



Metasurfaces leveraging solar energy for icephobicity

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Critical issues in surface icing

Automobiles



edie.net

Windows



blog.brack.xyz

Photovoltaics



www.cleanenergyauthority.com



www.boldmethod.com



nipgroup.com

Structure

- **Introduction, motivation, and state-of-the-art in icephobicity.**
Economic figures and problems due to ice accumulation. Surface engineering of icephobic surfaces.
- **Metasurfaces for heat concentration with renewable energy.**
Rational surface engineering: Fabrication and characterization.
- **Demonstrating a new icephobicity technology based on metasurfaces.** Ice formation prevention or removal.

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Economic impact of ice formation and accumulation on surfaces

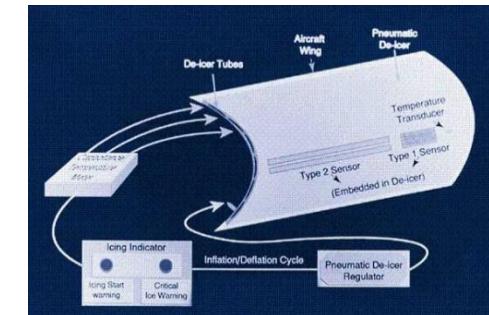
- Aircraft deicing market: \$1.30 billion (2020) [1]
- Global ice protection systems: \$10.17 billion (2021) [1]
- Vehicle windshield deicing: engine running for up to 30 min [2]



www.motorbeam.com



www.rt.com



sbir.gsfc.nasa.gov

[1] MarketsandMarkets Research Private Ltd. (2015, 2017)

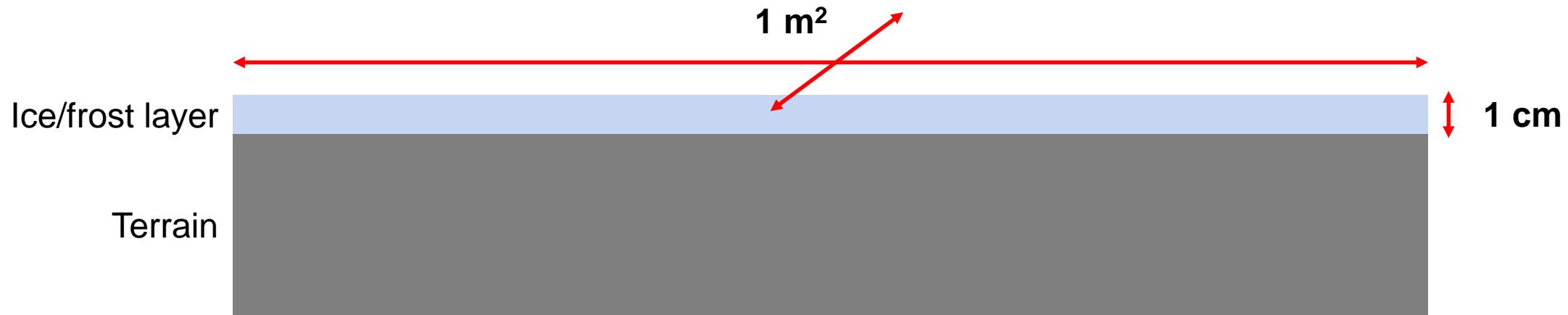
[2] Farag A & Huang L-J (2003)

Economic impact of ice formation and accumulation on surfaces

- “Severe winter weather caused 15 percent of all insured auto, home and business catastrophe losses in the United States in 2014.”
- “It costs an airline about \$6,000 to cancel a flight, according to masFlight. However, passengers also spend additional money as they accommodate their travel plans.”



Energy required to melt ice



Mass = volume x density (0°C) = $10^{-2} \text{ [m}^3\text{]} \times 916.7 \text{ [kg/m}^3\text{]} = \mathbf{9.2 \text{ kg of ice}}$

Energy = mass x enthalpy of fusion = $9.2 \text{ [kg]} \times 333.55 \text{ [kJ/kg]} = \mathbf{3070 \text{ kJ}}$

Energy required to melt ice

- Boeing 747 wing surface area: **524.9 m²**
- This leads to an energy consumption of **450 kWh** !



De-icing = energy-intensive task

Common de-icing methods

- Sodium chloride
- Mechanical scraping
- De-icing fluids

The city of Basel is using formates as thawing agents for de-icing.



cif.org



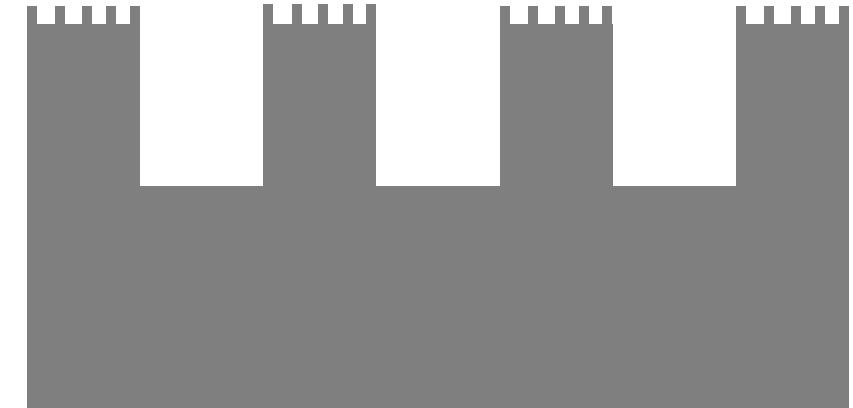
wikipedia.org

Icephobicity: State-of-the-art



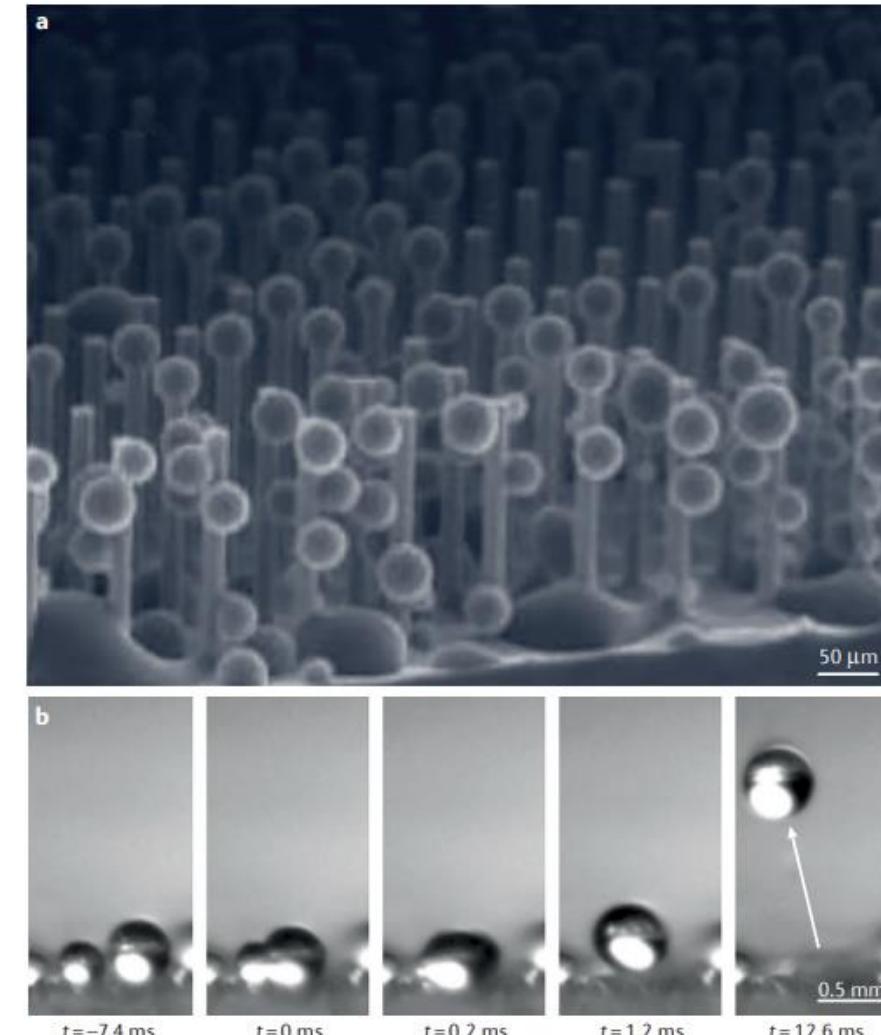
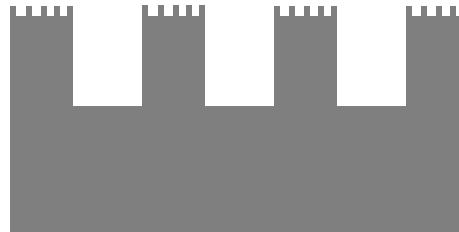
State-of-the-art passive icephobicity approaches

- Hierarchical superhydrophobic surfaces
- Lubricant infused surfaces



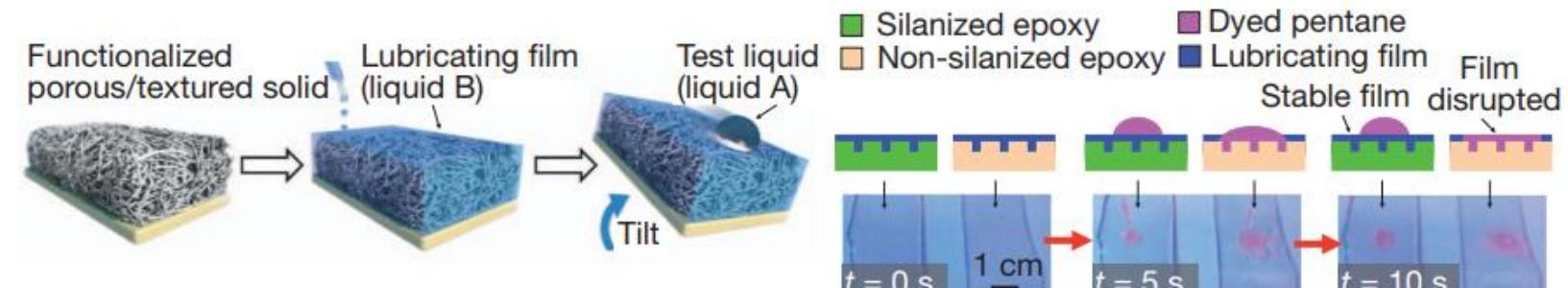
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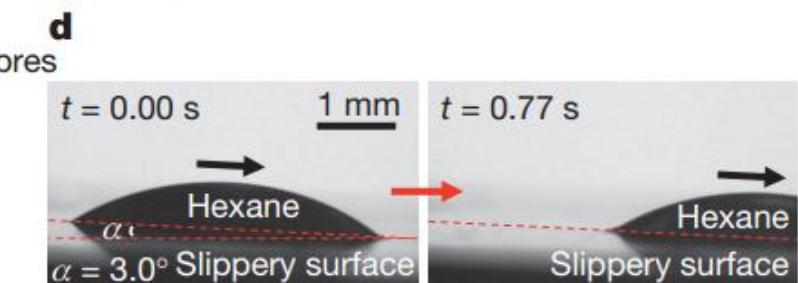
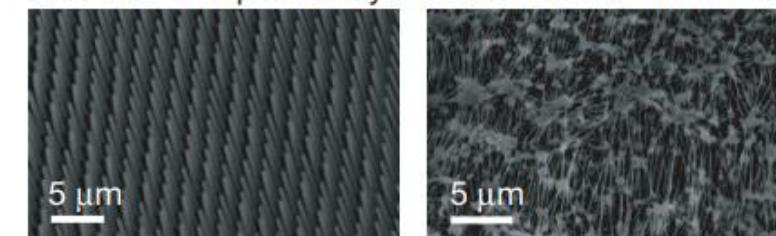


