



Wir schaffen Wissen – heute für morgen

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**Hydrothermal Gasification of Microalgae for
Biomethane Production (SunChem Process)**

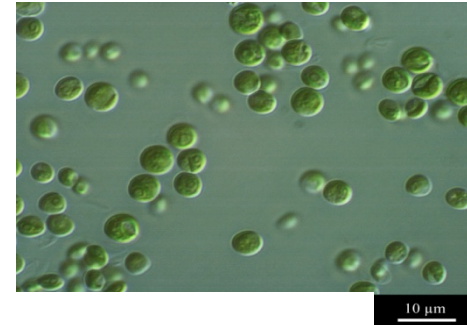
- Why microalgae as a bio-energy source? Why biomethane?
 - Hydrothermal Gasification Process (SunCHem Process)
 - Ru/C catalysts for CSCWG of isopropanol as organic model compound
 - Conclusion and perspectives
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Why microalgae as a bio-energy source?

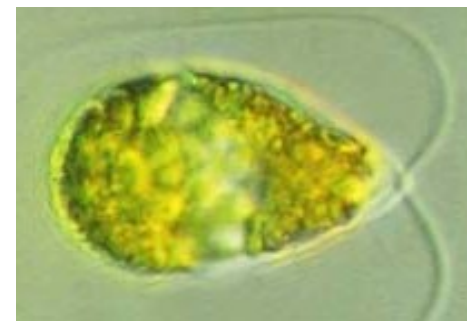
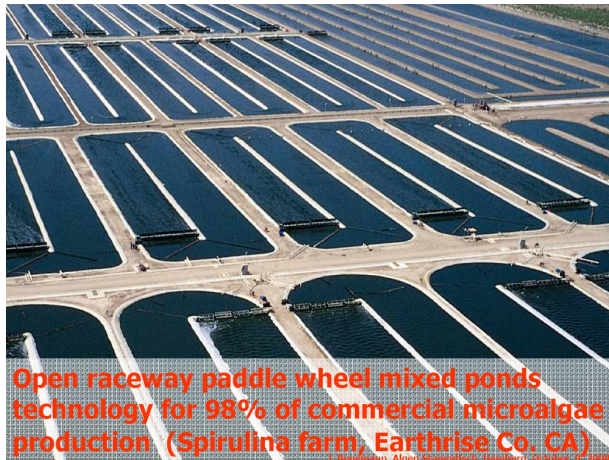
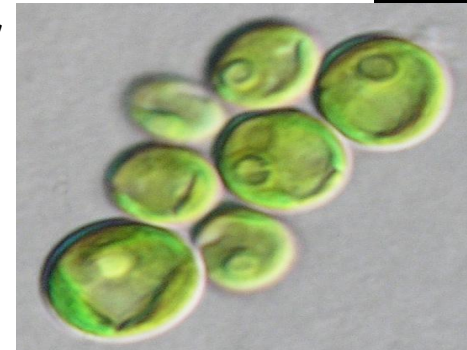
No competition with food crops (3th generation biofuels)

High photosynthetic efficiency and biomass production rates

- Forest **8-12 t/ha/yr**
- Algae (open pond) similar to tropical agriculture **30 t/ha/yr**
- Algae (photobioreactor) > **150 t/ha/yr**



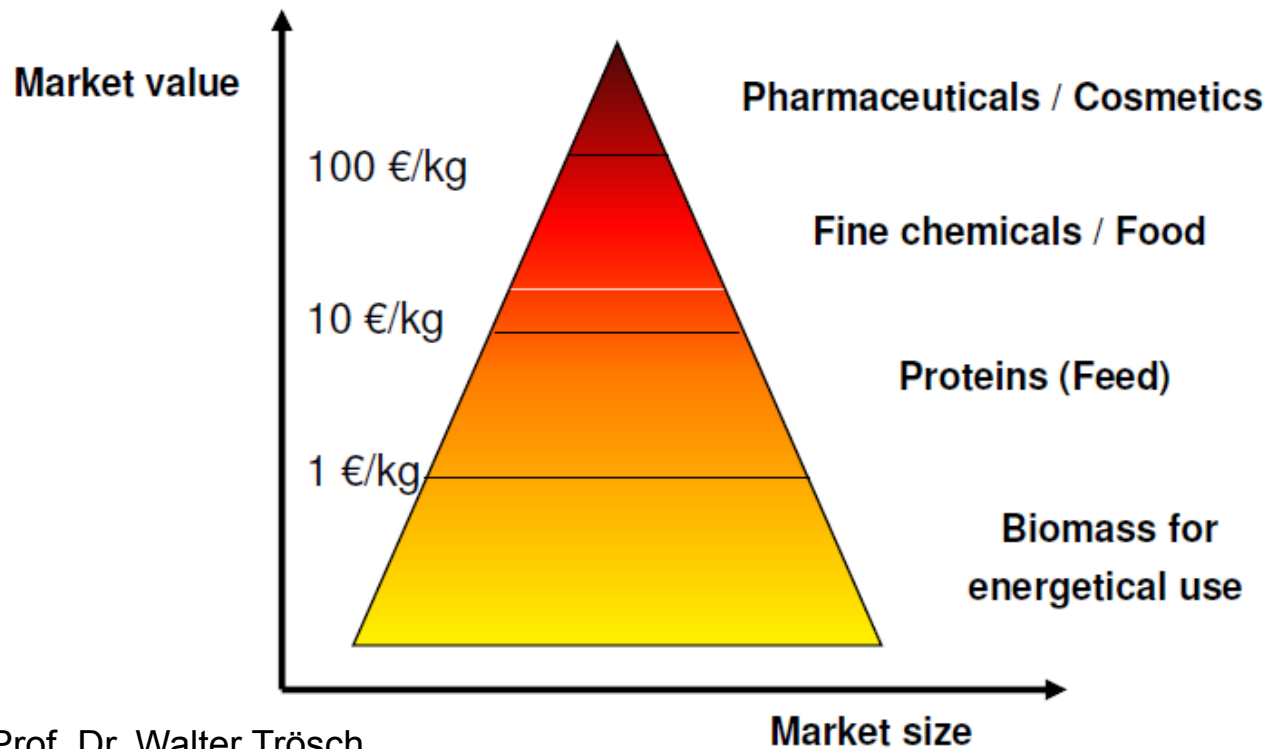
10 µm



Algae supplier: SAG

Poly generation pathways with production of **high cost chemicals**

- Lipids (PUFAs)
- Proteins
- Pigments (Carotenoids)



Why biomethane?

High diversity of use:

For Mobility, Household (SOFC), Plant (CHP)



