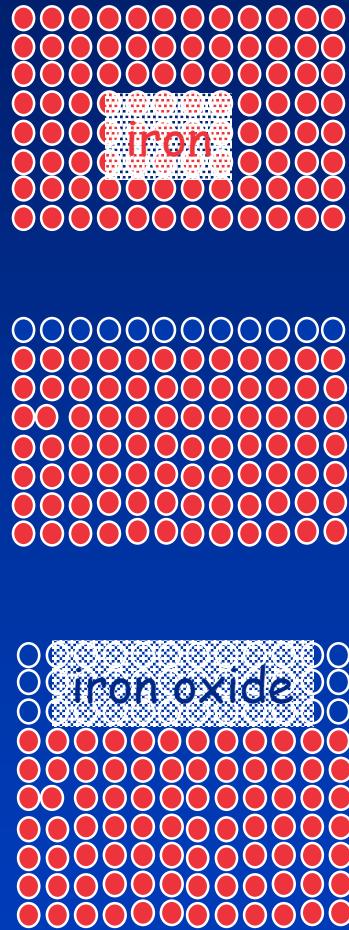
- The need for quantitative XPS information
- Real surfaces
- Modeling of real surfaces: the three layer model
- Examples
- Outlook

XPS depth resolution in the nanometer range



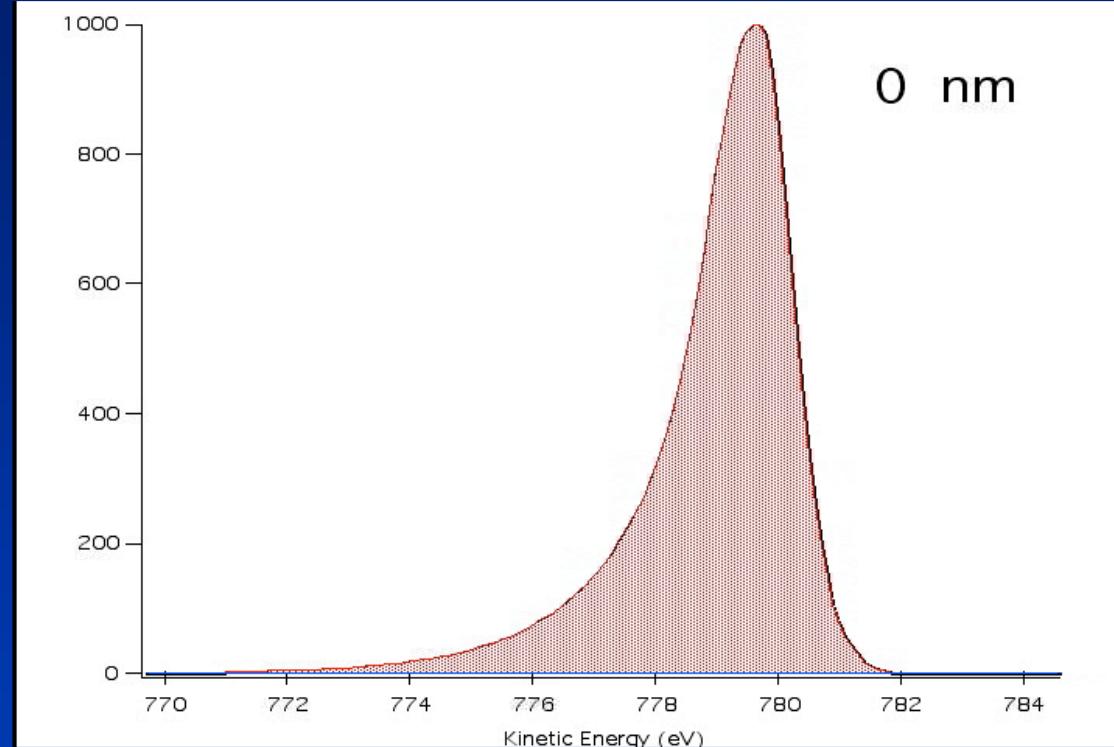
Attenuation of XPS photoelectrons

Example of an iron oxide layer on iron



$t = 0 \text{ nm}$

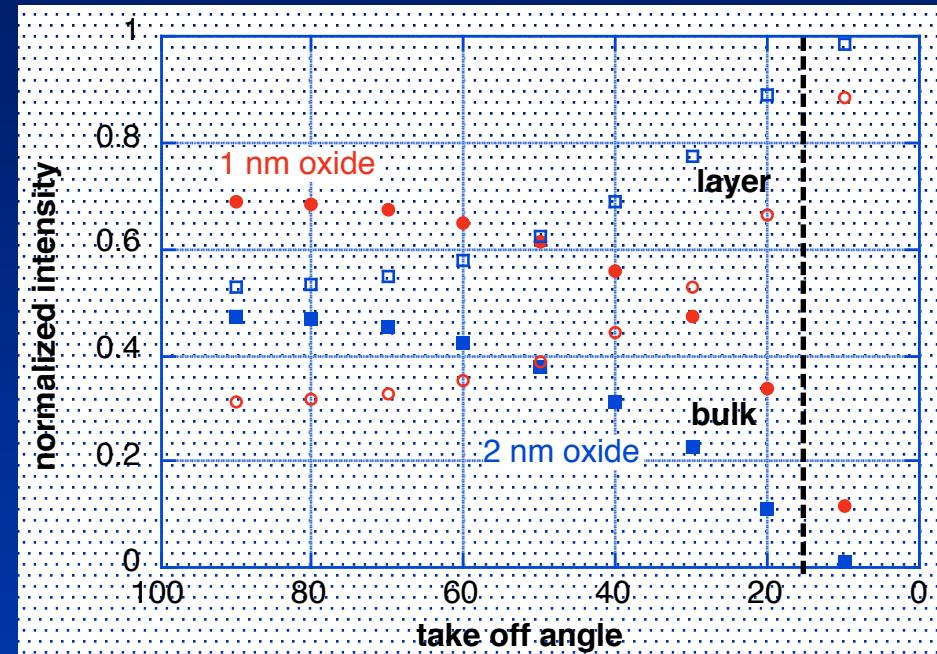
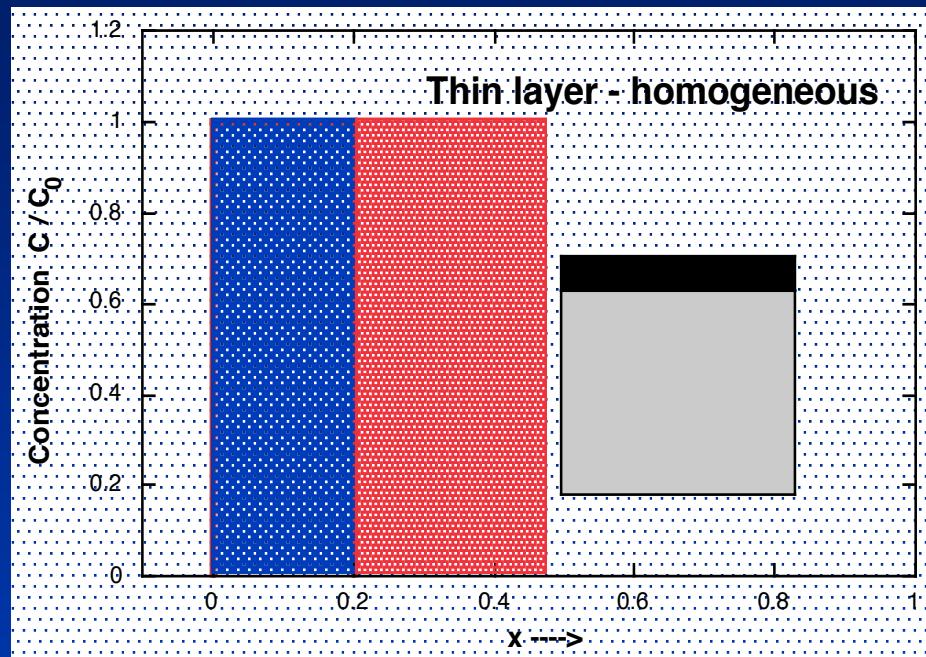
$t = 9 \text{ nm}$