State-of-the-art passive icephobicity approaches

- Lubricant infused surfaces [3]

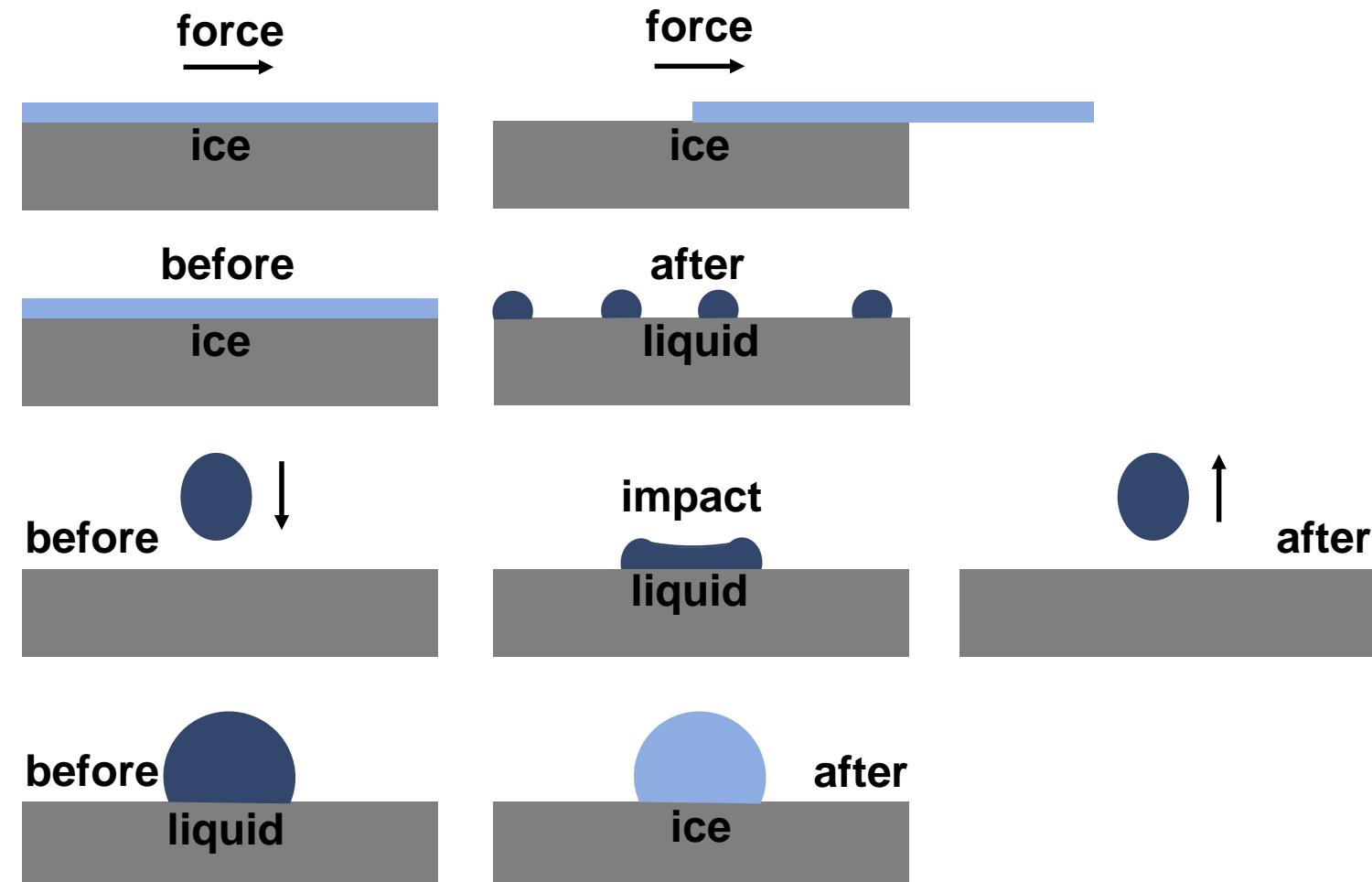


c
Ordered nano-post array Random network of nanofibres



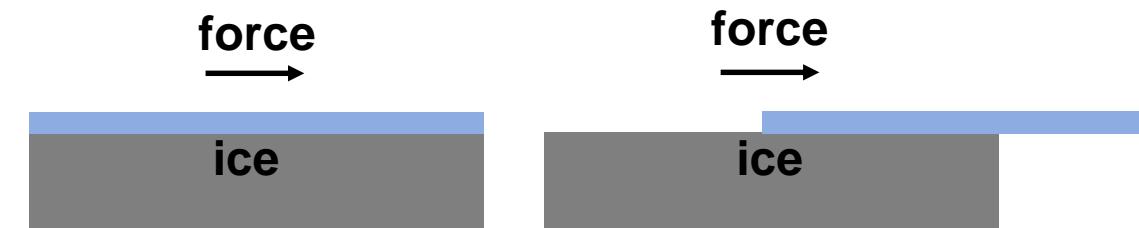
How do we define icephobicity?

- Reduced ice adhesion
- Reduced defrosting time
- Reduced droplet contact time
- Nucleation delay

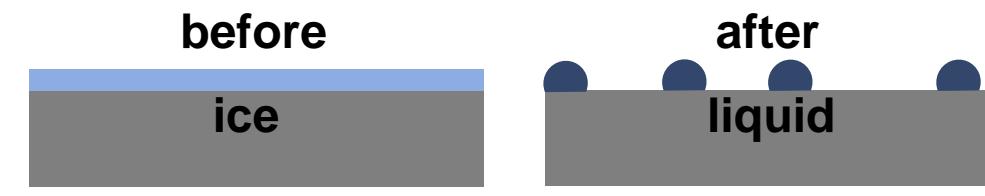


How do we define icephobicity?