1-4.6 kW_{el}
1.5-7 kW_{th}



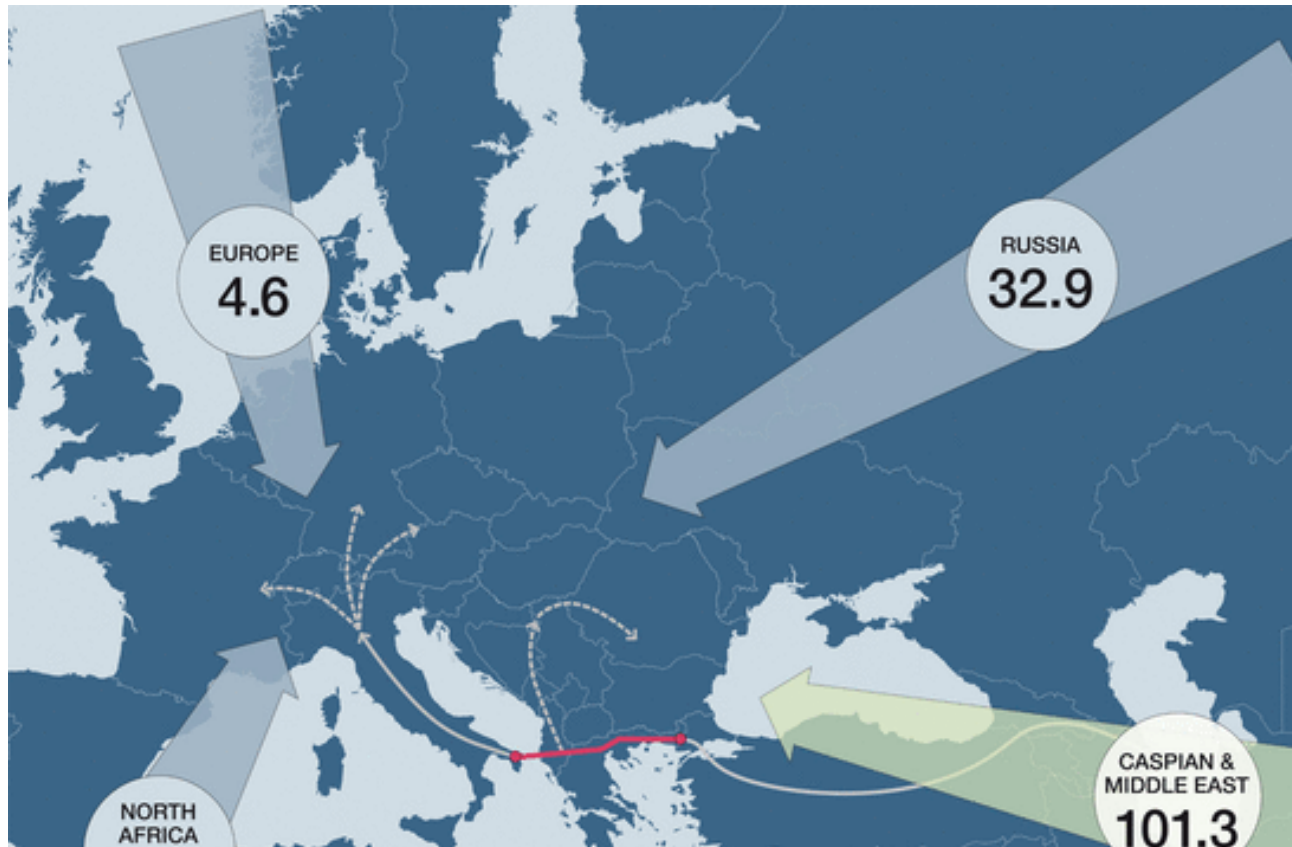
1 kW_{el}
2.5 kW_{th}



Already available and affordable!
139 g CO₂/km (gasoline car: 169 g CO₂/km)

Combined cycle power plant η_{el} up to 60%





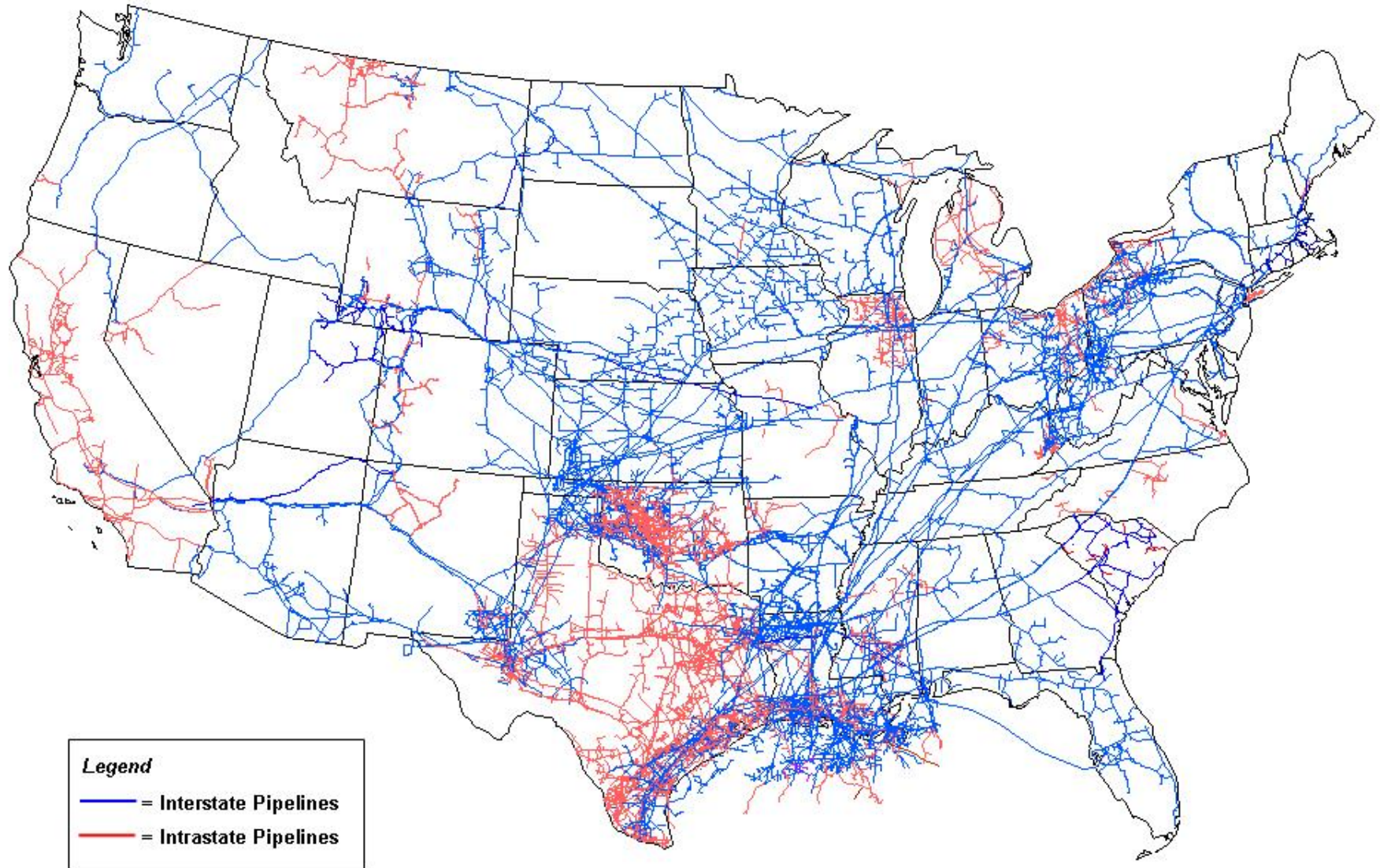
[in trillion m³]

Source: BP World Energy Statistical Review 2012

Construction

- Gas infrastructure is existing and will be extremely well developed in the near future
- Biomethane has a big potential!!!