Electrons from the bulk are attenuated by any layer on the surface

Modeling oxide layer - ARXPS

Example of an iron oxide layer on iron



$$I_{\text{bulk}} = I_{\text{bulk}}^0 \exp(-x_L / (\mu_L \sin(\theta)))$$

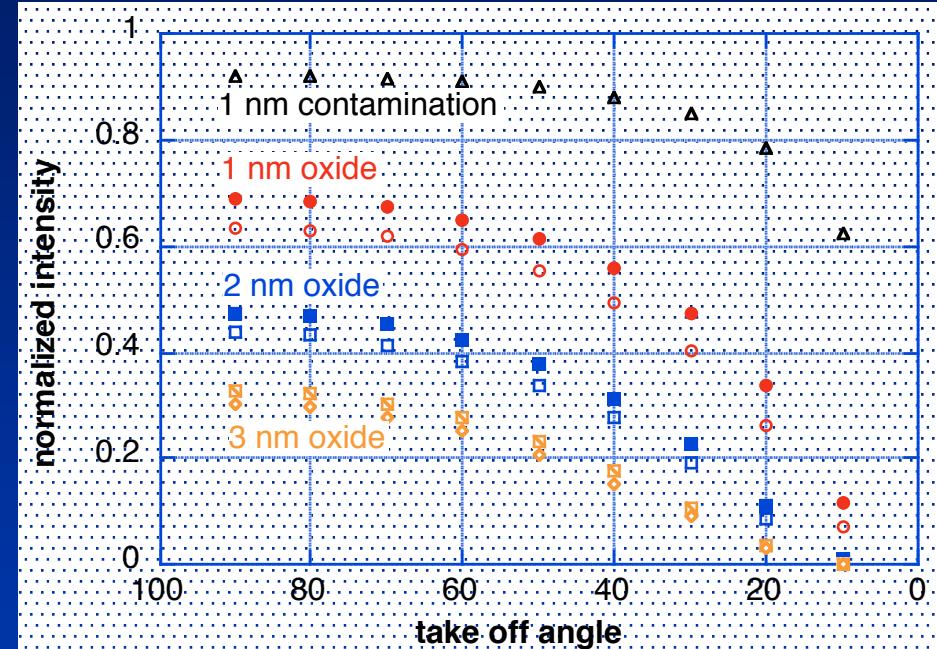
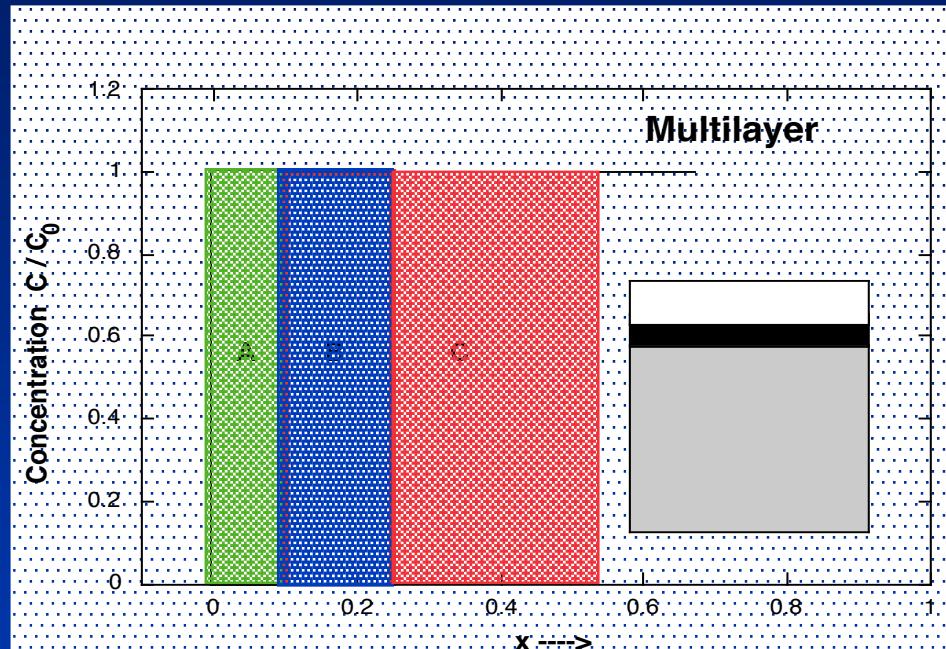
$$I_{\text{Layer}} = I_L^\infty (1 - \exp(-x_L / (\mu_L \sin(\theta)))$$

decreases with x_L and angle
increases with x_L and angle

Electrons from the bulk are progressively attenuated at lower take off angles.

Modeling contamination overlayer - ARXPS

Example of an iron oxide layer on iron with contamination



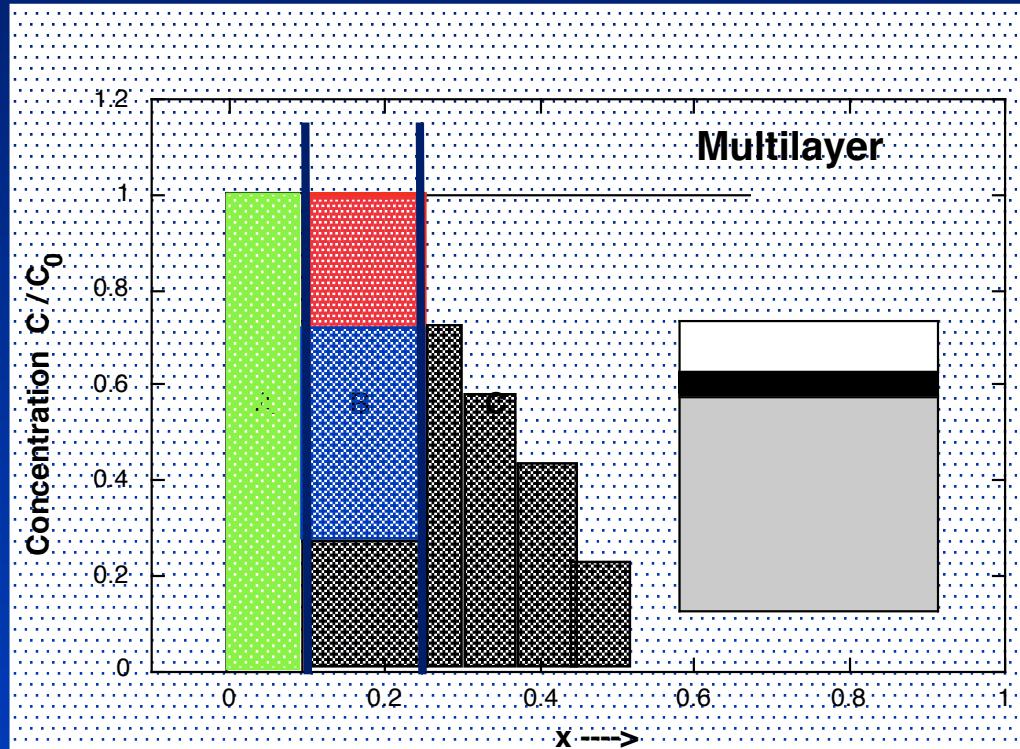
$$I_{\text{bulk}} = I_{\text{bulk}}^0 \exp(-x_L / (\mu_L \sin(\theta)) * \exp(-x_c / (\mu_L \sin(\theta)))$$