- Reduced ice adhesion



- Reduced defrosting time

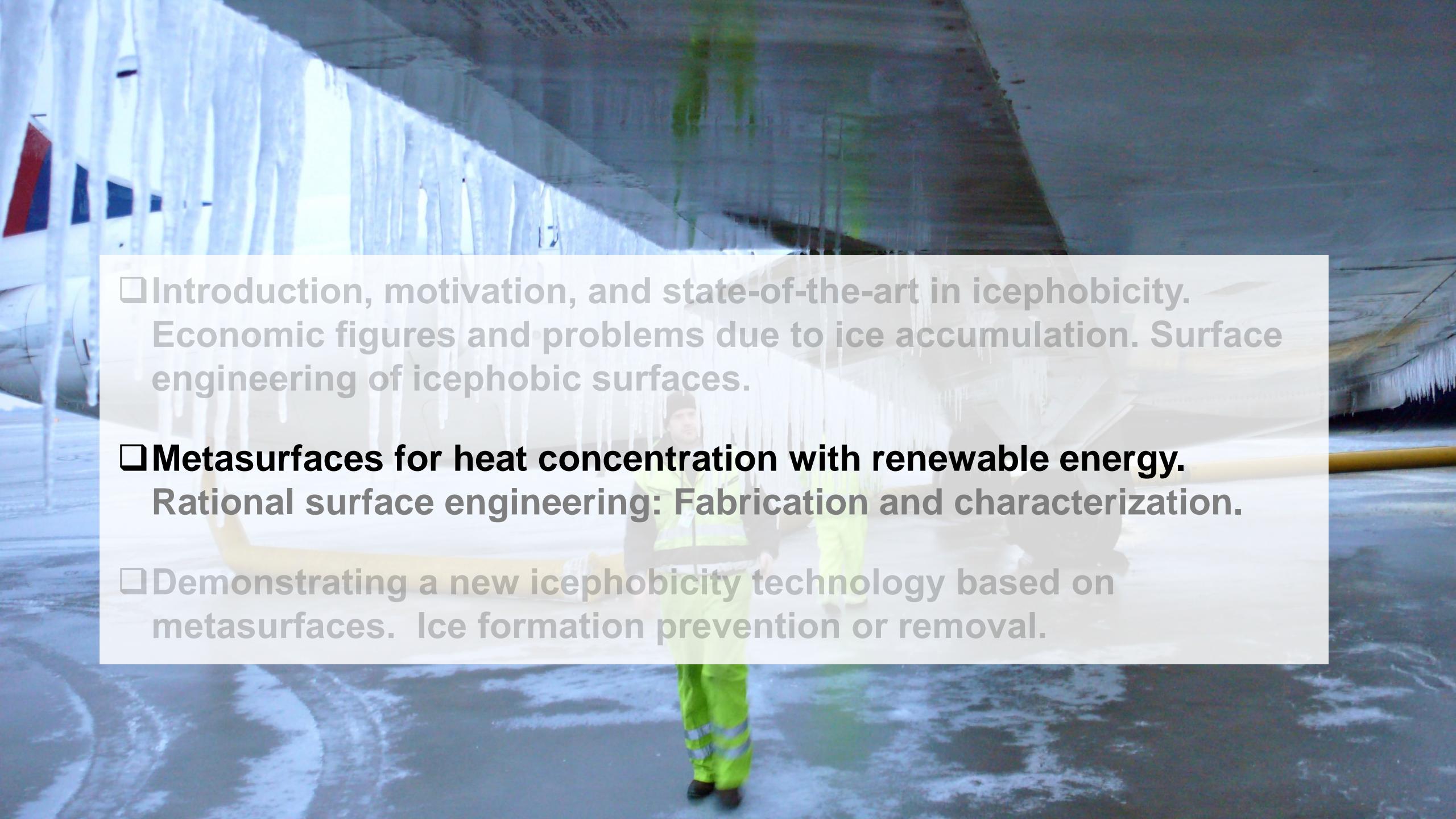


- Reduced droplet contact time



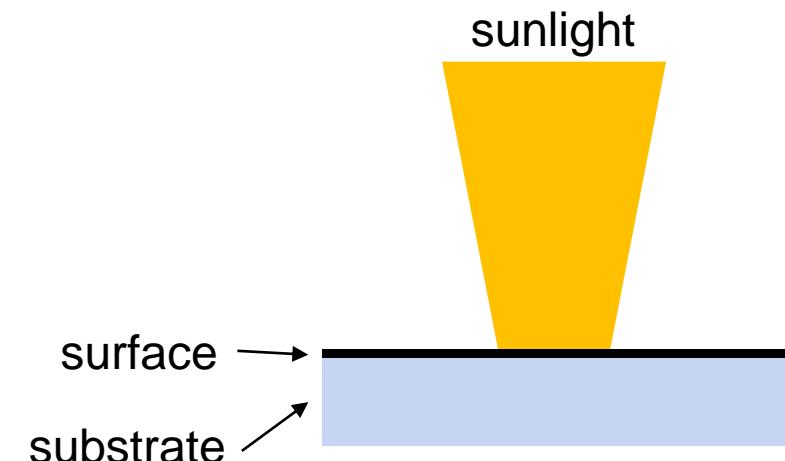
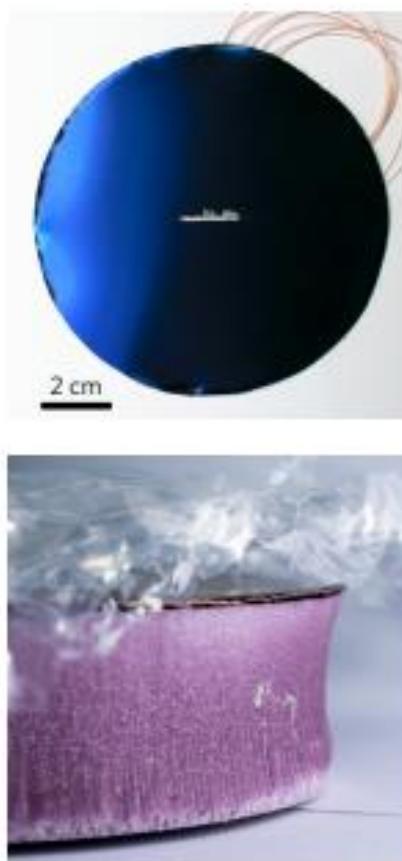
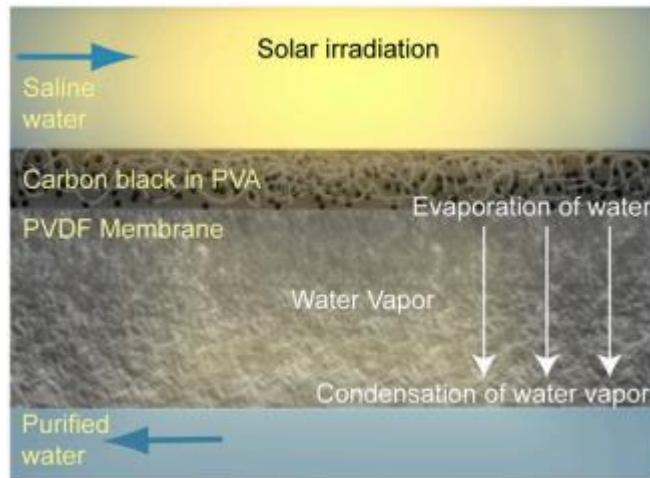
- Nucleation delay



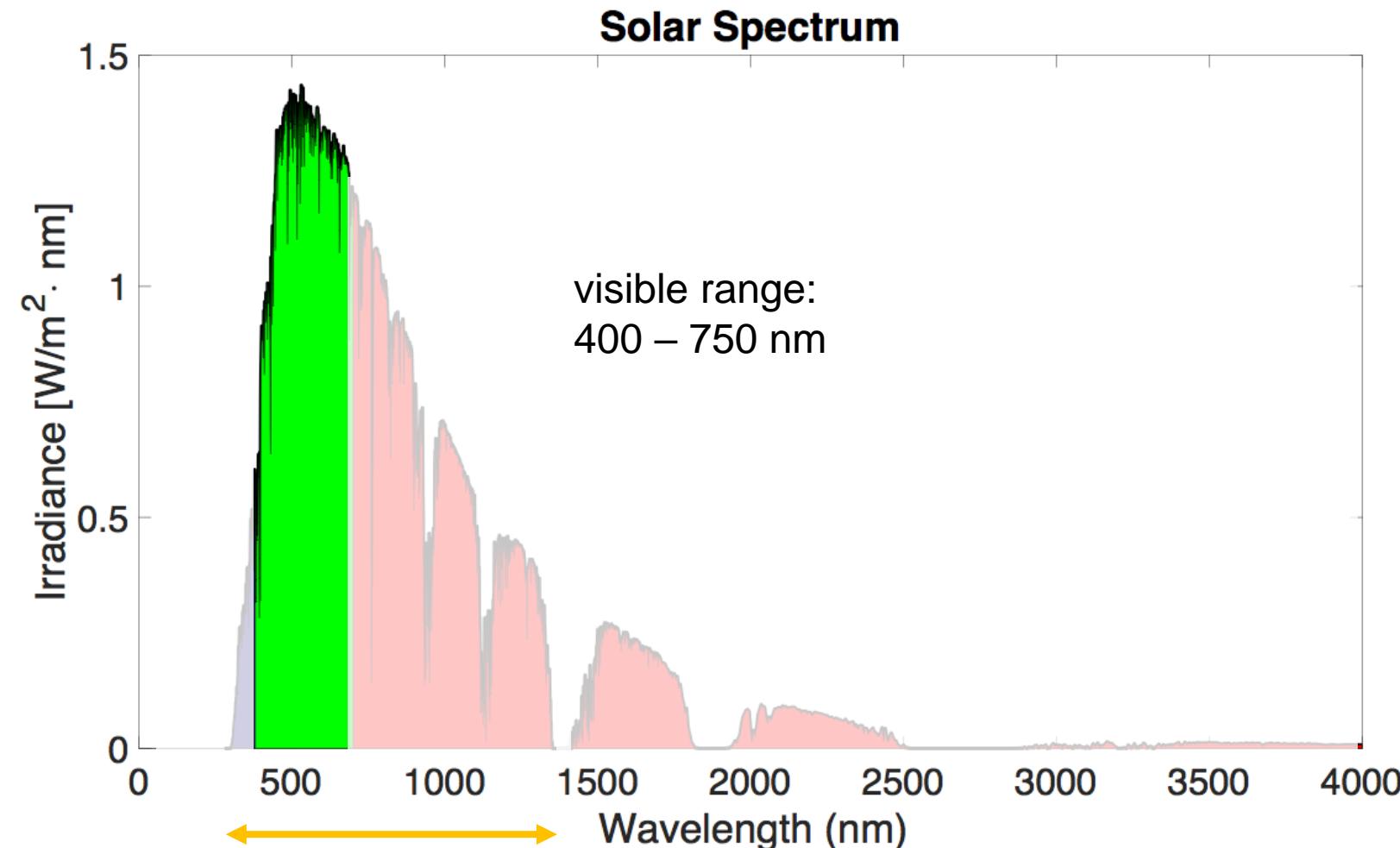
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Inspiration

- Solar steam generation [1]
- Solar-enabled desalination [2]



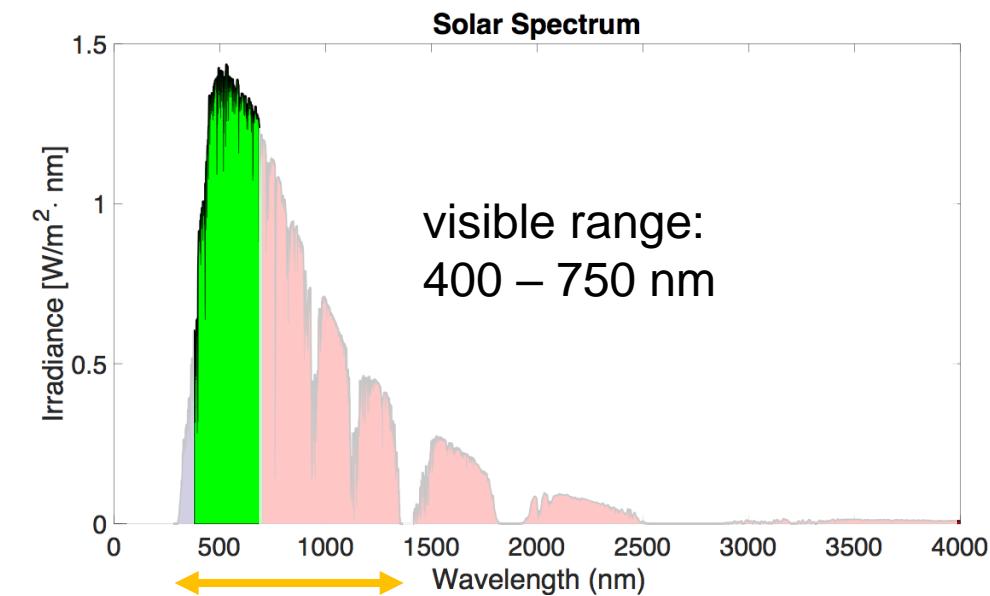
Metamaterials for solar energy absorption



Metasurfaces for solar energy absorption

Facts:

- ❑ We need surfaces that absorb sunlight.
- ❑ Absorption also in the visible range => loss of transparency.
- ❑ The absorption spectrum has to be as broadband as possible.



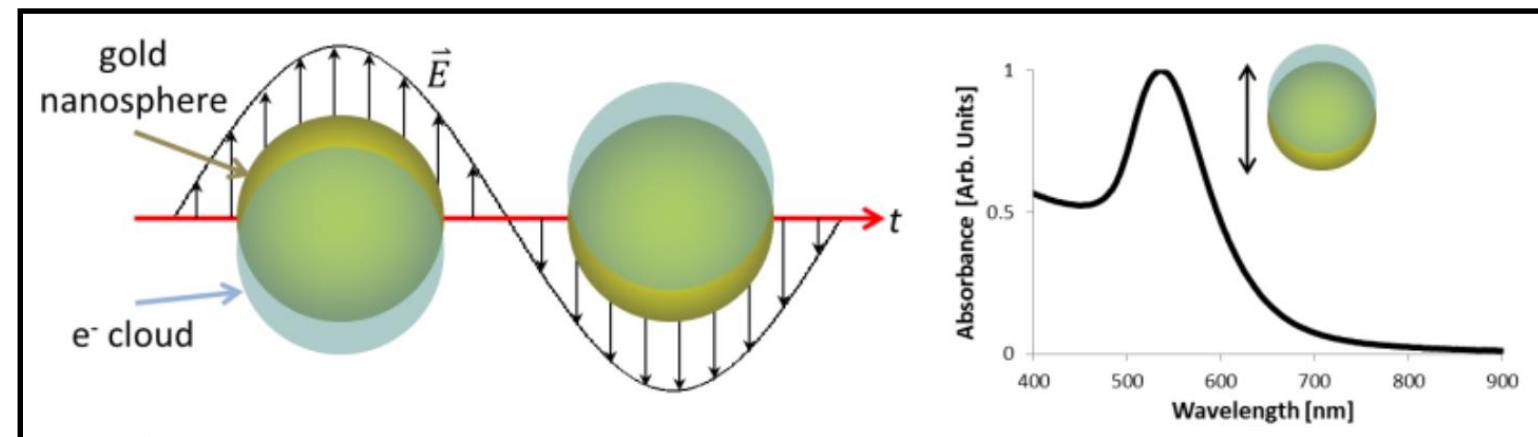
Metamaterials for solar energy absorption

Plasmonics: The most efficient way of creating sunlight absorbers that are ultra-thin at the same time.

Hint: Their properties cannot be found in natural materials.