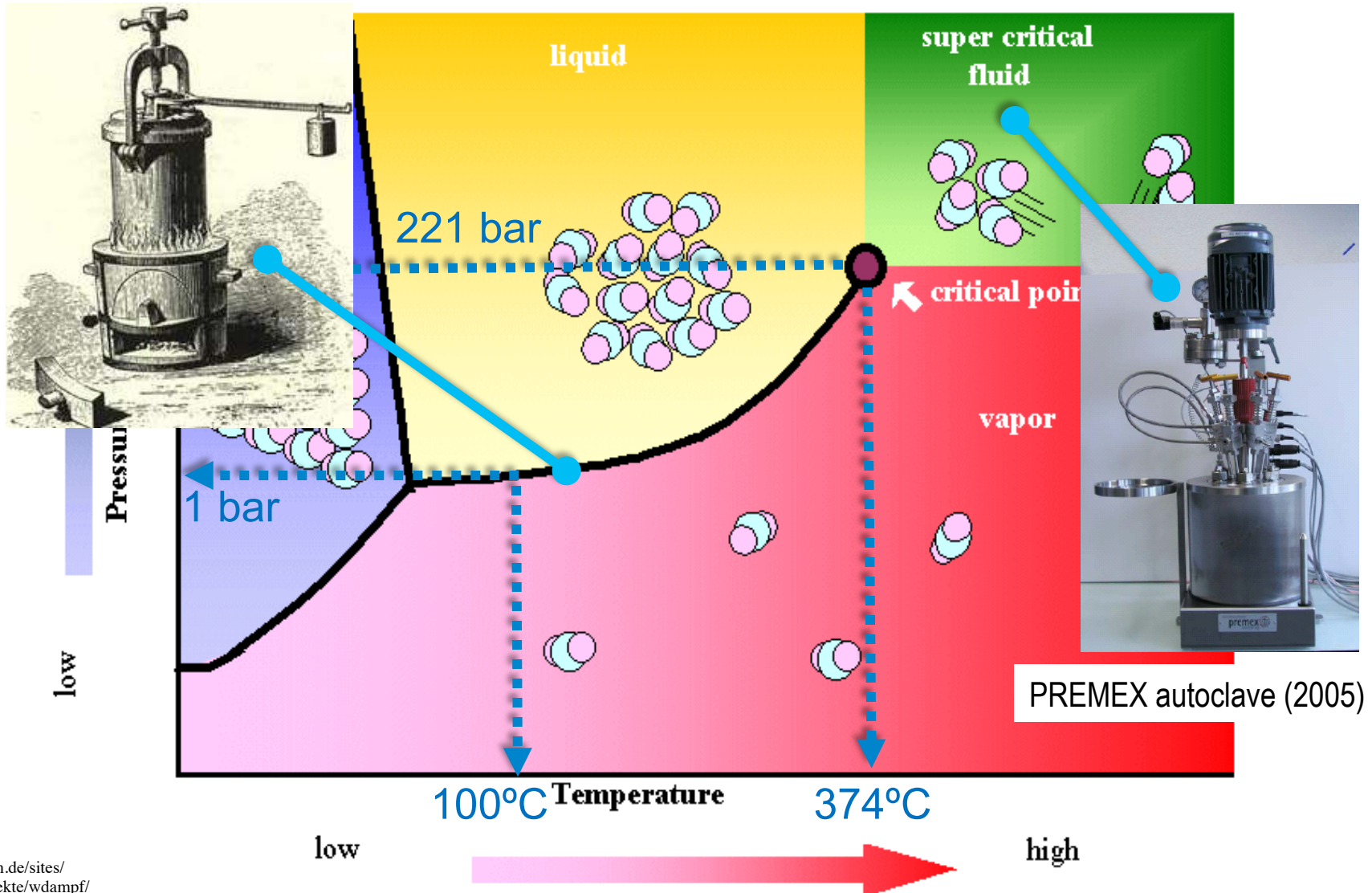
Distribution via natural gas grid in USA