$$I_{\text{Layer}} = I_L^\infty (1 - \exp(-x_L / (\mu_L \sin(\theta)) * \exp(-x_c / (\mu_L \sin(\theta)))$$

Layer thickness x_L and x_c are corrected for density

Modeling of multicomponent real surfaces

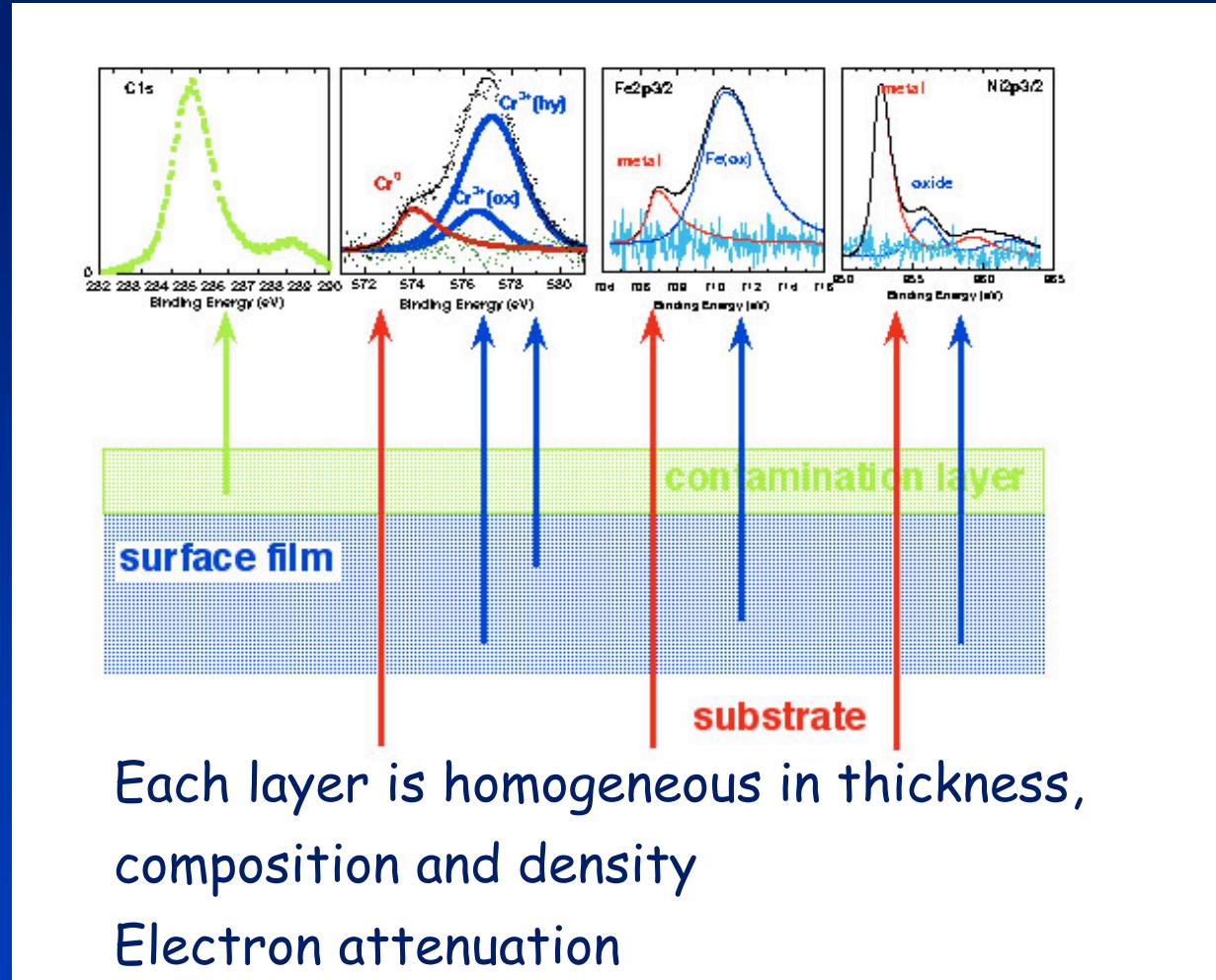
Real surfaces are multicomponent with unknown in-depth
Distribution and a contamination overlayer



- Electron attenuation
- Effect of layer density
- Gradients in composition
- Contamination overlayer

How to calculate composition and thickness from XPS data ?

Three layer model for real surfaces



K. Asami, K. Hashimoto, Corrosion Science, 24, 83, 1984

A. Rossi and B. Elsener, Surface and Interface Analysis, 18, 1992, 499-504.

B. Elsener and A. Rossi, Electrochimica Acta, 37, 1992, 2269-2276

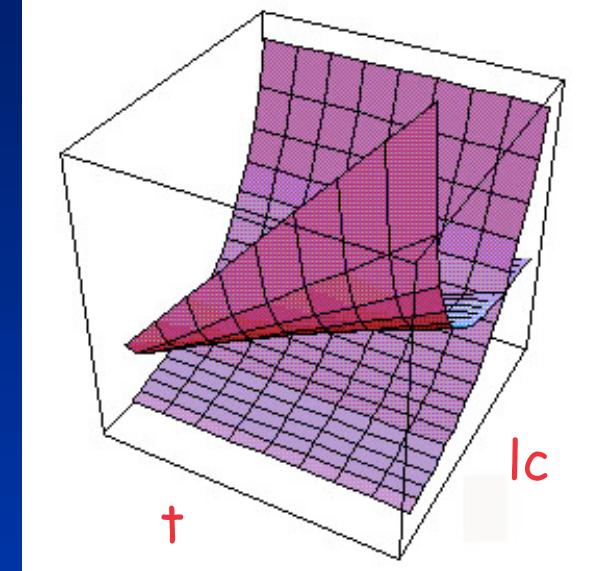
Three-layer model: the equations

$$I_i^{ox} = \frac{[g_i \square_i^{ox} C_i^{ox} \square_i^{ox} \square_i^{ox}]}{A_i} * \exp \left(\frac{-t}{\square_i^{ox}} \right) * \exp \left(\frac{-I_c}{\square_i^c} \right)$$

$$I_i^{sub} = \frac{[g_i \square_i^{sub} C_i^{sub} \square_i^{sub} \square_i^{sub}]}{A_i} * \exp \left(\frac{-t}{\square_i^{ox}} \right) * \exp \left(\frac{-I_c}{\square_i^c} \right)$$