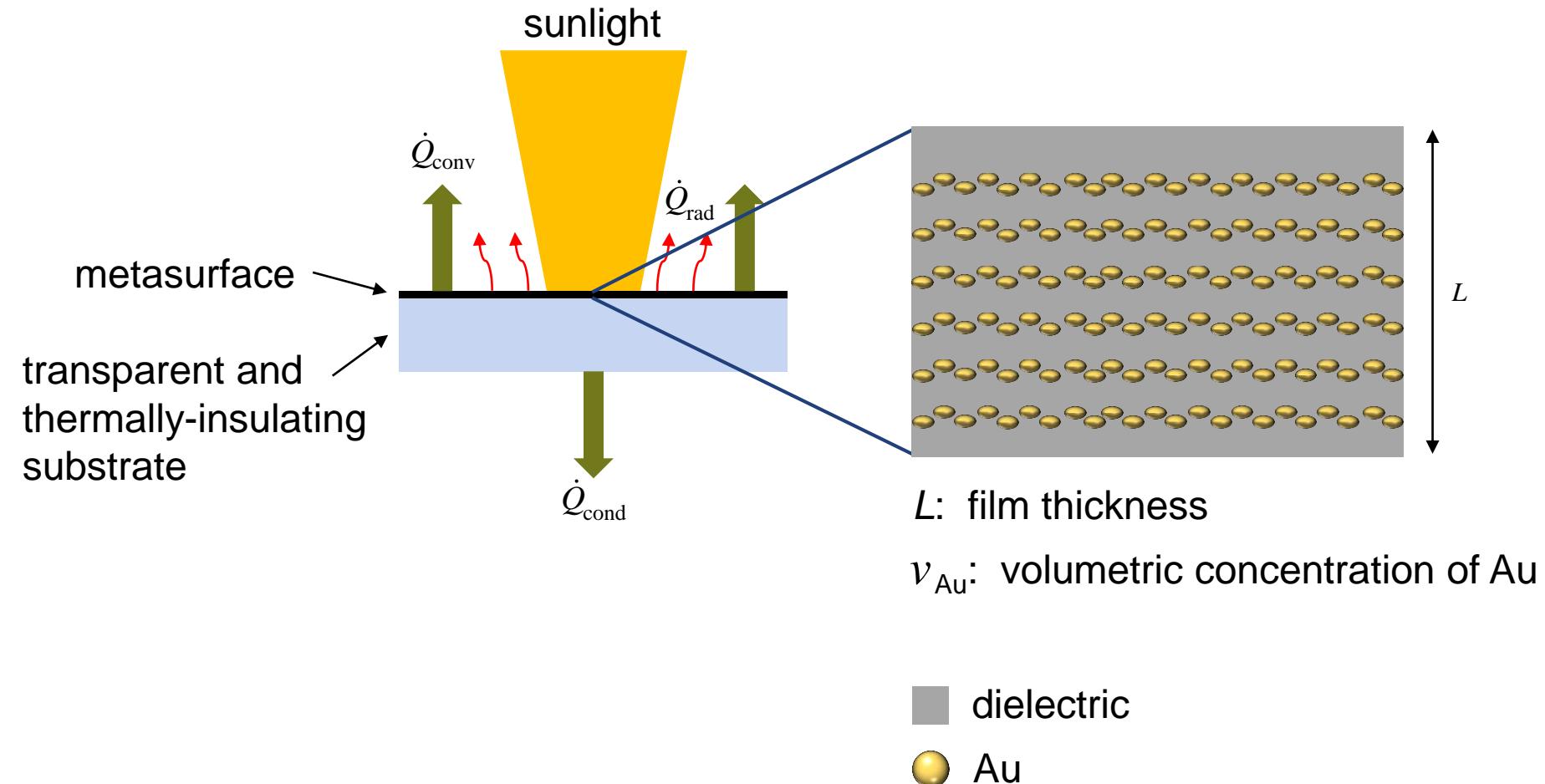
Material: **Gold** (Au) nanoparticles

**Localized
Surface
Plasmon
Resonance
(LSPR)**



scholar.harvard.edu/ndurr/pages/multiphoton-luminescence-imaging

Design: Metamaterials for solar energy absorption



Design: Metamaterials for solar energy absorption

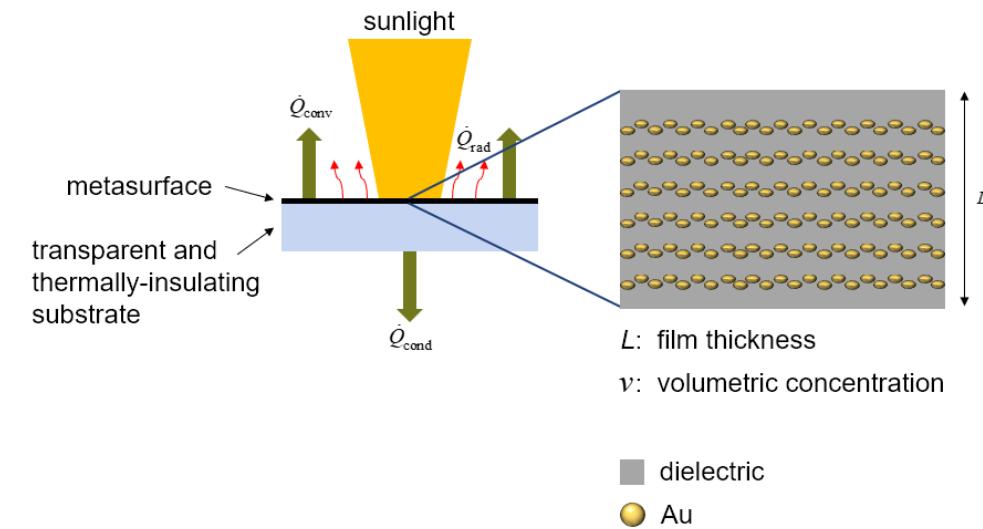
Bruggeman effective medium theory:

$$\varepsilon_{r,nc} = \frac{1}{4} \left\{ \frac{(3v_{Au} - 1)\varepsilon_{r,Au} + (3v_{diel} - 1)\varepsilon_{r,diel}}{\sqrt{[(3v_{Au} - 1)\varepsilon_{r,Au} + (3v_{diel} - 1)\varepsilon_{r,diel}]^2 + 8\varepsilon_{r,Au}\varepsilon_{r,diel}}} \pm \right\}$$

$\varepsilon_{r,nc}$: effective permittivity of the nanocomposite

Facts:

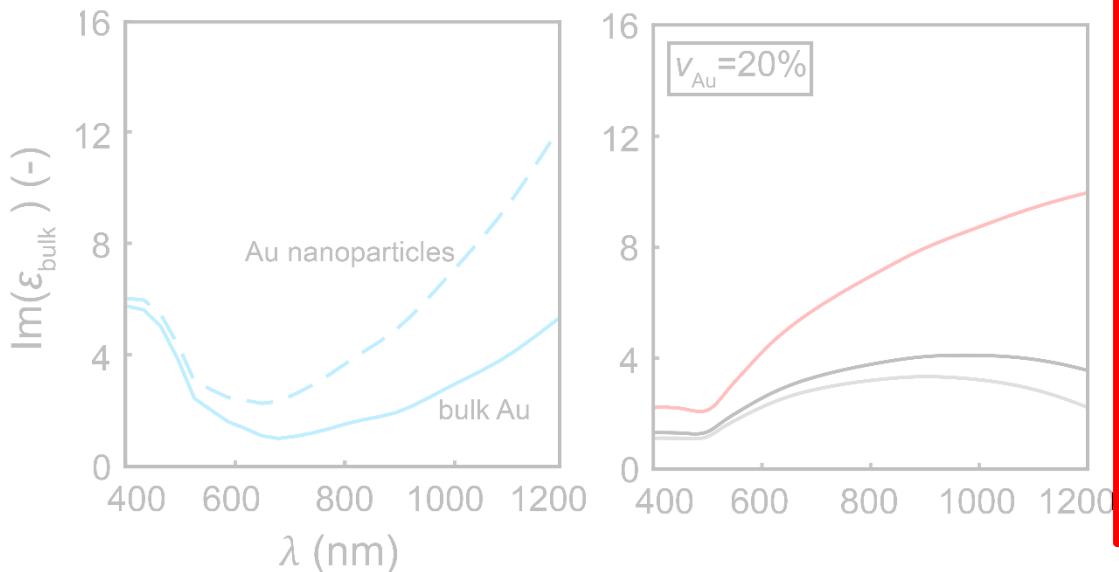
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- Absorption also in the visible range.
- The absorption has to be as broadband as possible.



Design: Metamaterials for solar energy absorption

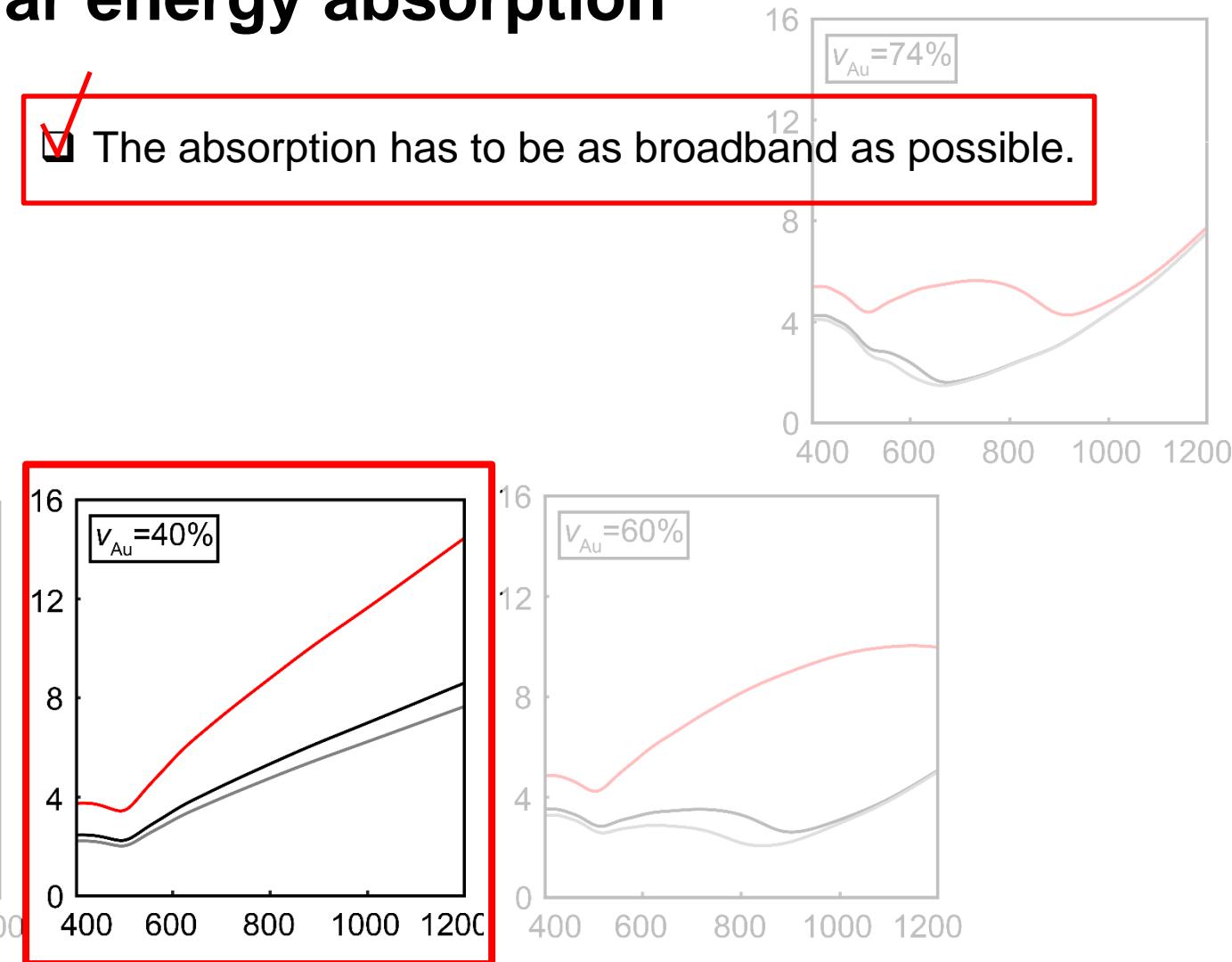
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λ : wavelength of light

✓ The absorption has to be as broadband as possible.