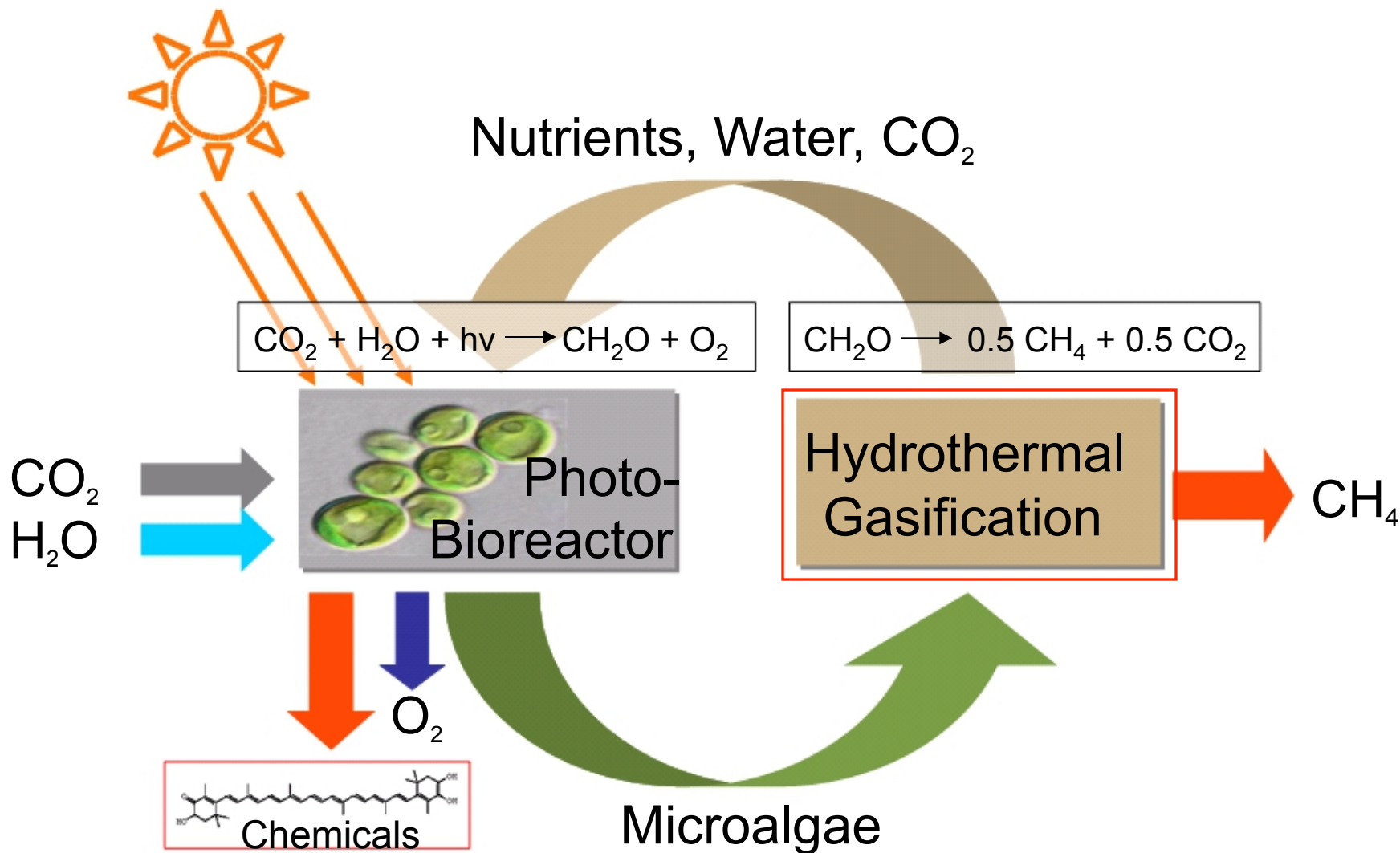


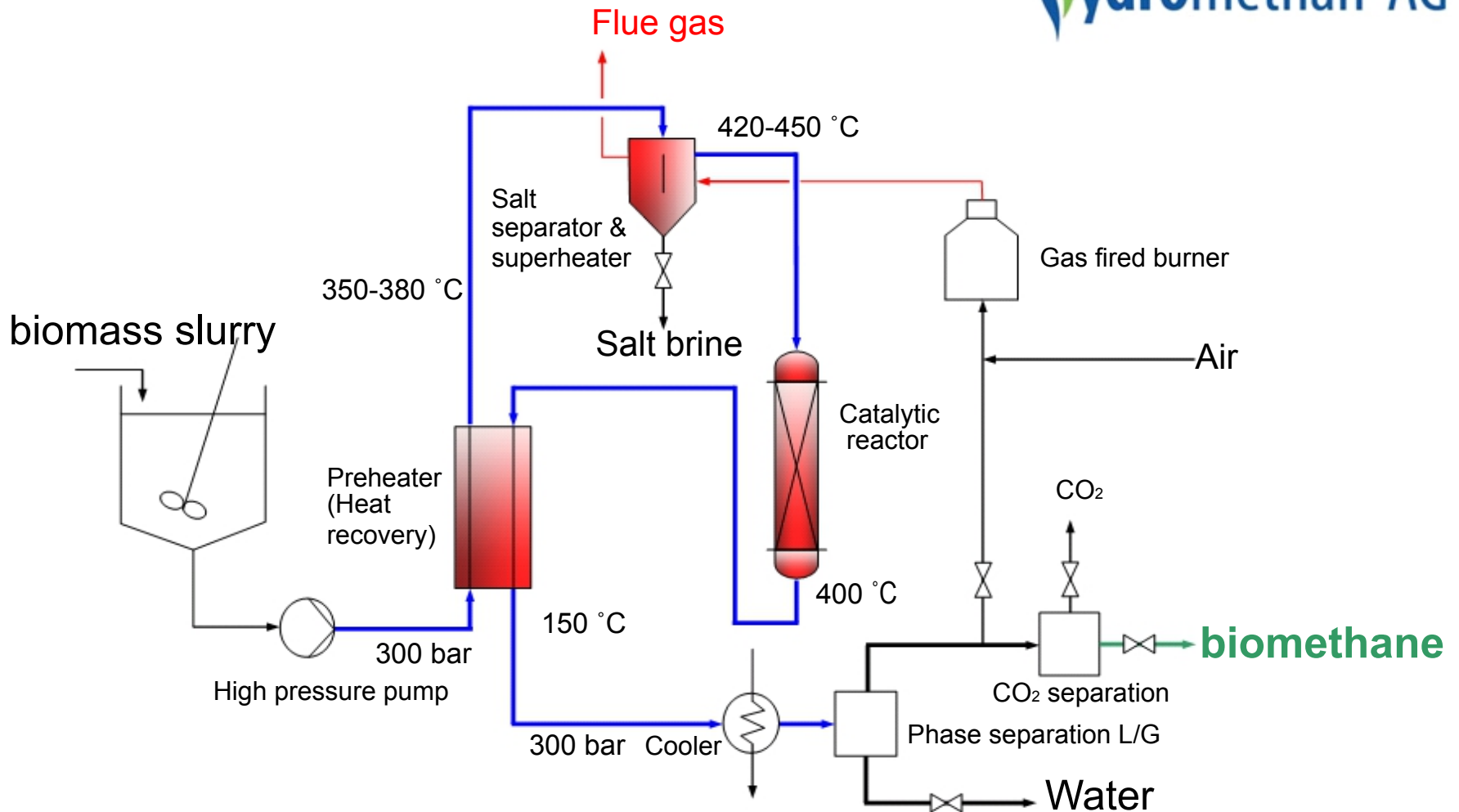
Characteristic	Hydrothermal gasification	Anaerobic digestion	Conventional gasification & methanation
Feed type	Most wet types ($m_{\text{water}} > 60 \%$)	Manure, Household residues, sewage sludge, marine algae	Wood, grass, ($m_{\text{water}} < 15 \%$)
Thermal efficiency (biomass to SNG)	62-71 % (Manure, wood)	25-35 % (< 8 wt % DM manure)	54-58 % (absolutely dry wood)
Residence time	< 30 min	20-22 days	< 10 min
Technological readiness	R&D	Very good (commercially available)	Good (PDU 1 MW _{SNG} in Güssing 2008)

The super pressure cooker: avoiding evaporation

Pressure cooker by Denis Papin (1679)







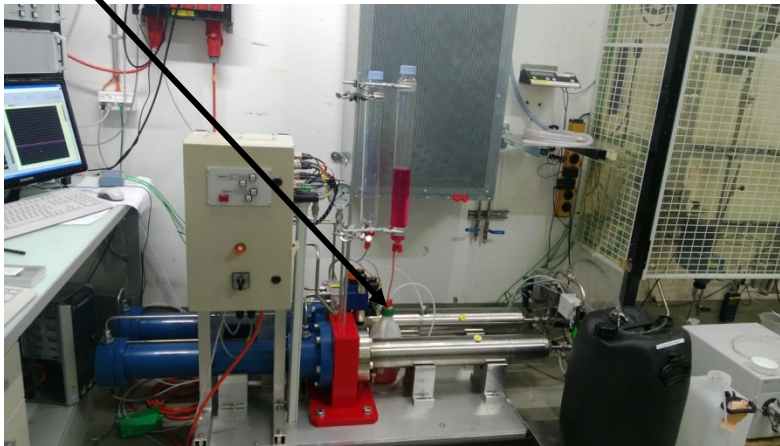


Catalytic reactor

Salt separator

Preheater

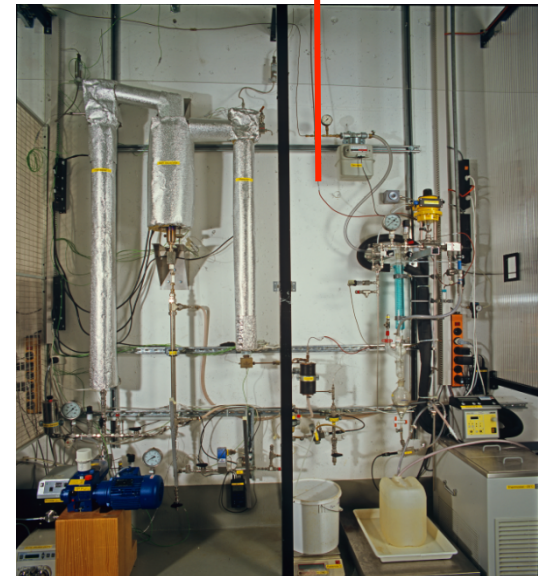
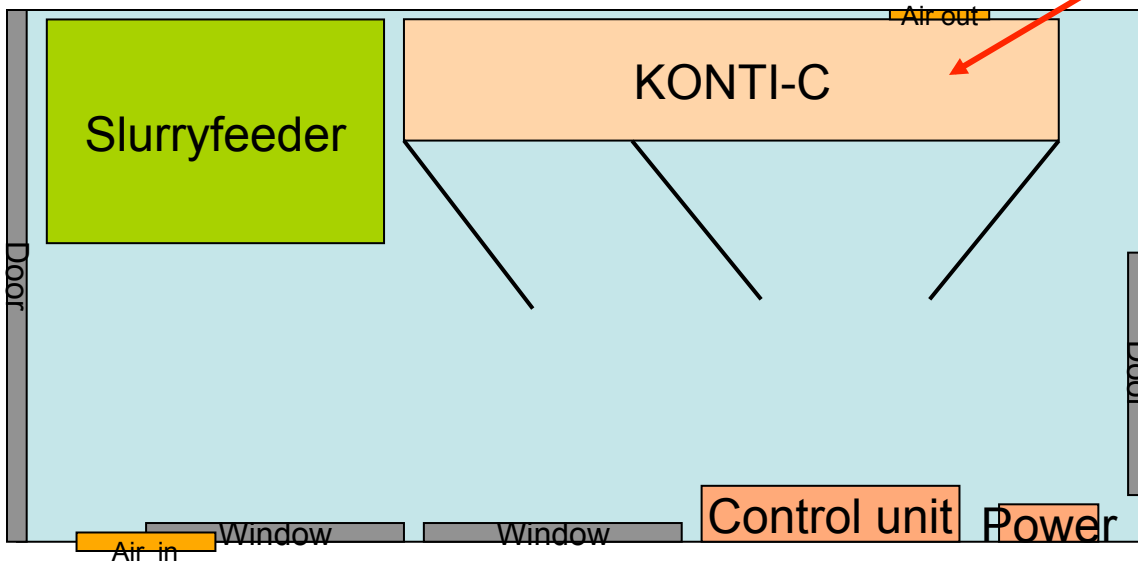
Slurry feeder