Three-layer model: parametric equations

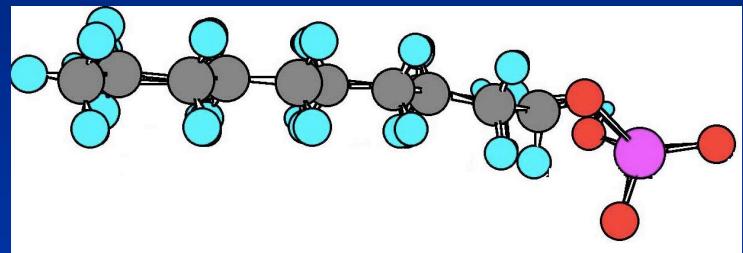
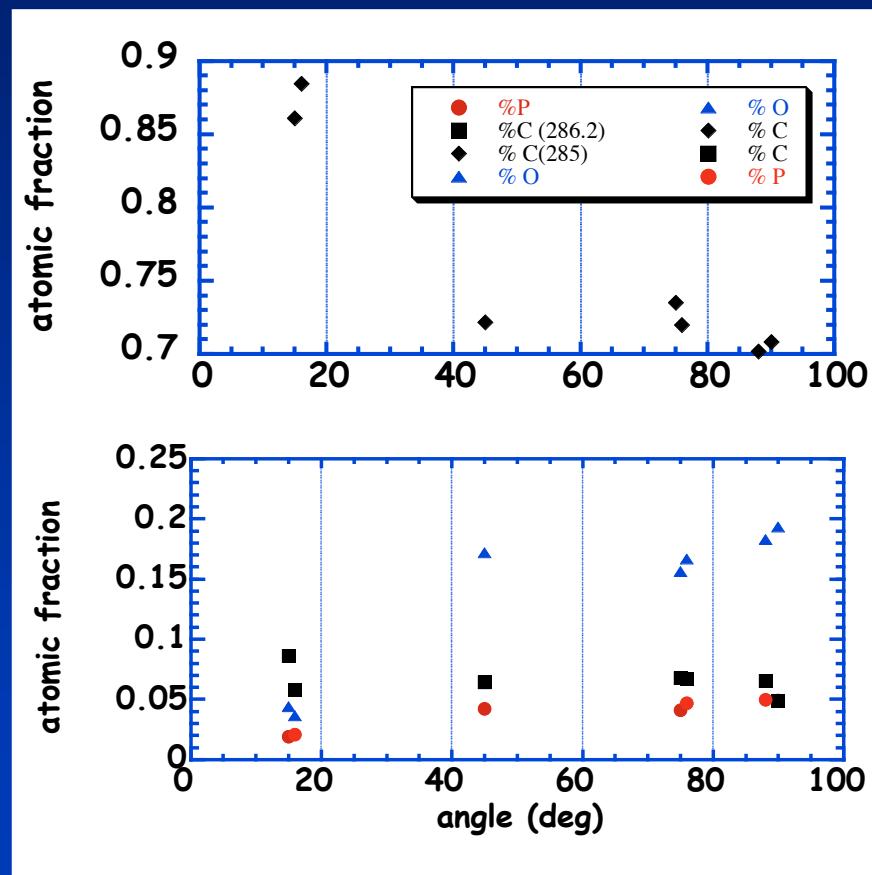
$$f_1(t, I_c) = \frac{I_i^{\text{sub}} k_i^{\text{sub}} \exp\left(\frac{t}{I_i^{\text{ox}}}\right)}{\exp\left(\frac{I_c}{I_i^{\text{c}}}\right) * \exp\left(\frac{I_c}{I_i^{\text{ox}}}\right)}$$
$$f_2(t, I_c) = \frac{I_c K_c I_i^{\text{ox}}}{\exp\left(\frac{I_c}{I_i^{\text{c}}}\right) \exp\left(\frac{I_c}{I_i^{\text{ox}}}\right)}$$



Numerical methods based on versions of Newton's method are used to find numerical approximations to the solutions of the equations. The composition of each layer are calculated simultaneously.

Self-Assembled Monolayers

ODP on Ta_2O_5 ARXPS

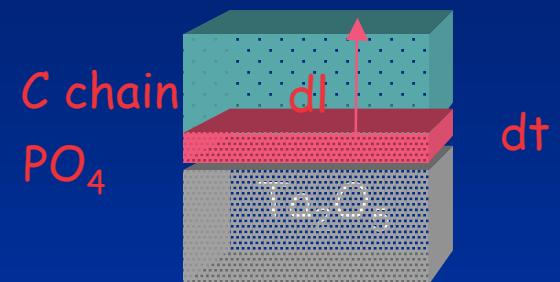
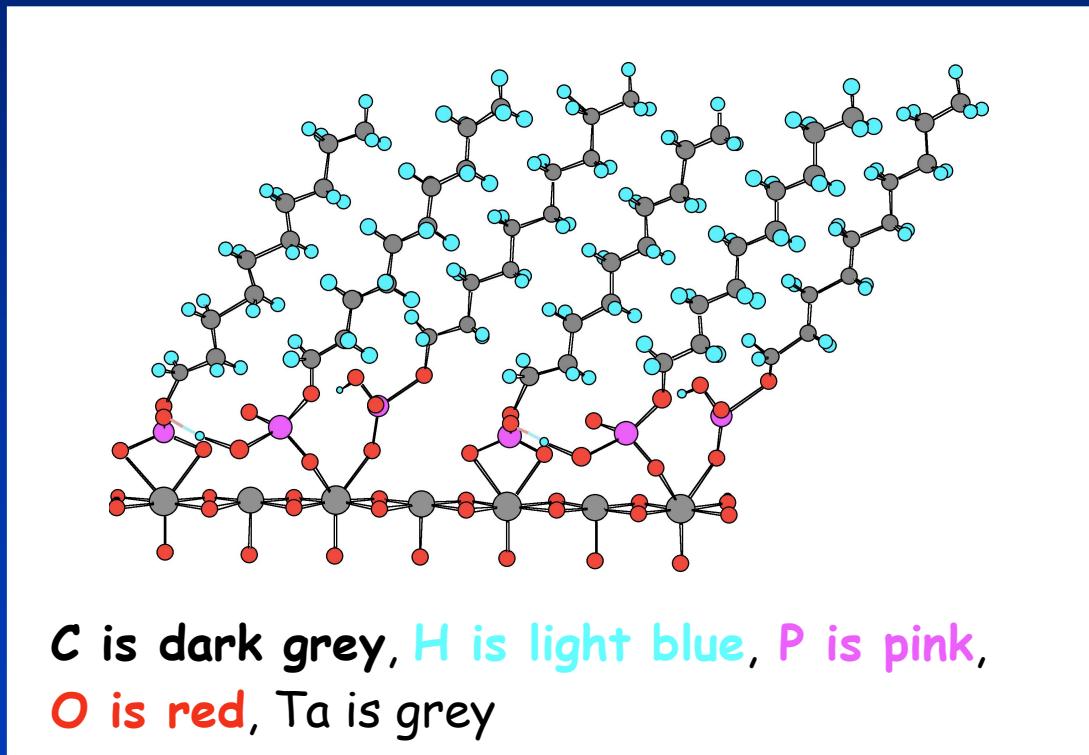