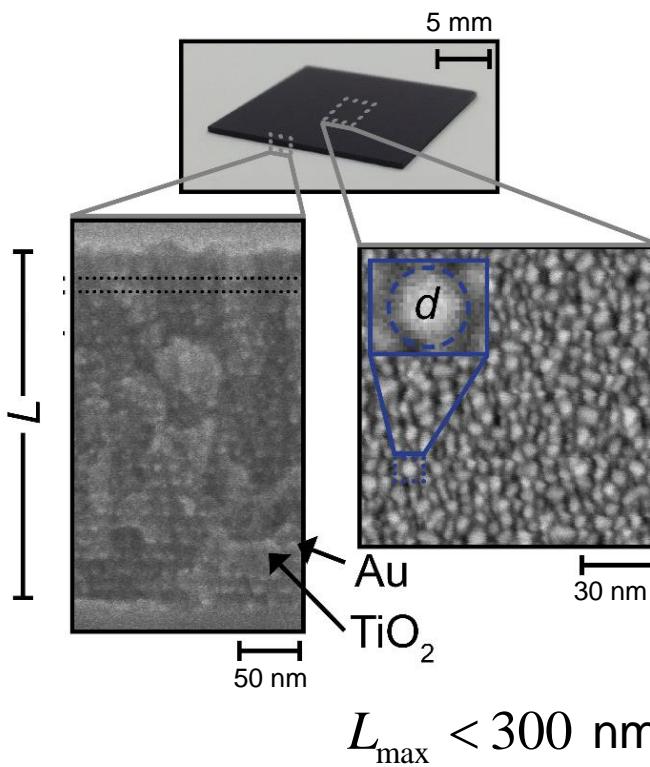


Metasurface engineering: Fabrication and characterization

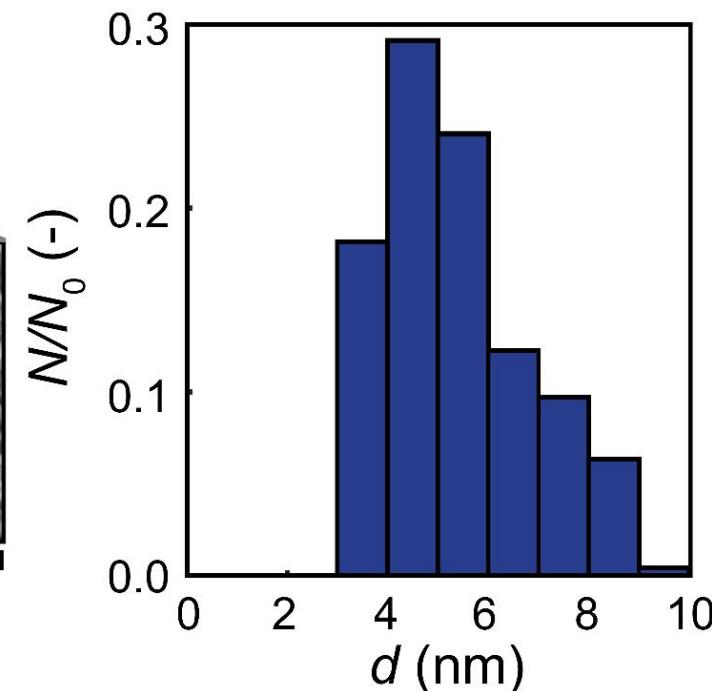


Fabrication and optical characterization

Sputter deposition: Au and TiO_2 .



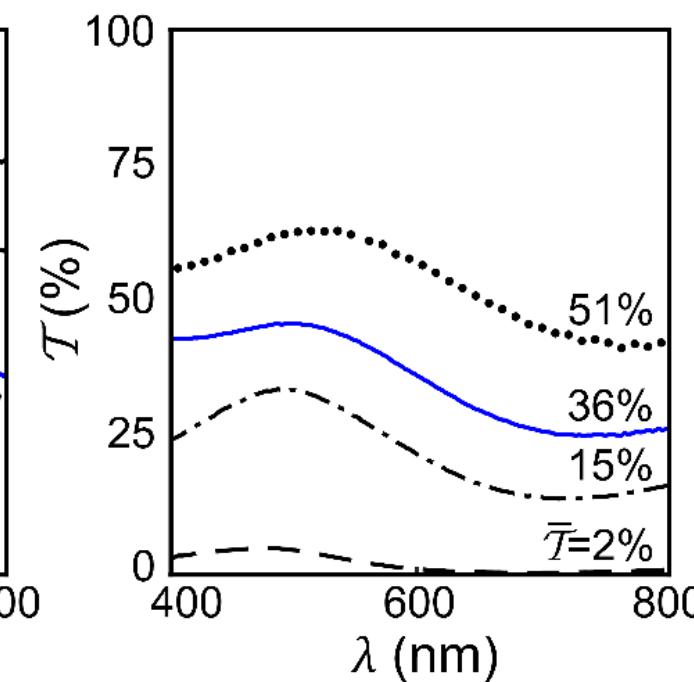
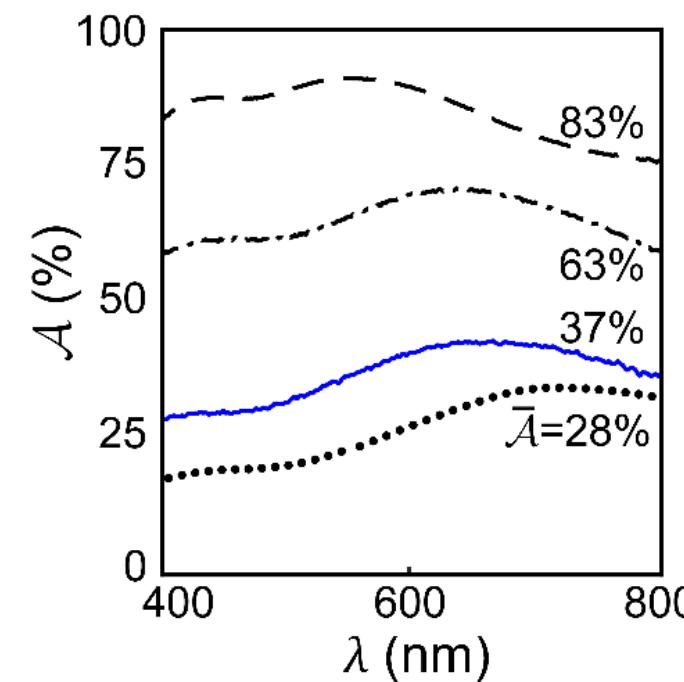
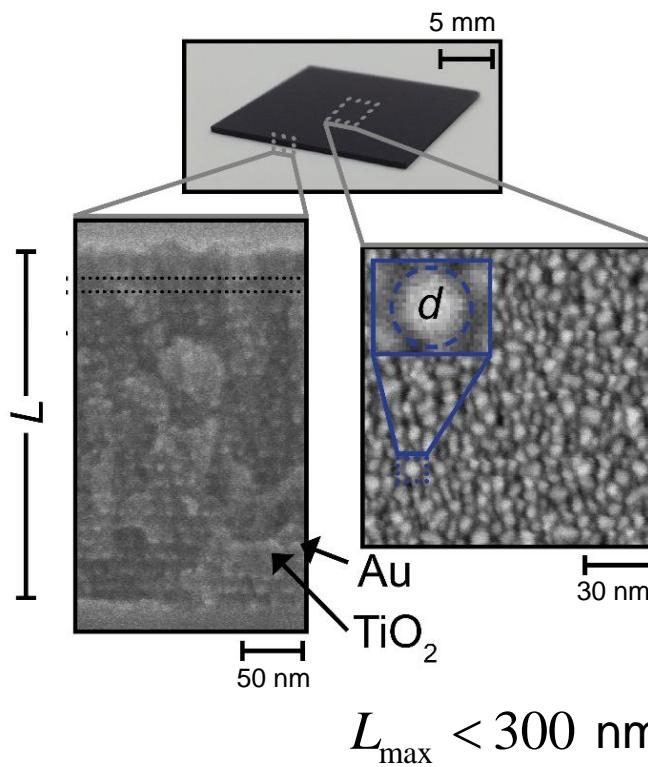
d : nanoparticle diameter



Fabrication and optical characterization

Sputter deposition: Au and TiO_2 .

... 40 nm
— 60 nm
- - 100 nm
- - - 270 nm

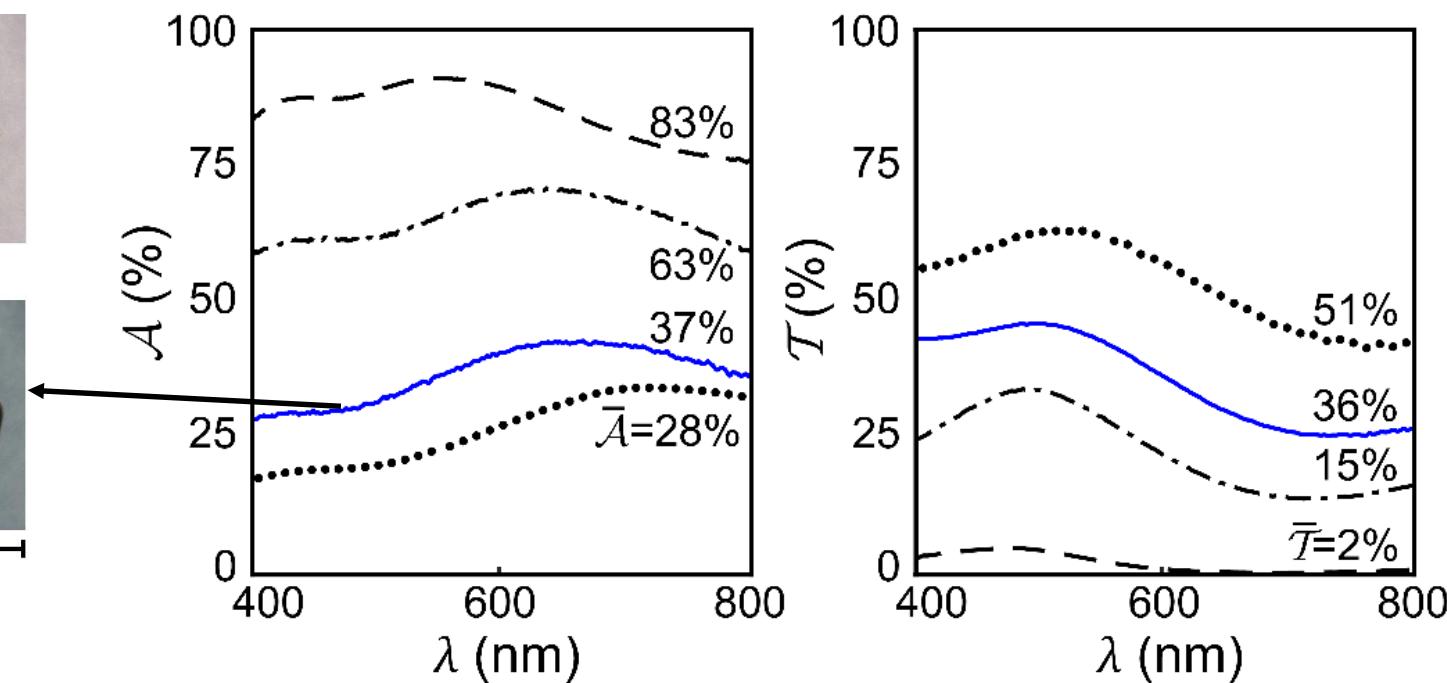
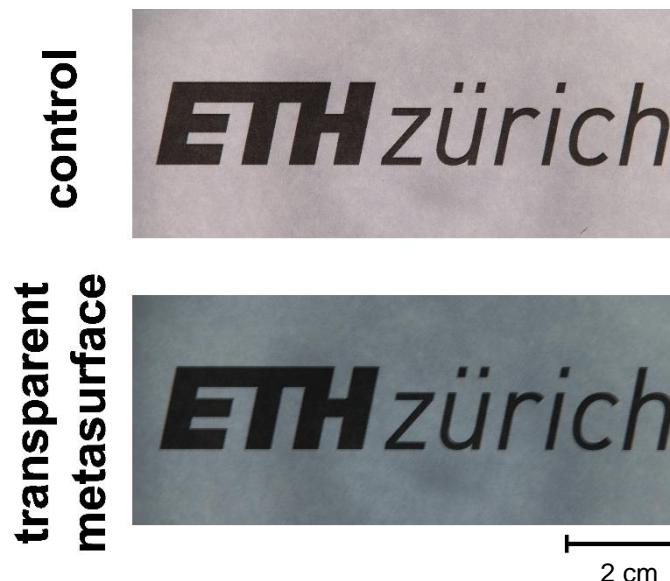