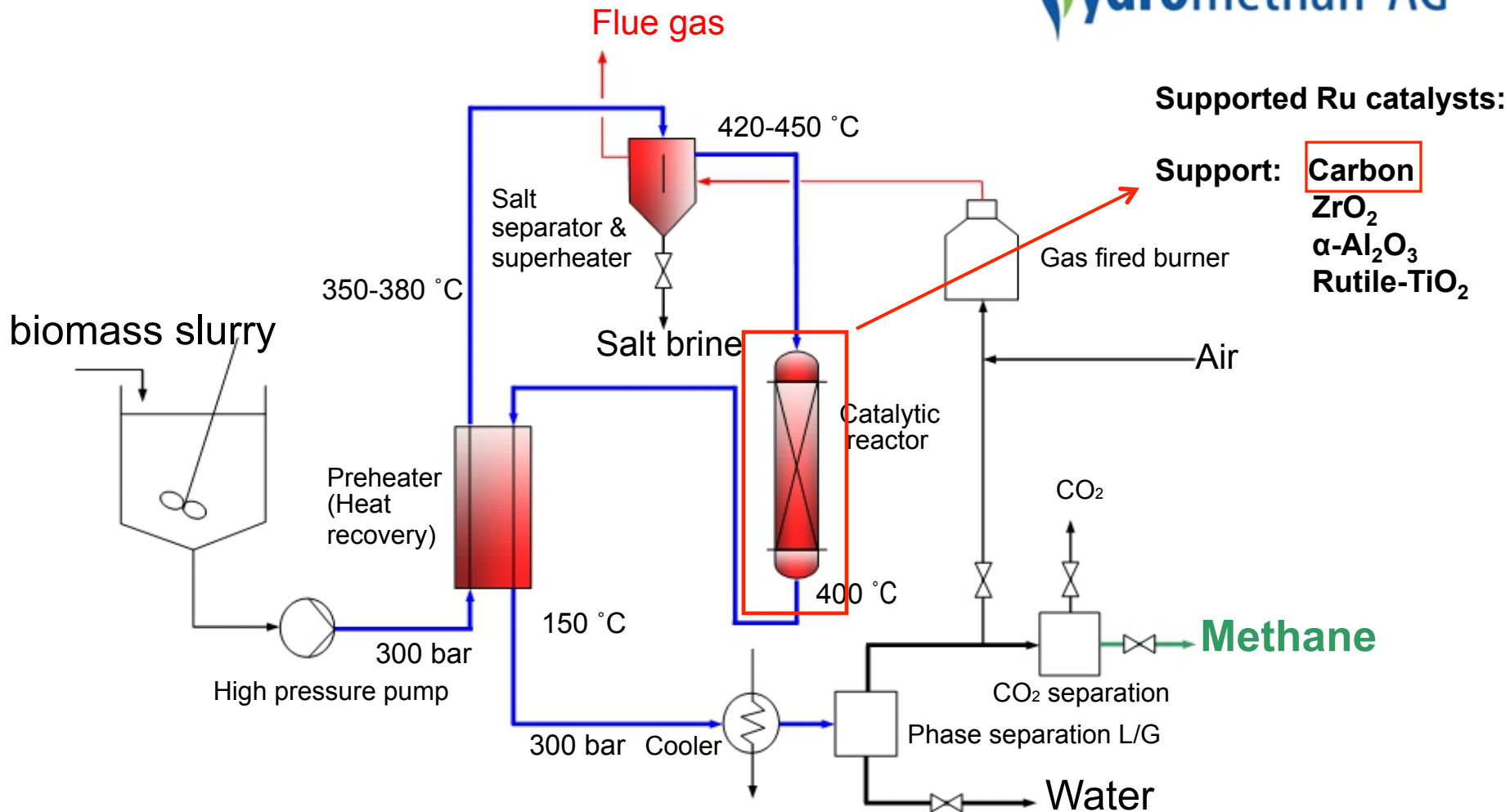


Container KONTI-C

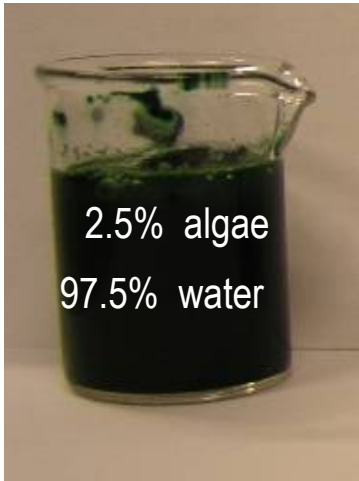


swisselectric
research

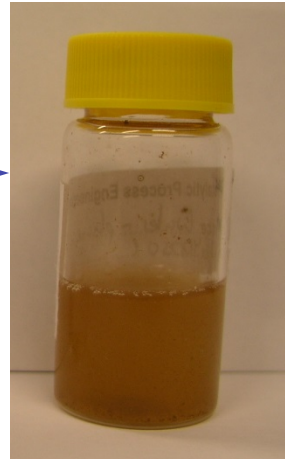




Experiment at 400 °C, 300 bar



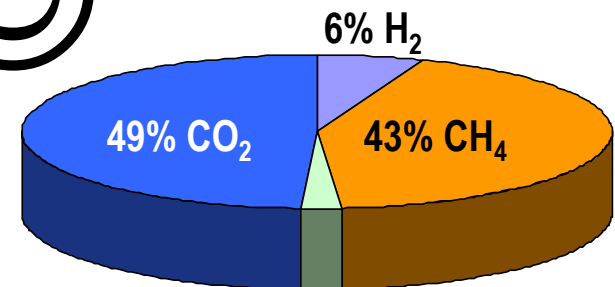
without catalyst



with catalyst



Composition of the product gas
(dry basis)

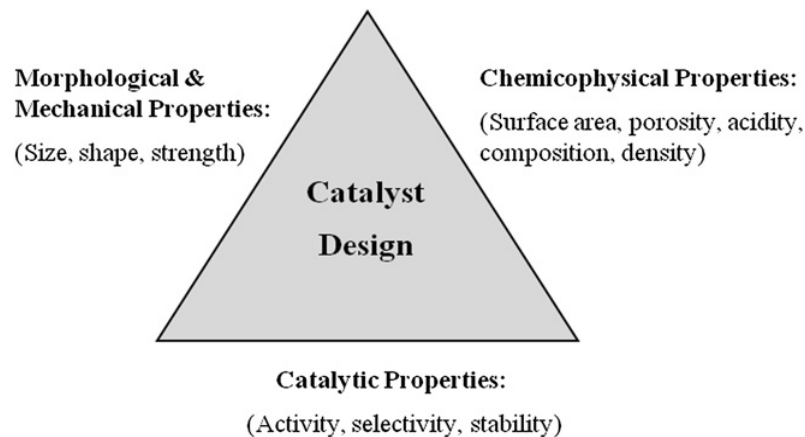


2% higher hydrocarbons (ethane, ...)

Batch reactor (54 ml)

« **Better understanding of the relationships between catalyst formulation and structure related to its catalytic performance** »

That's the key to further advance the effectiveness of the Ru/C catalysts for CSCWG!



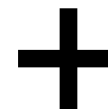
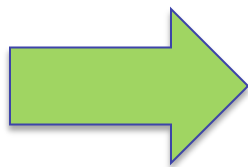
Relevant properties:

- Meso/microporous structure
- Impurities (**S**, P, Cl,...)
- Surface acidity
- Graphitization degree
- Ru dispersion
- Ru loading

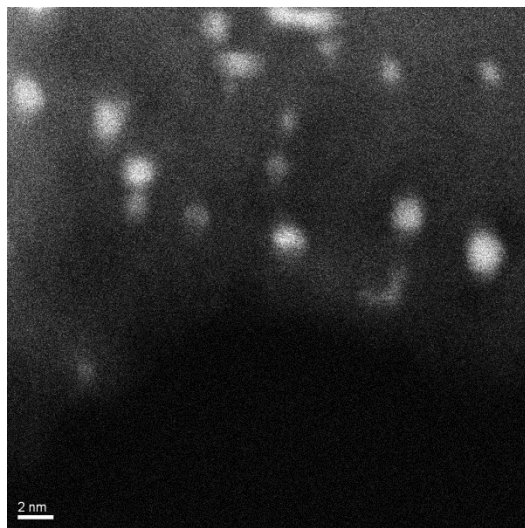
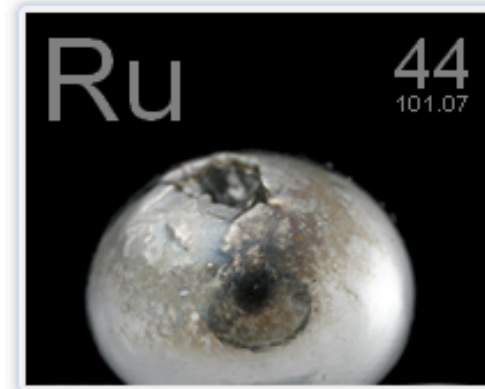
Synthesis of the Ru/C catalysts