Phosphate head is
at the interface



Self-Assembled Monolayers

ODP on Ta_2O_5



layer homogeneity
e- attenuation according
to Lambert-Beer law
no gradients



Self-Assembled Monolayers

ODP on Ta_2O_5

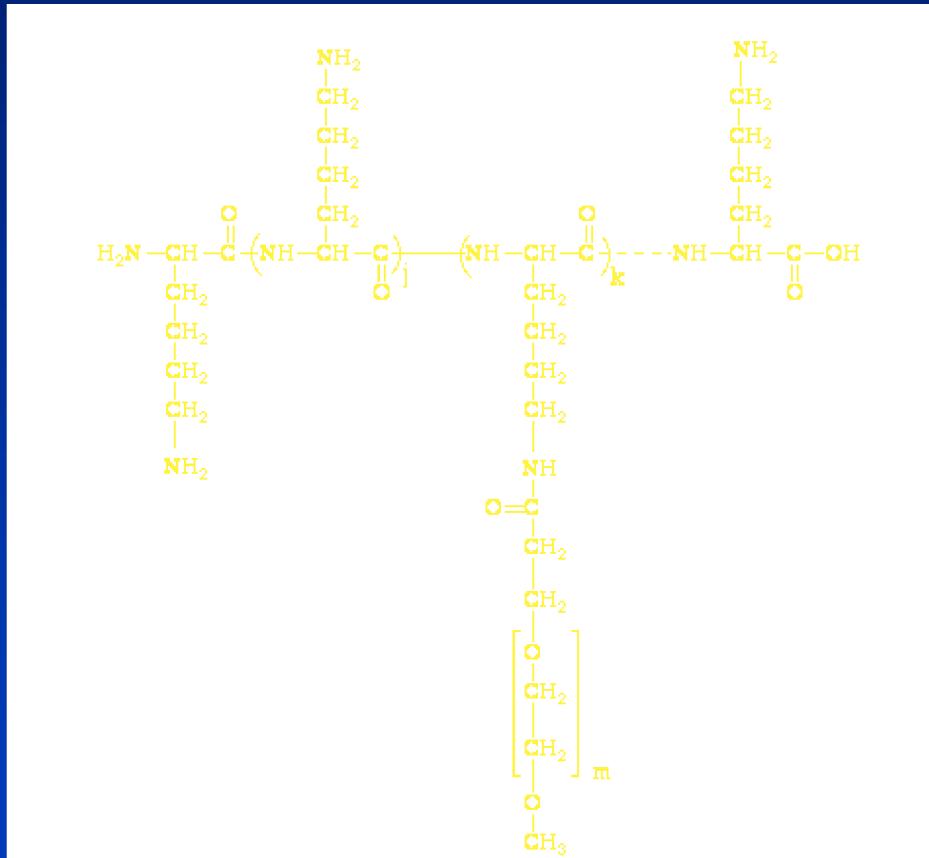
Thickness and composition of self-assembled ODP monolayer on Ta_2O_5 *.

	C	O	P	O	Ta
	ODP chain	polar head		substrate Ta_2O_5	
thickness (nm)	$l_c = 1.2 \pm 0.2$	$t = 1.2 \pm 0.1_5$		semi infinite	
composition weight%					
theoretical	-	67.4	32.6	18.1	81.9
experimental	-	69±1	31±1	17.3±1	82.7±1