A : light absorption
 T : light transmission

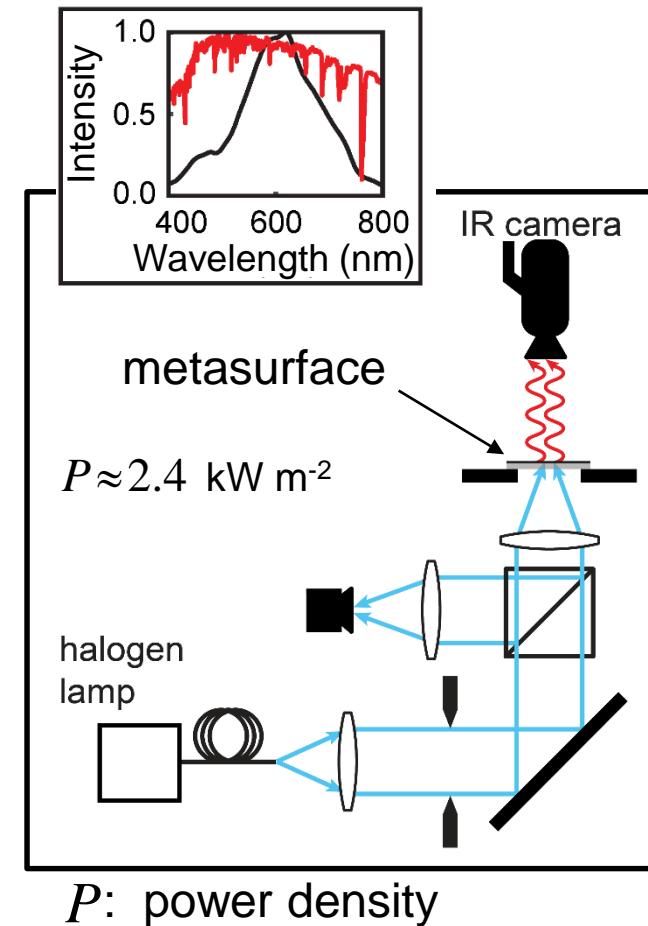
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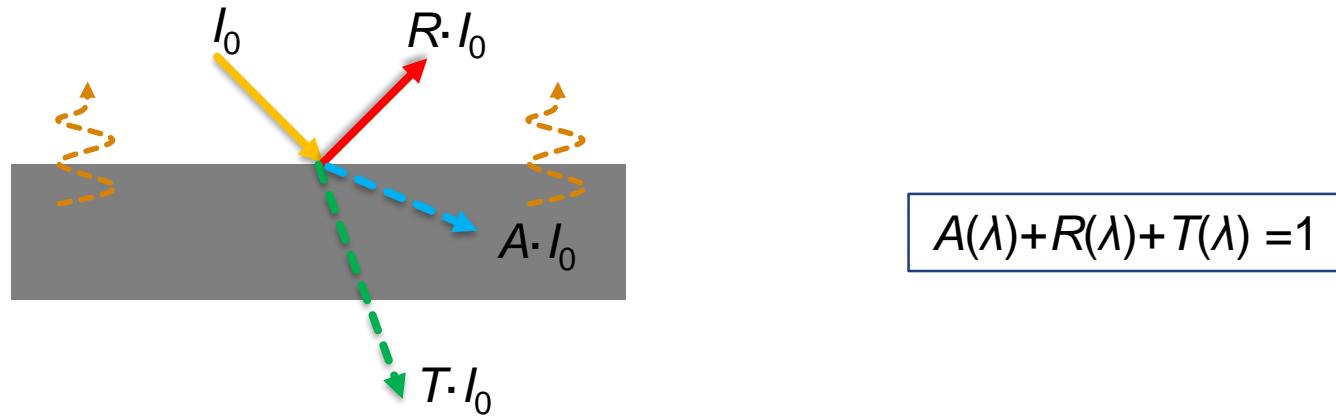


Light-induced heating and characterization with high-speed infrared imaging



Infrared imaging: Basic principles

□ Total radiation



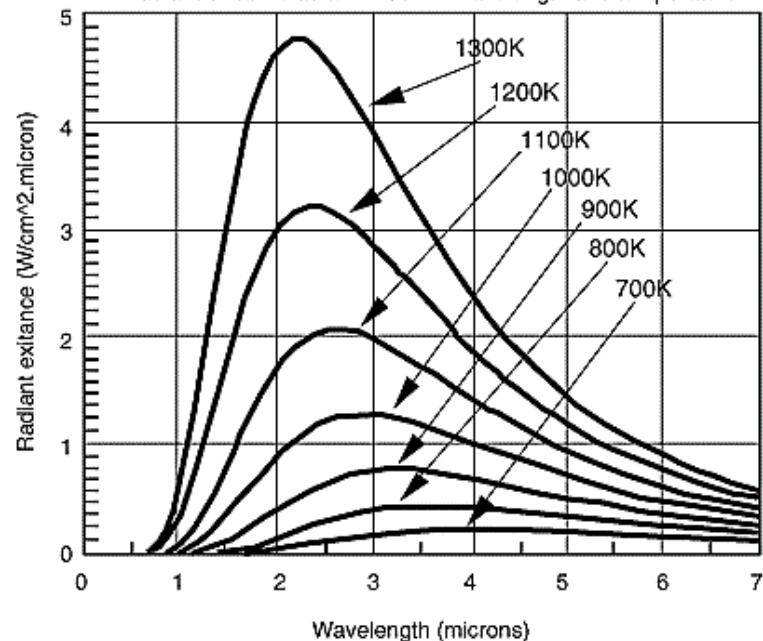
- For object temperatures <500 °C, thermal radiation lies completely in the infrared.

Infrared imaging: Basic principles

□ Thermal emissivity (ε)

The effectiveness of emitting energy as thermal radiation.

$$\varepsilon(\lambda) = A(\lambda)$$



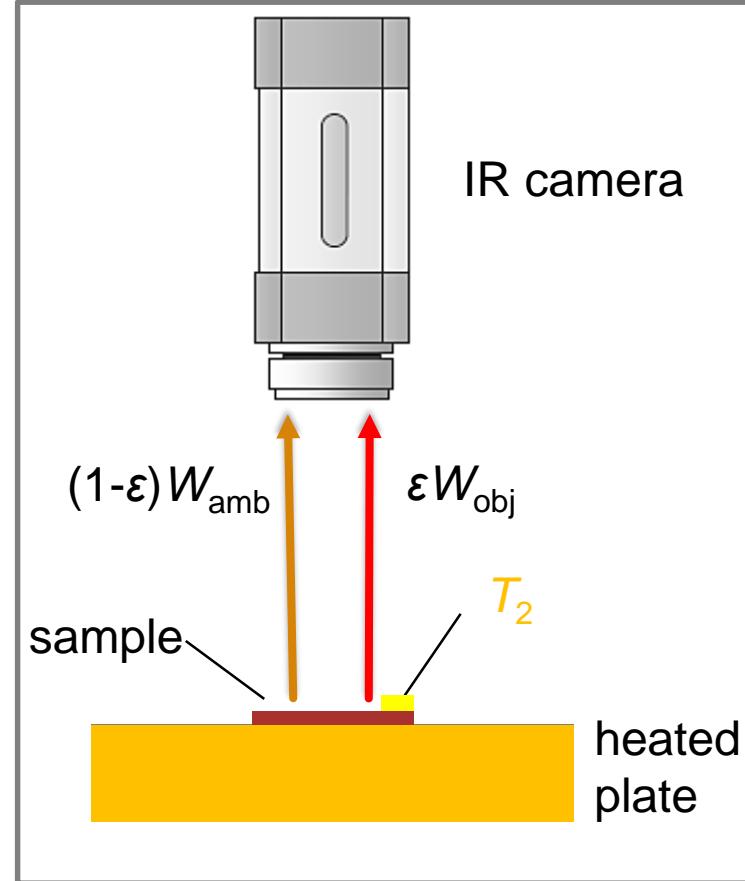
- The maximum emittance wavelength is inversely proportional to temperature.

$$\lambda_{\max} = 2898/T$$

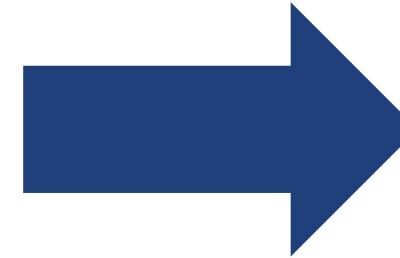
T : temperature [K]