Source: www.dampfkoenig.ch



Source: www.dfgoldsmith.com



HAADF-STEM image of the 2% Ru/C

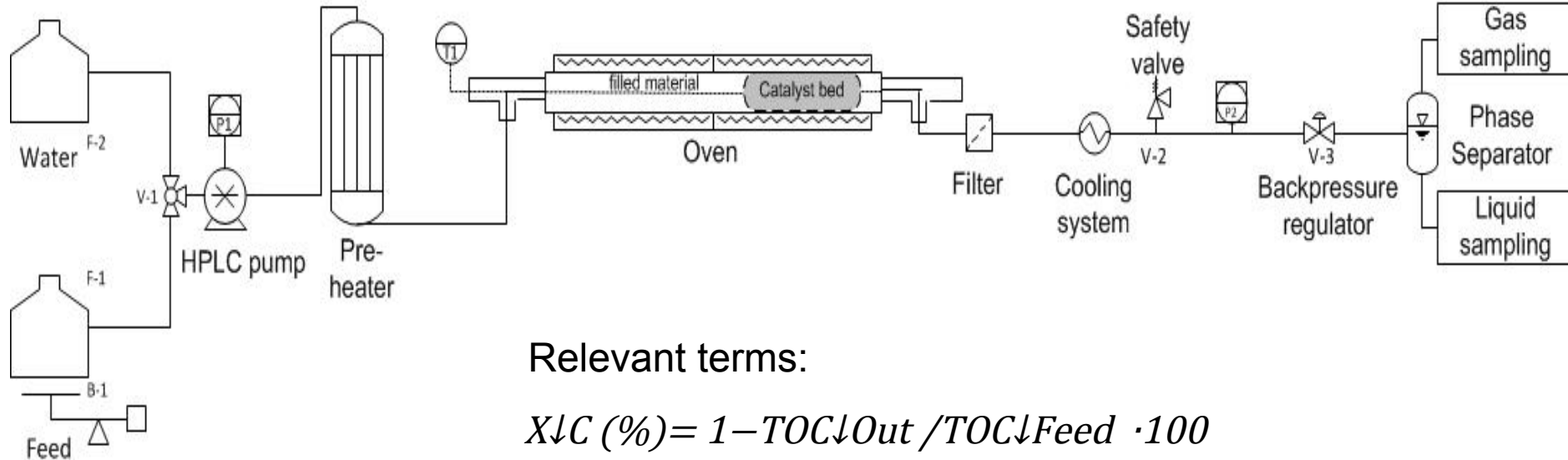
$\text{RuCl}_3 \cdot x\text{H}_2\text{O}$
or $\text{Ru}(\text{NO})(\text{NO}_3)_x(\text{OH})_y$
by wet impregnation



Objective

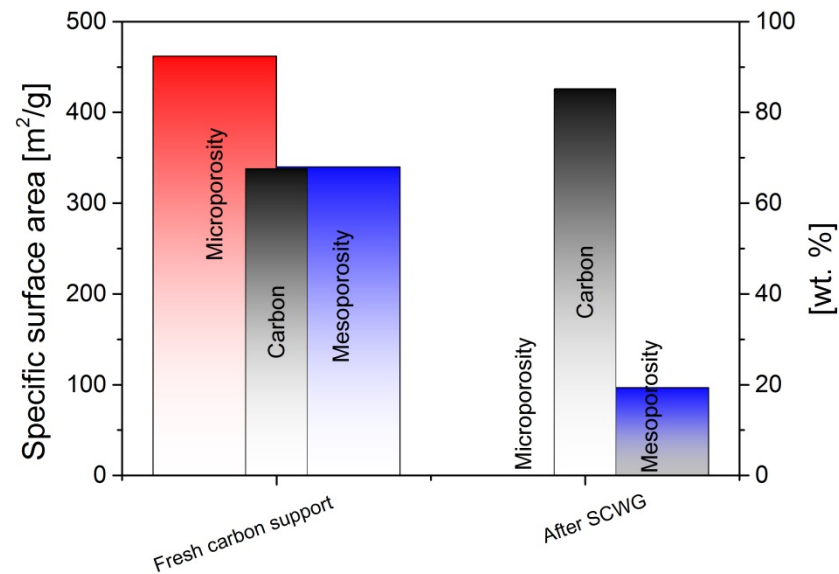
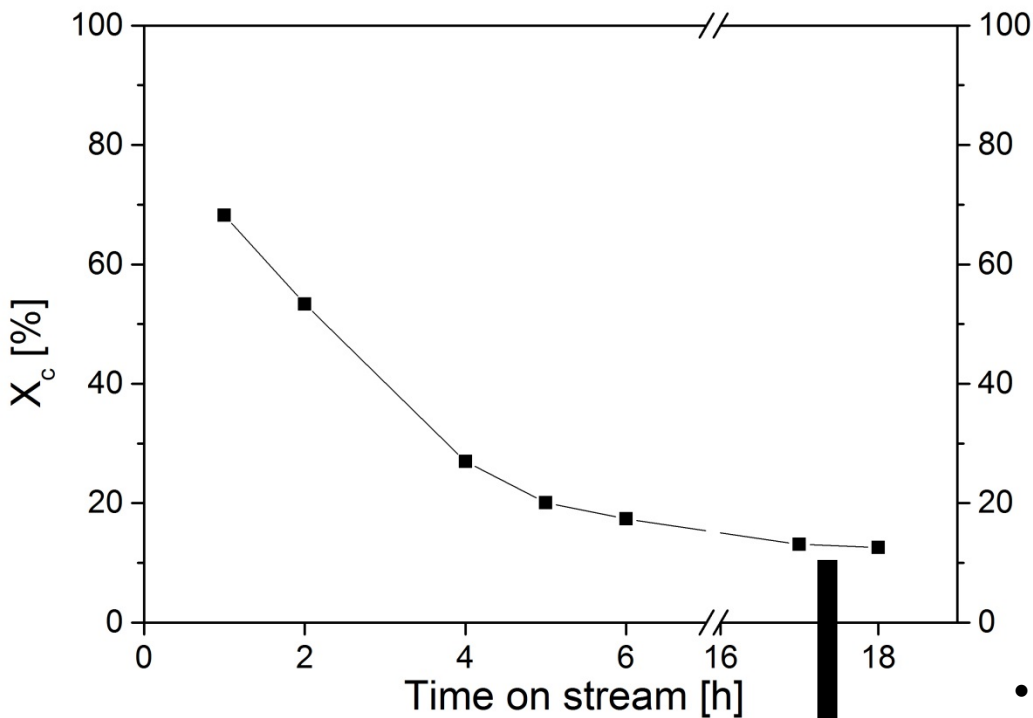
- Catalytic performances assessment of the Ru/C catalysts with 10 wt. % isopropanol

Working conditions: $T = 450\text{ }^{\circ}\text{C}$, $P = 300\text{ bar}$, $F = 3\text{ g/min}$