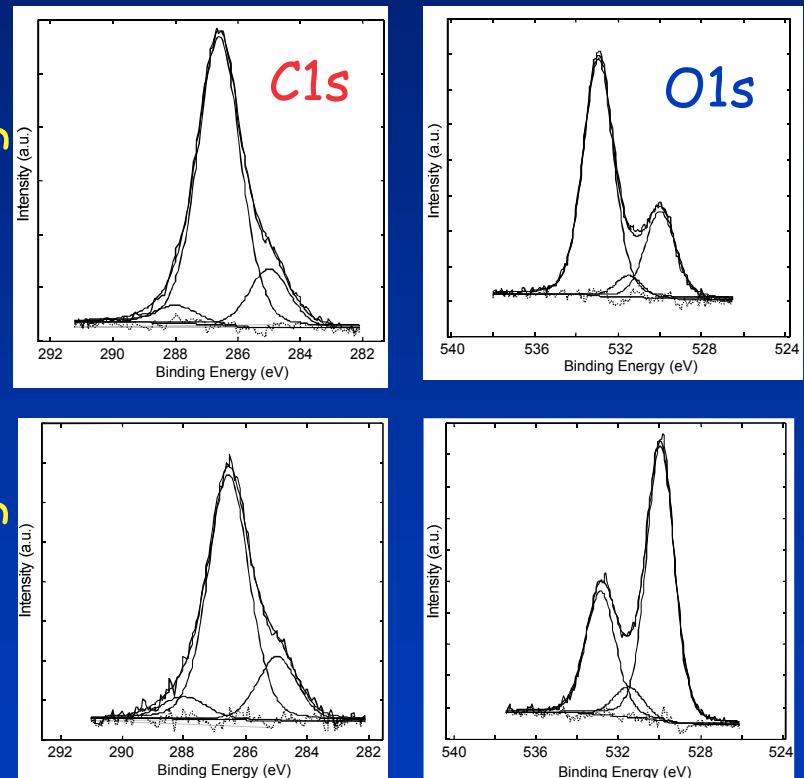


Protein resistant biomaterials

PLL-g-PEG on Nb_2O_5



15 deg
75 deg



Protein resistant biomaterials

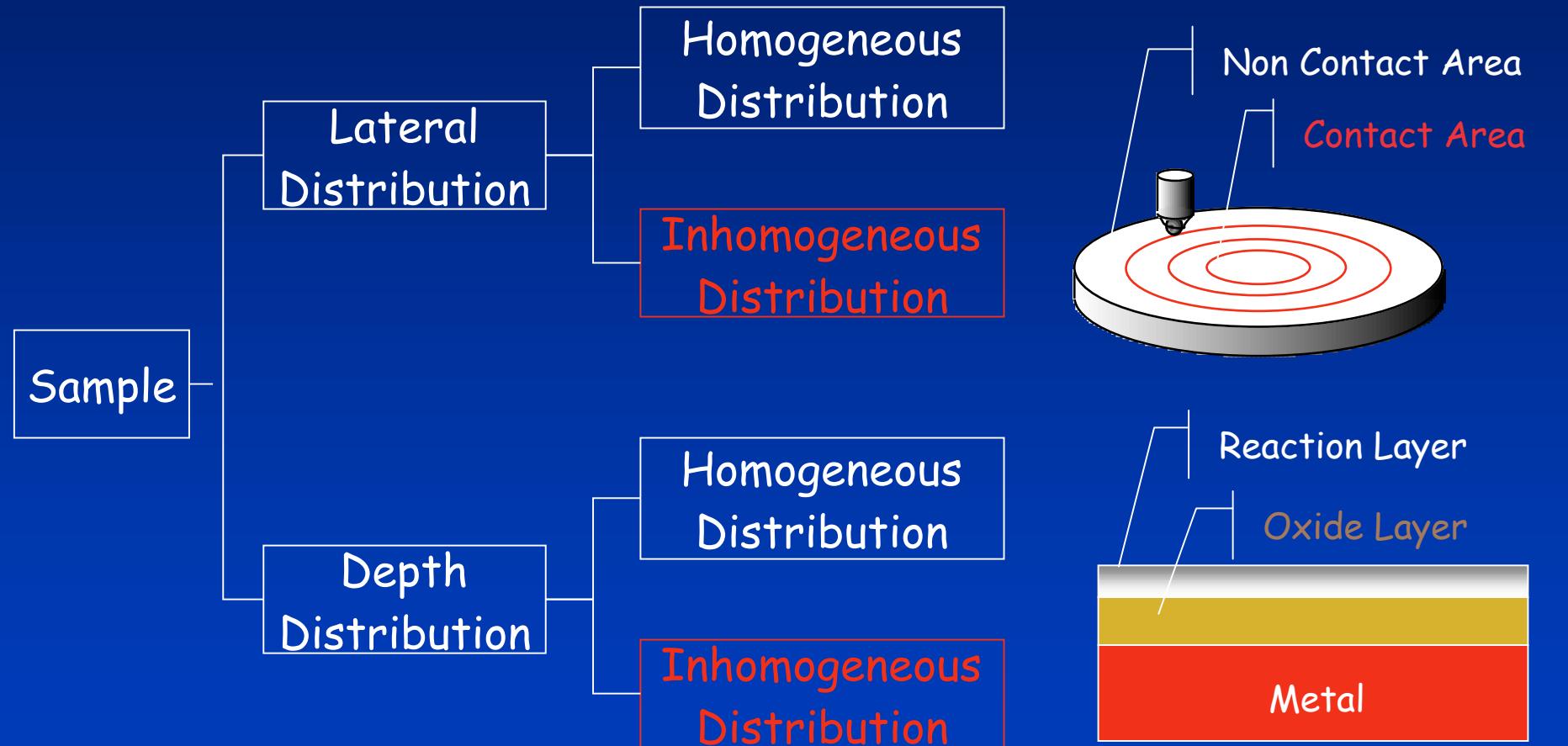
PLL-g-PEG on Nb_2O_5

angle	thickness (nm)		composition				
			interface (PLL)			bulk	
	PEG	PLL	C1s	N1s	O1s	metal	O1s
Exper.	1.1±0.3	0.6±0.2	59	21	20	72	28
Calc.	-	-	59	23	18	70	30

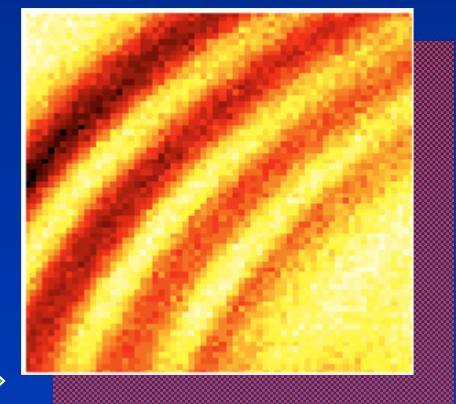
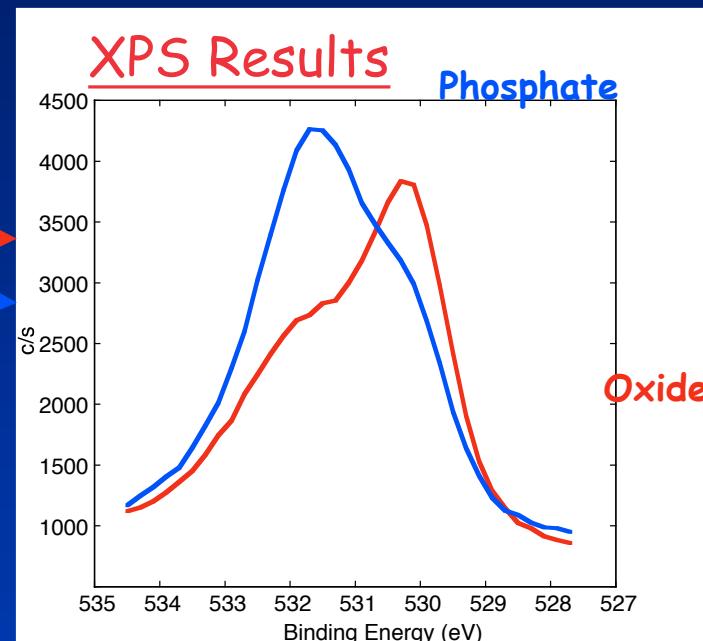
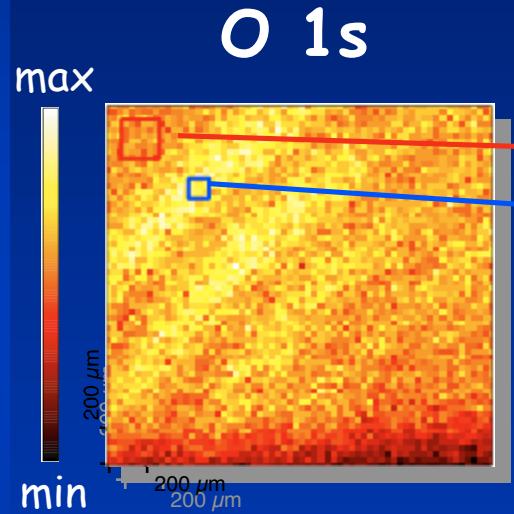
- PEG = 2 g cm⁻³
- PLL = 1 g cm⁻³
- Nb_2O_5 = 4.47 g cm⁻³

The thicknesses are fully consistent with monolayer coverage of PLL-g-PEG and with a surface coverage of 148 ng cm⁻³ measured by OWLS.