Infrared imaging: Basic principles

□ Calibration

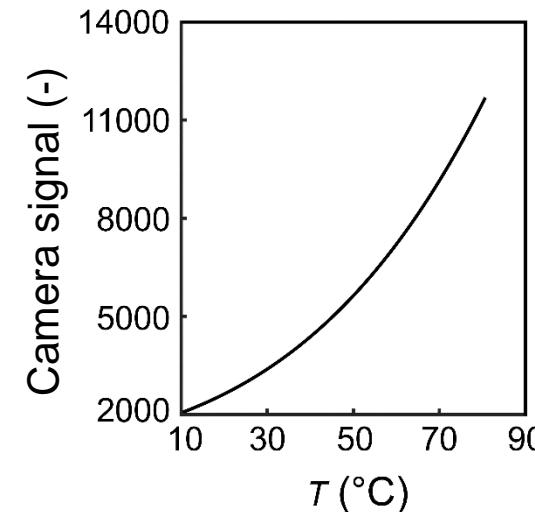


low ε enclosure

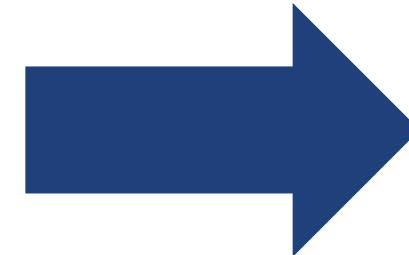


Calculate ε

Infrared imaging: Basic principles

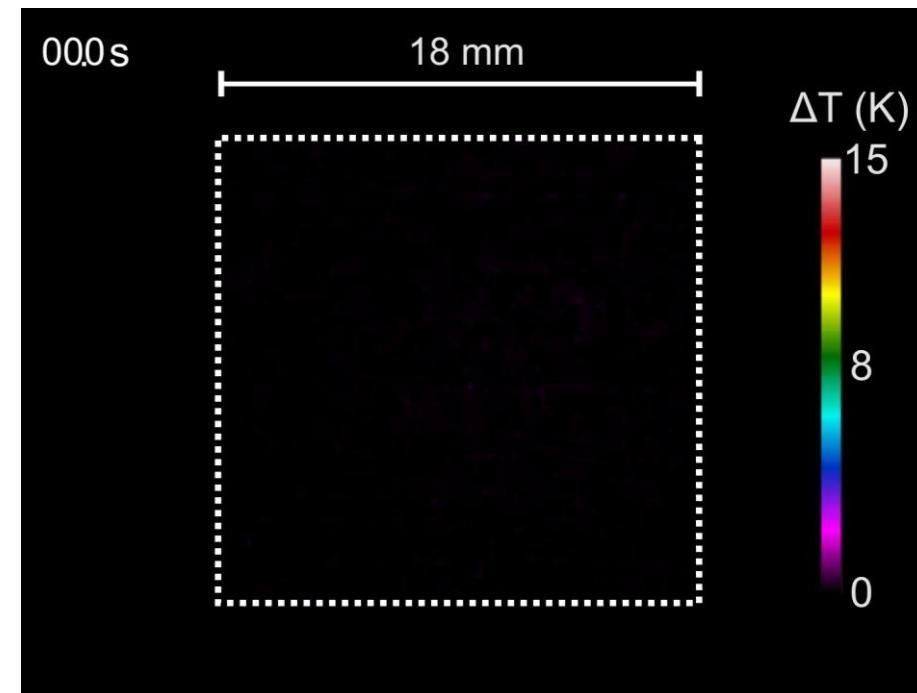
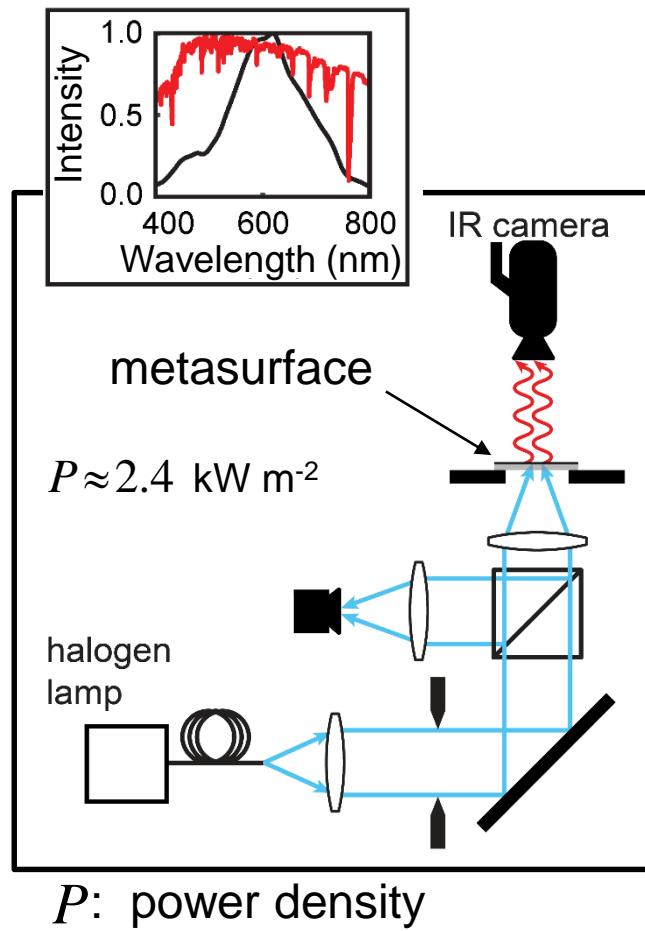


Calculate ϵ



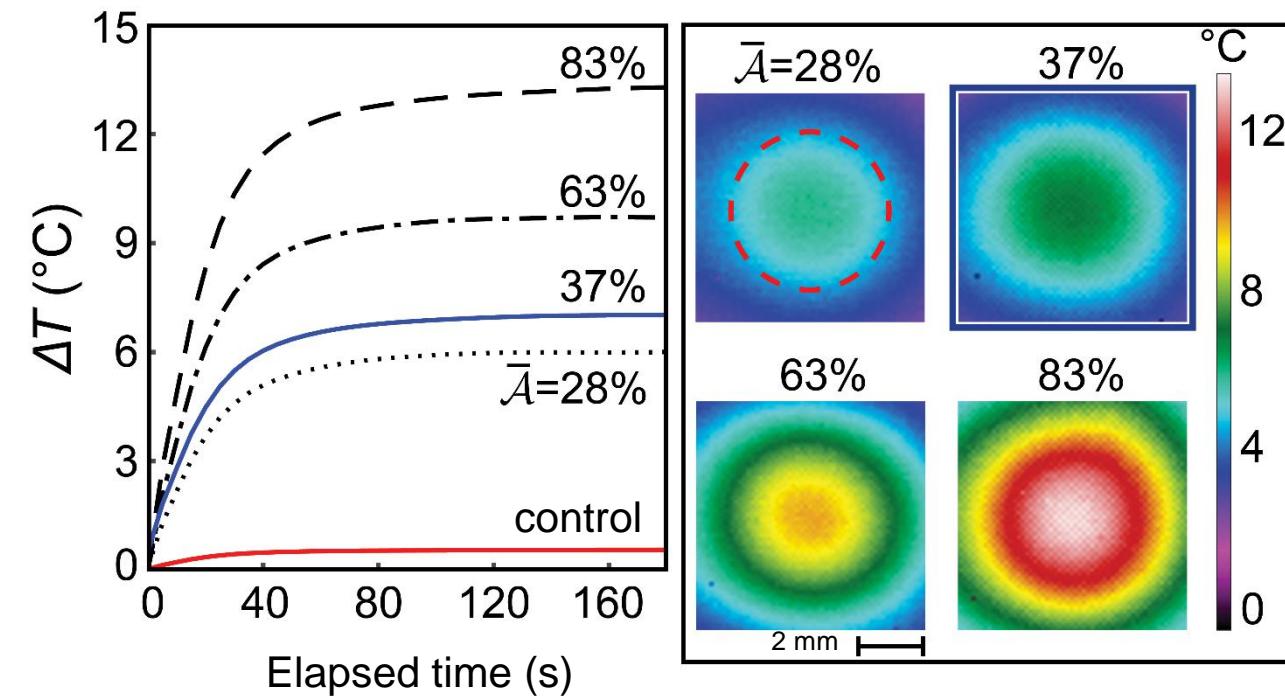
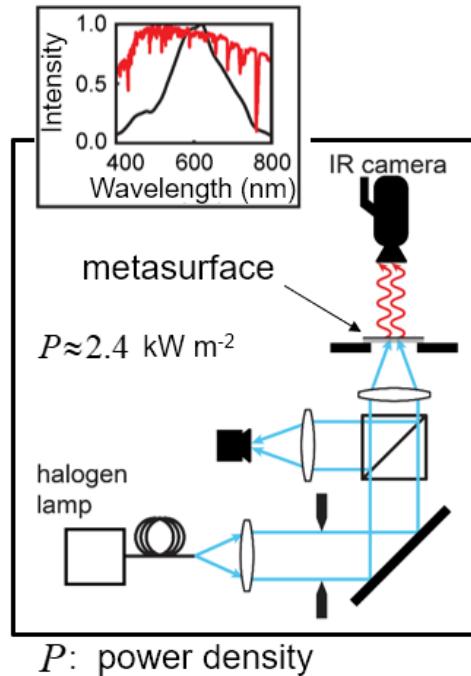
Calculate temperature