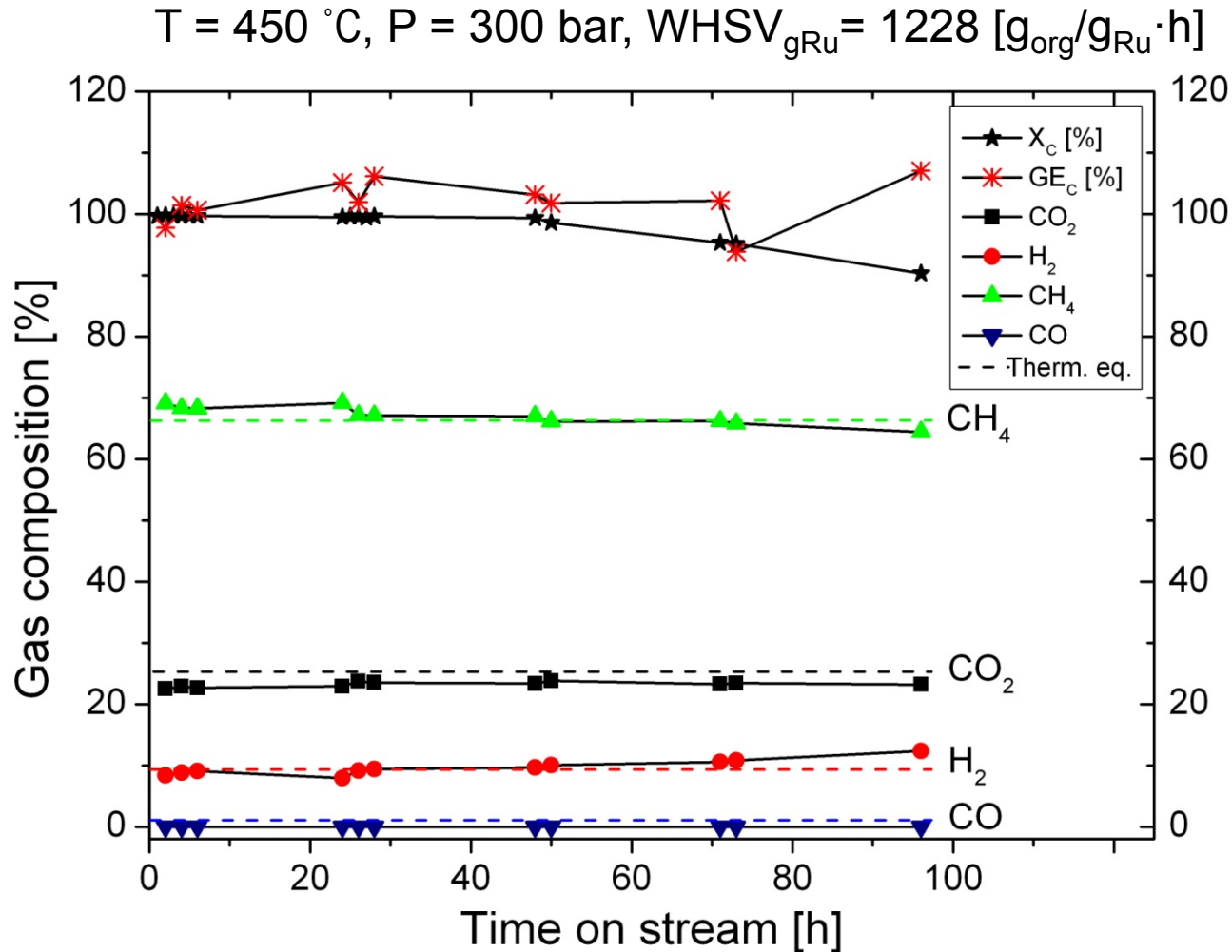
$$GE\downarrow C (\%) = Total\ molC\downarrow Gas / Total\ molC\downarrow Feed \cdot 100$$

T = 450 °C, P = 300 bar, 10 wt. % isopropanol



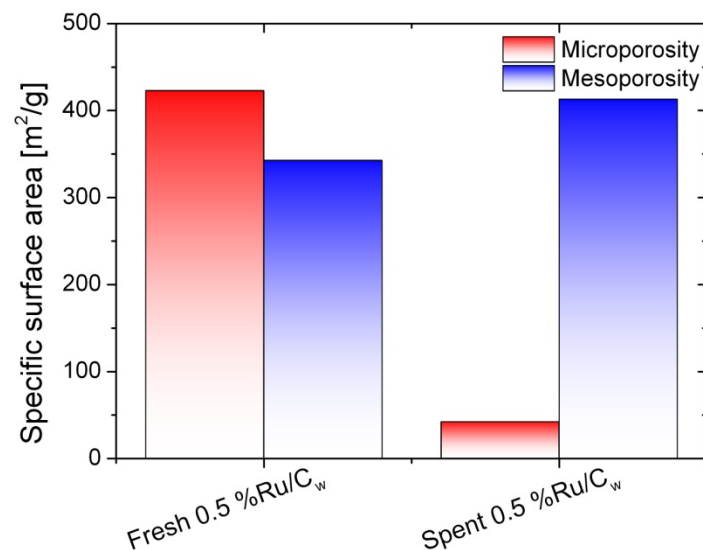
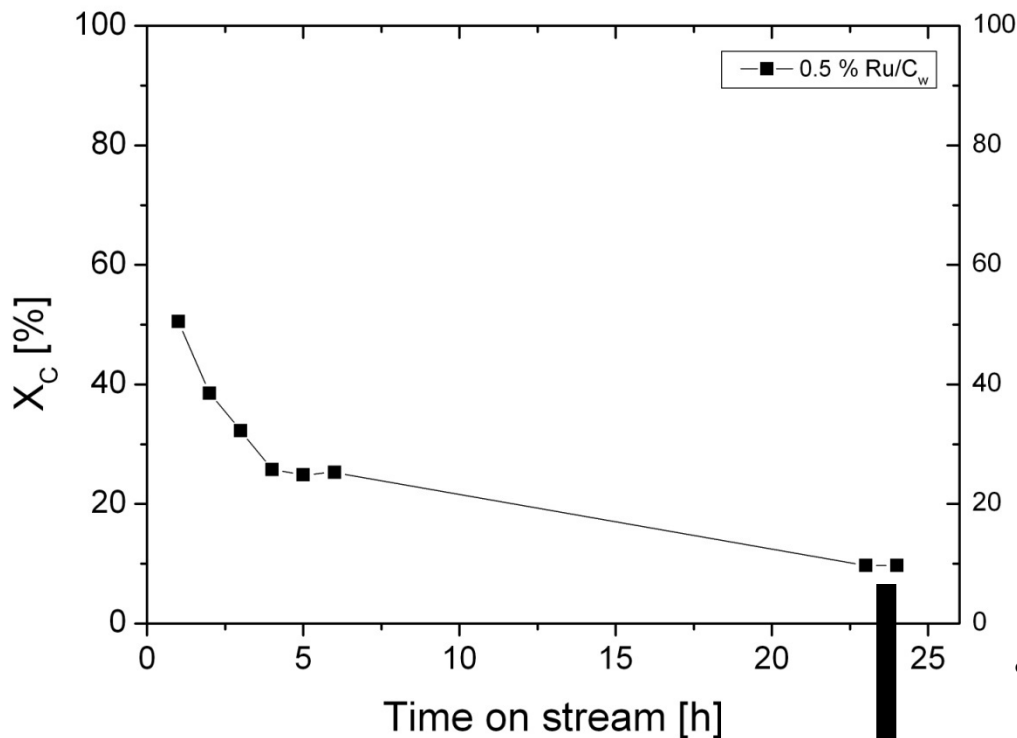
- Coke deposition during SCWG!