Real surfaces



Imaging XPS



Extract Spectra

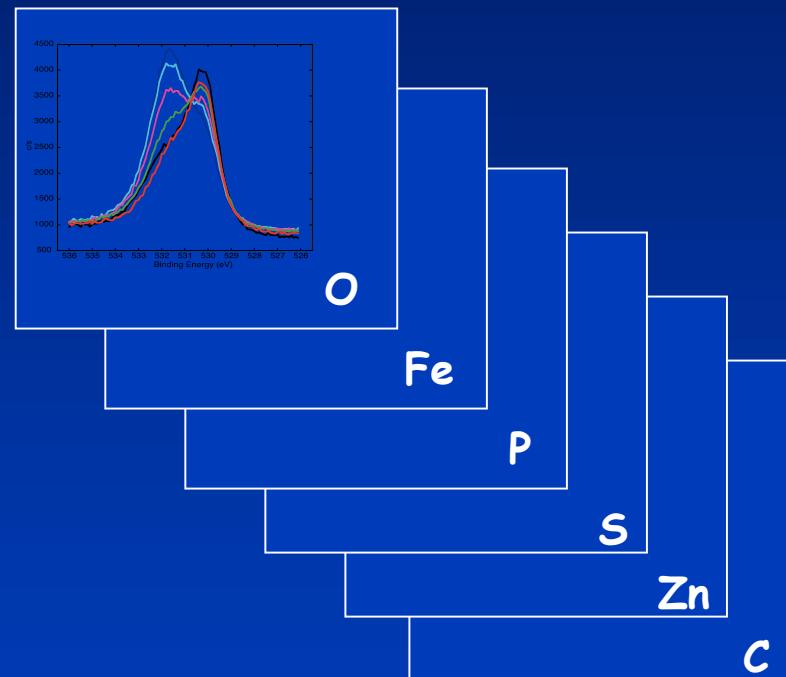
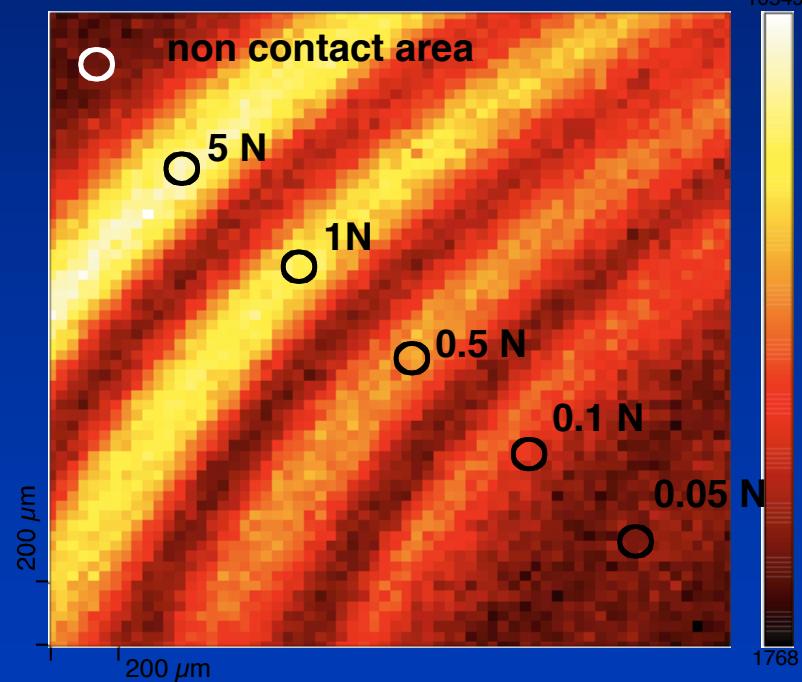


Reconstruct
Map



From Imaging to Spectroscopy

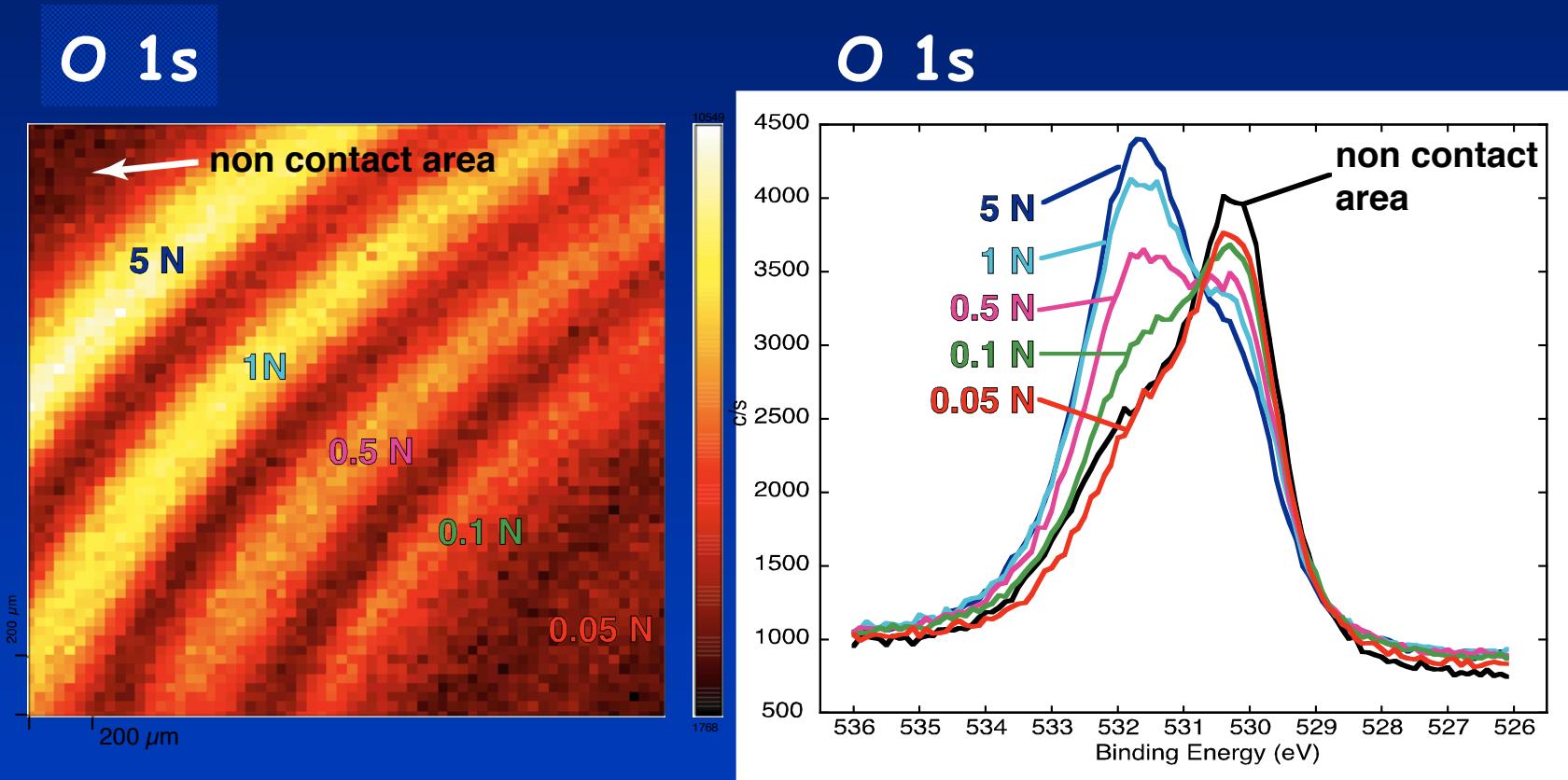
$O\ 1s$



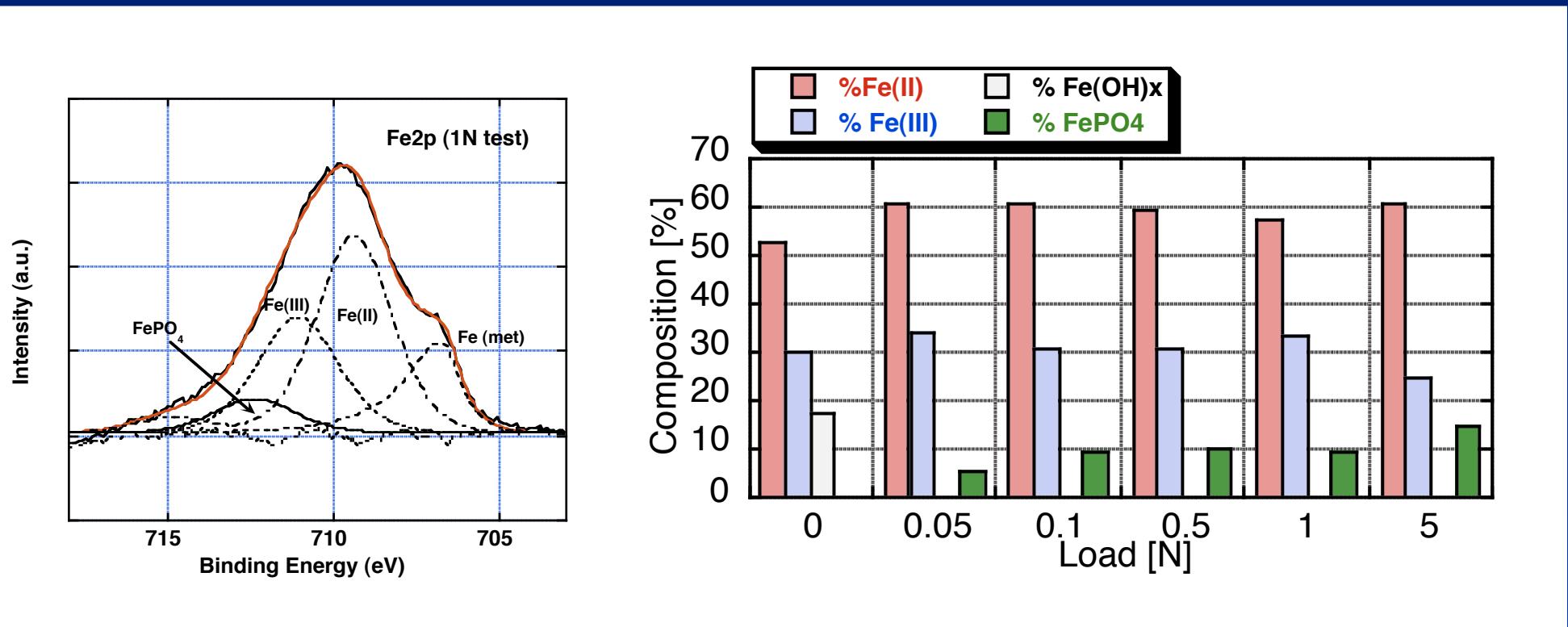
Select Areas for High Resolution-Small Area XPS



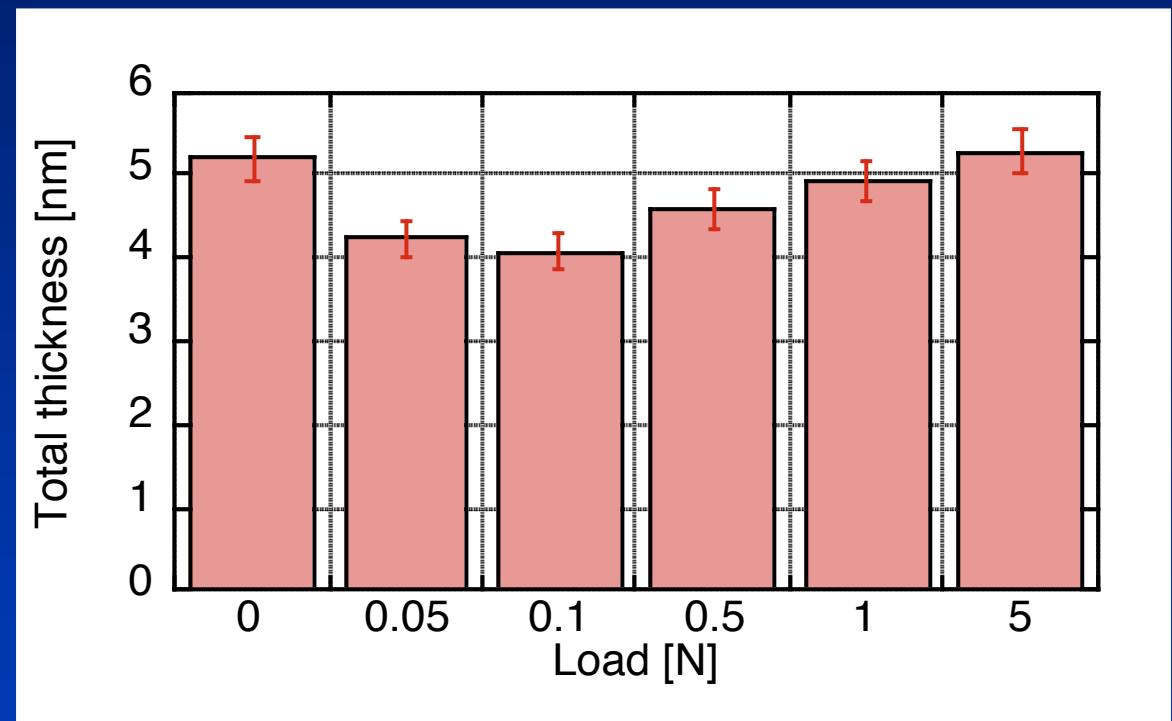
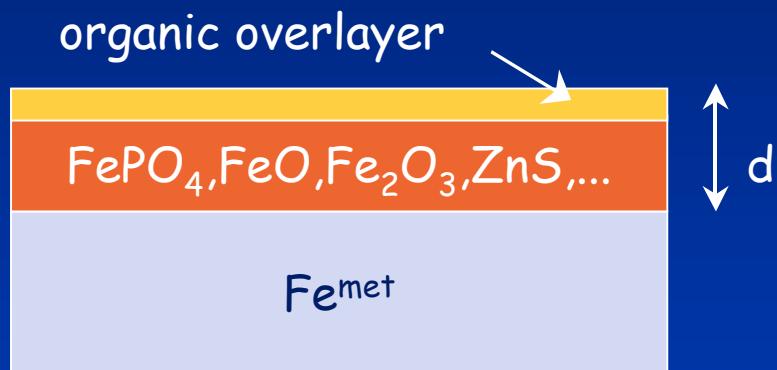
XPS Results



XPS Results (Iron)



XPS Results - film thickness



Outlook

The newest spectrometers allow imaging quantification with:
higher lateral resolution ($<3\mu\text{m}$)
collection of spectroscopic data with high sensitivity

The application of the **three-layer model** at any point of the image will provide information:

- local thicknesses variations
- local layer composition changes
- local interface composition variations

Acknowledgments

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Michael Eglin, Ning-Ping Huang, Laurence Ruiz