Time on stream [h]	X _C [%]	GE _C [%]	R _{Coke dep.} [mmol C/min]	Gas composition [vol. %]				
				CH ₄	CO ₂	H ₂	CO	C ₃ H ₈
18	11	3.6	1.1	3.3	0.5	96	0.2	5



- Good catalytic performances over 96 hours with gas composition close to thermodynamic chemical equilibrium

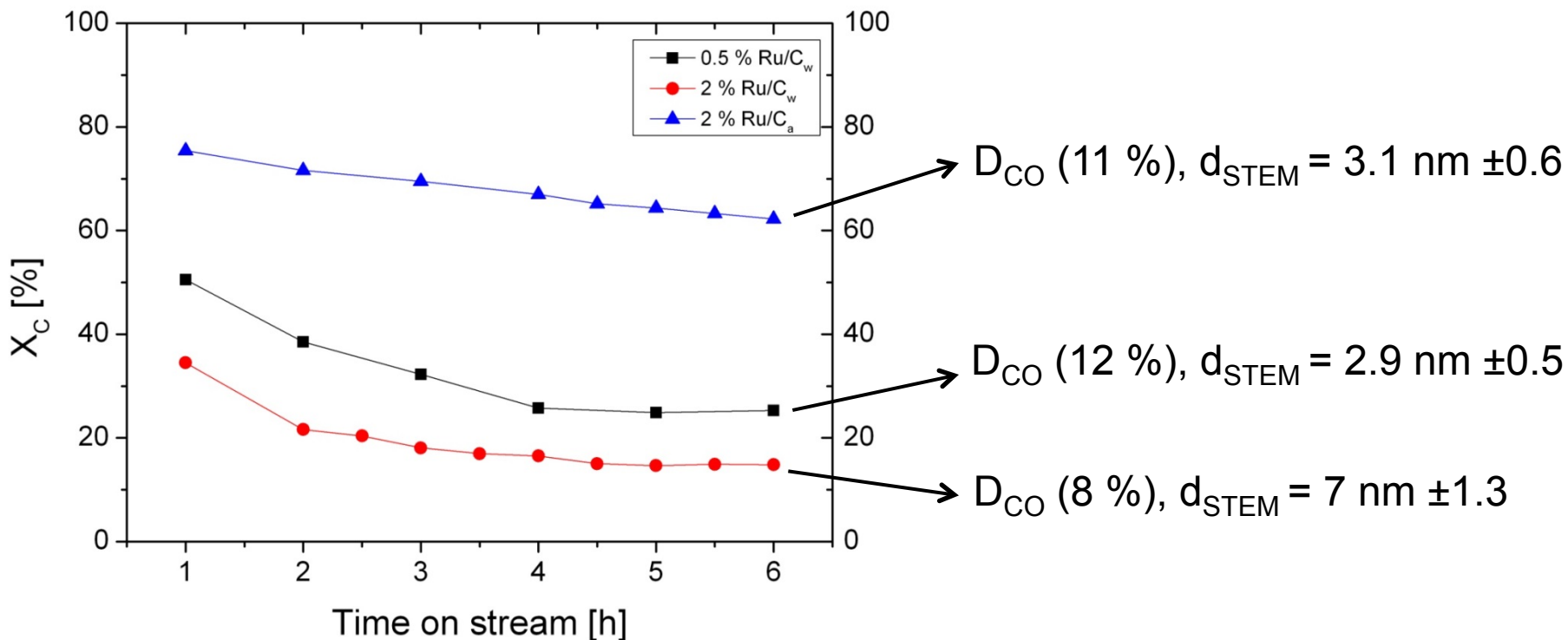
T = 450 °C, P = 300 bar, WHSV_{gRu} = 5202 [g_{org}/g_{Ru}·h]



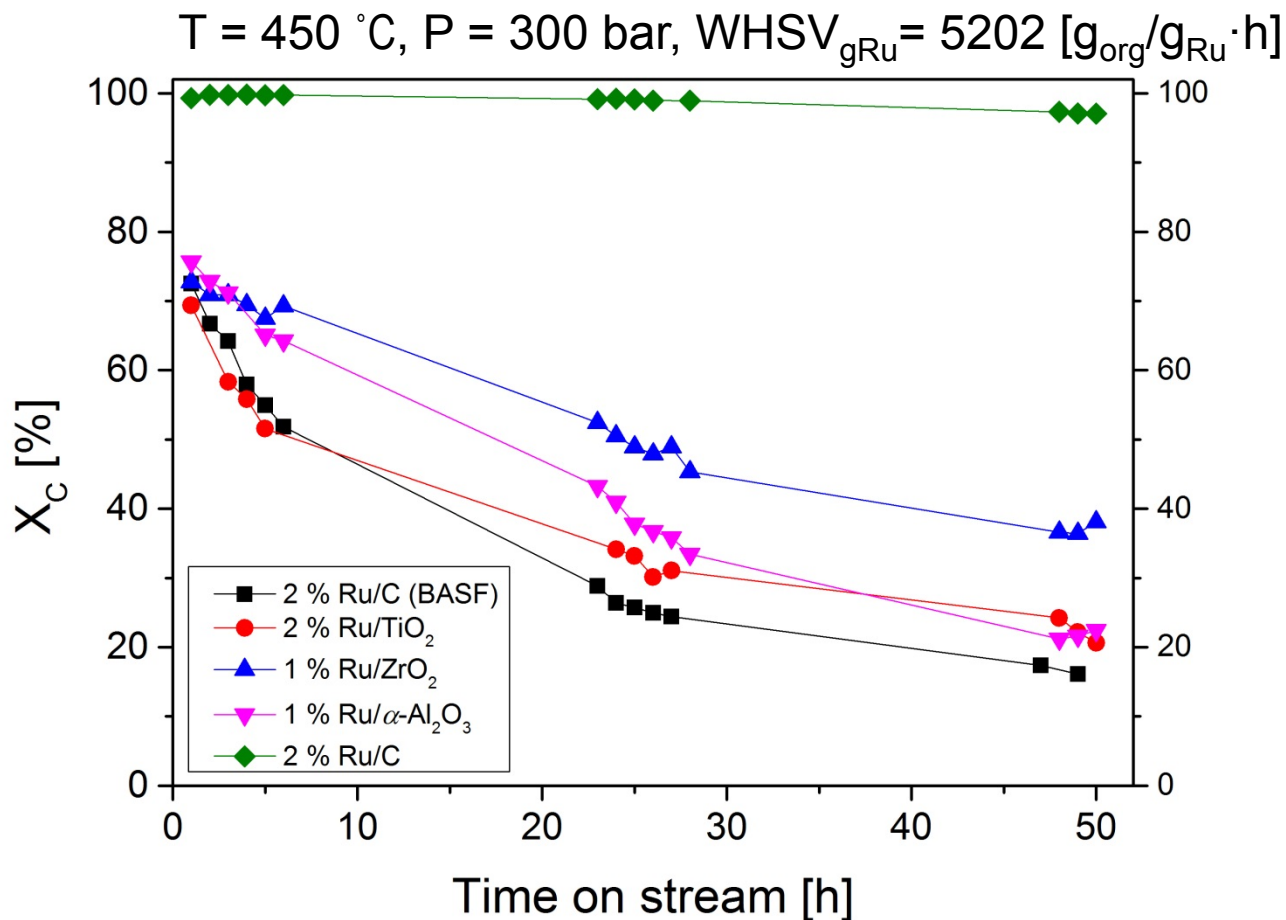
- Catalyst deactivation by coke deposits!

X _C	GE _C	R _{Coke dep.}	Gas composition [vol. %]				
			CH ₄	CO ₂	H ₂	CO	C ₃ H ₈
[%]	[%]	[mmol C/min]					
9.7	3.4	0.9	3.2	0.6	96	0.2	3.7

$T = 450\text{ }^\circ\text{C}$, $P = 300\text{ bar}$, $\text{WHSV}_{\text{gRu}} = 5202\text{ [g}_{\text{org}}/\text{g}_{\text{Ru}} \cdot \text{h}]$



- High Ru dispersion (smaller Ru NPs) lead to a better activity
- The use of acetone improved significantly the activity (higher Ru dispersion and a lower concentration of residual Cl⁻ coming from RuCl₃·xH₂O)



- The Ru/C catalyst is the most suitable catalyst
- Better catalytic performances than the 2% Ru/C (BASF) (use of $\text{Ru}(\text{NO})(\text{NO}_3)_x(\text{OH})_y$)

- Hydrothermal gasification process is a promising technology for biomethane production
- Remaining challenges are:
 - Efficient salt separation
 - Ensure stable working operations (plugging due to tar formation)
 - Sulfur removal by S-adsorbent agent
 - Develop robust catalysts for CSCWG (regarding coking, sulfur,...)