Light-induced heating



P : power density

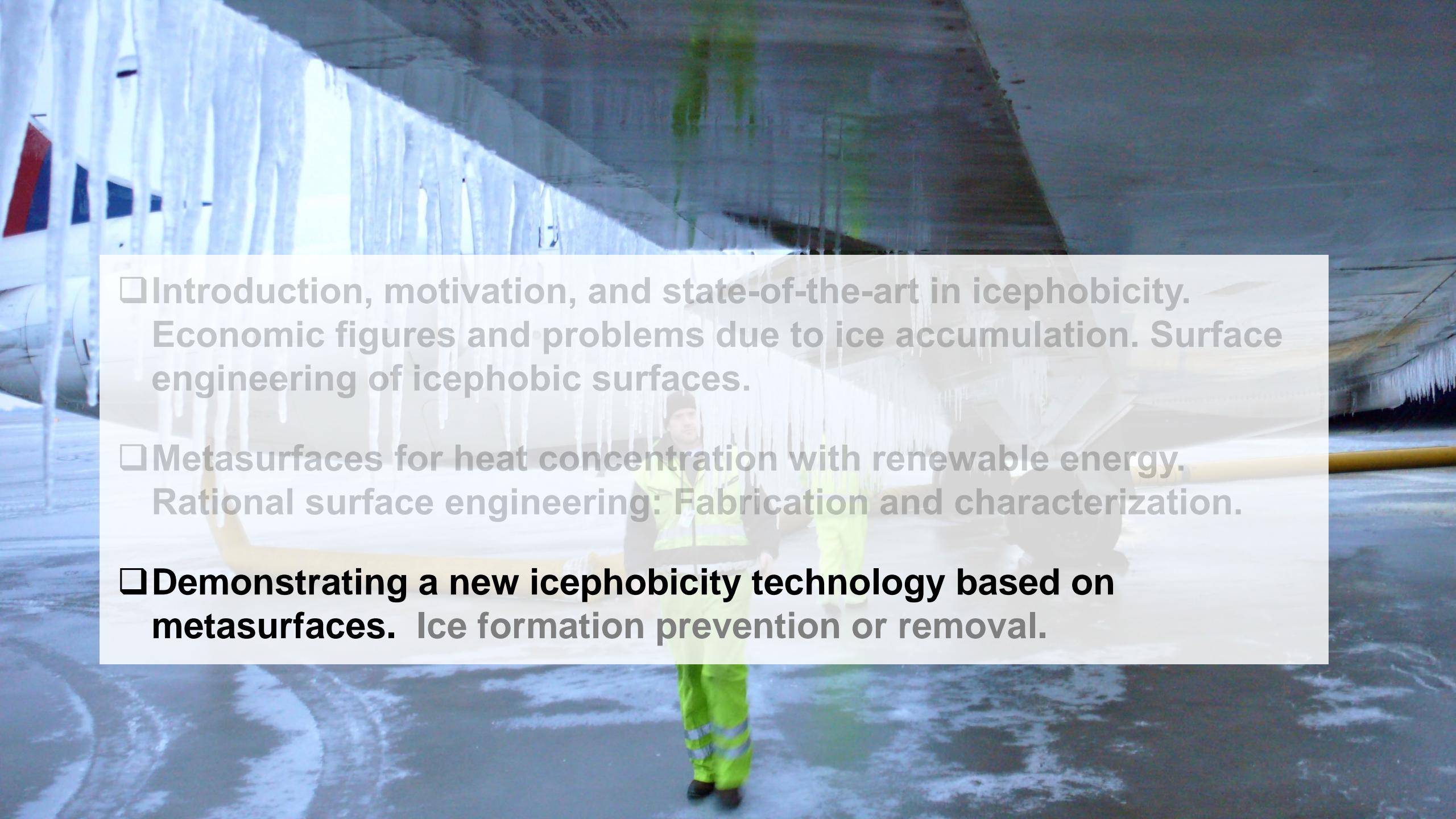
Light-induced heating



ΔT : temperature increase

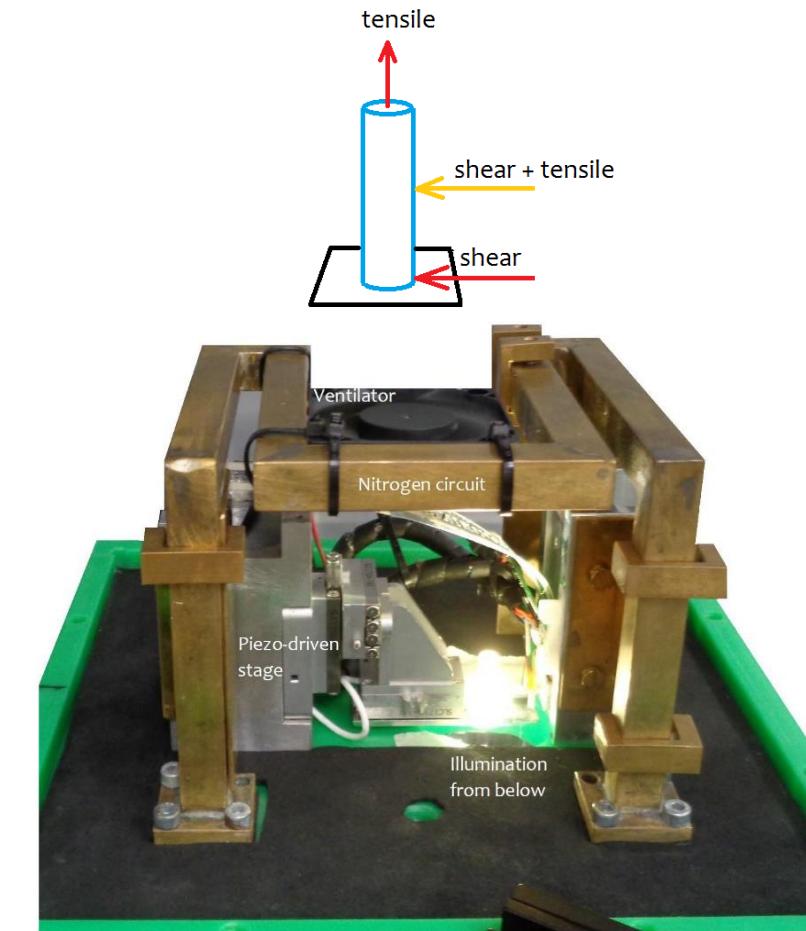
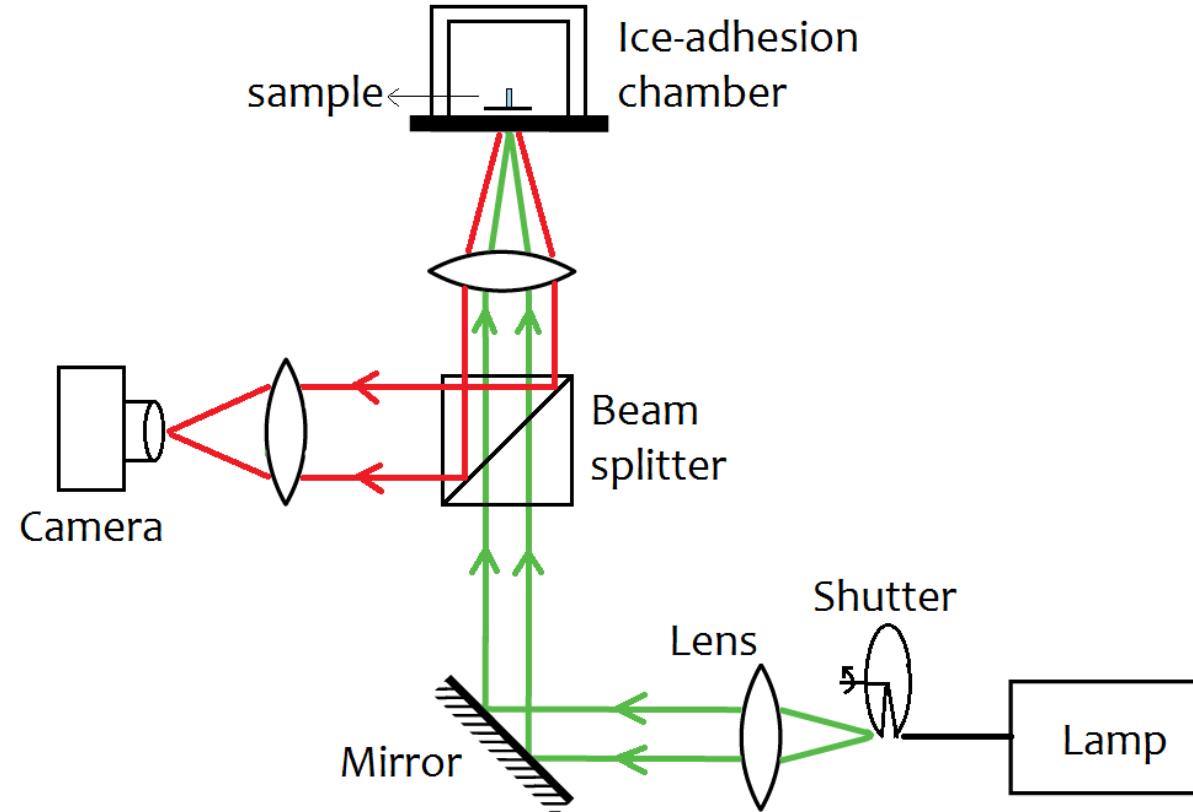
Mitridis, E.; et al. Under review (2018)

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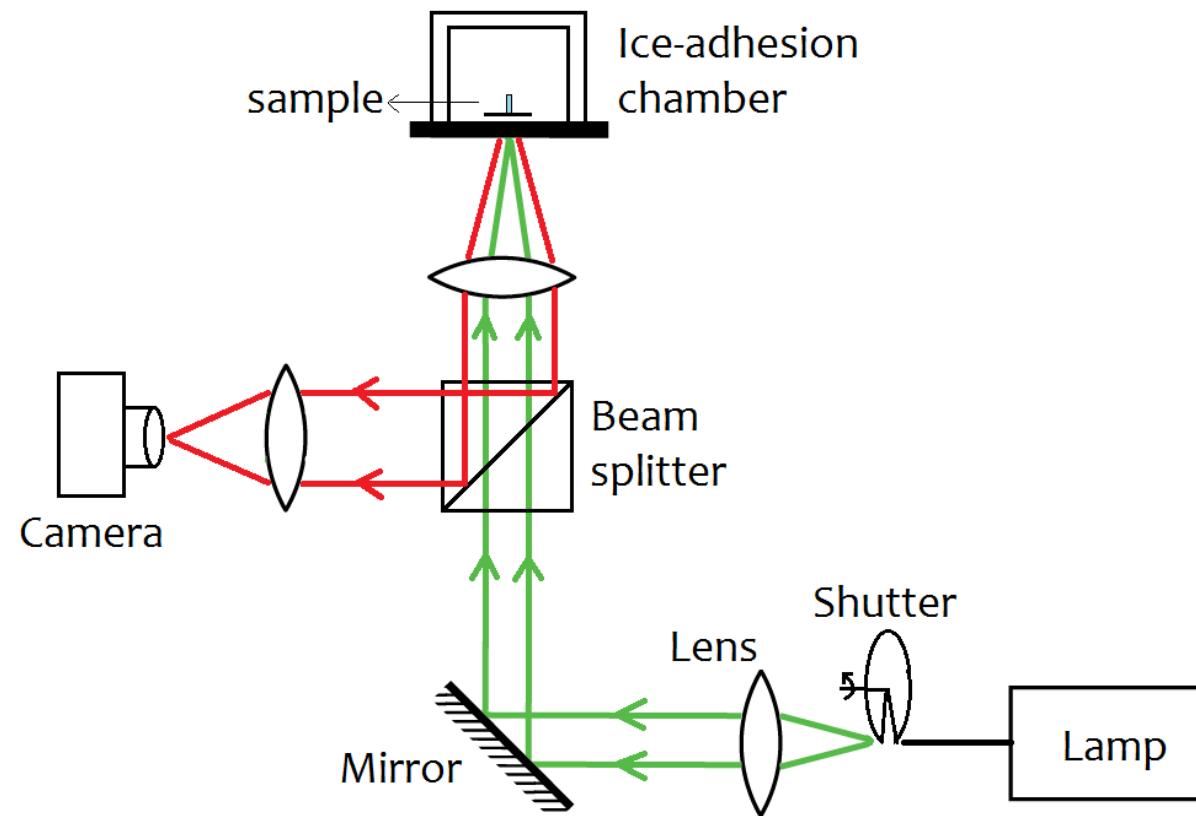
De-icing, anti-icing and defrosting

□ Ice adhesion measurements under visible light

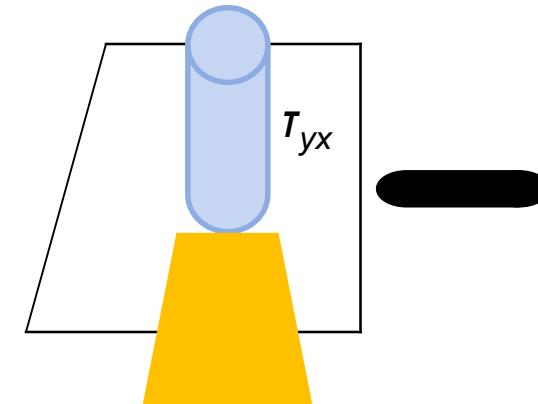
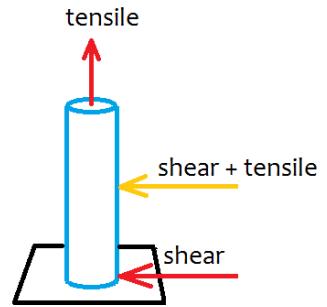


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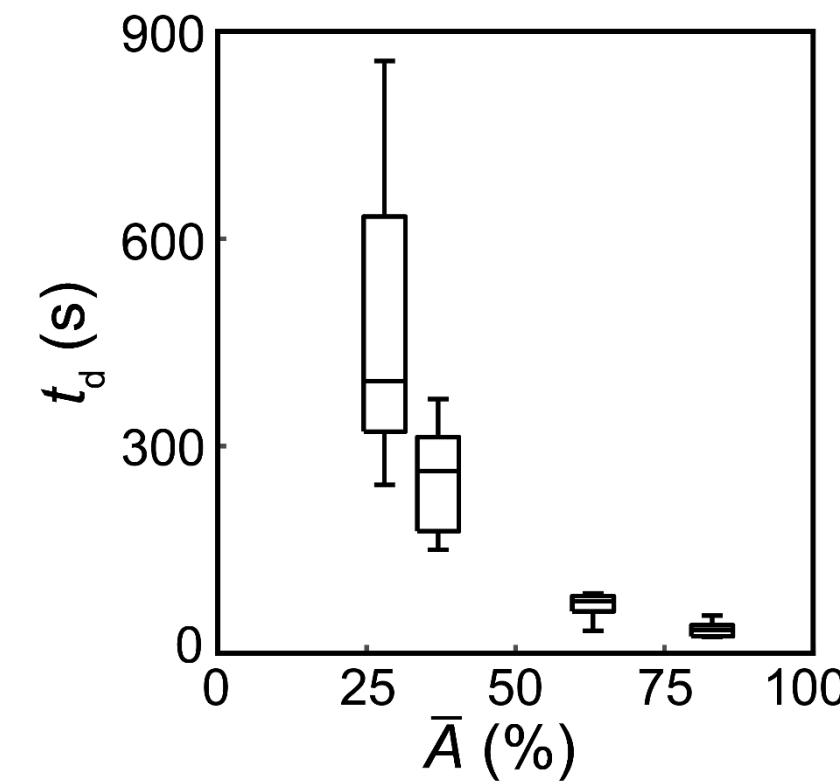
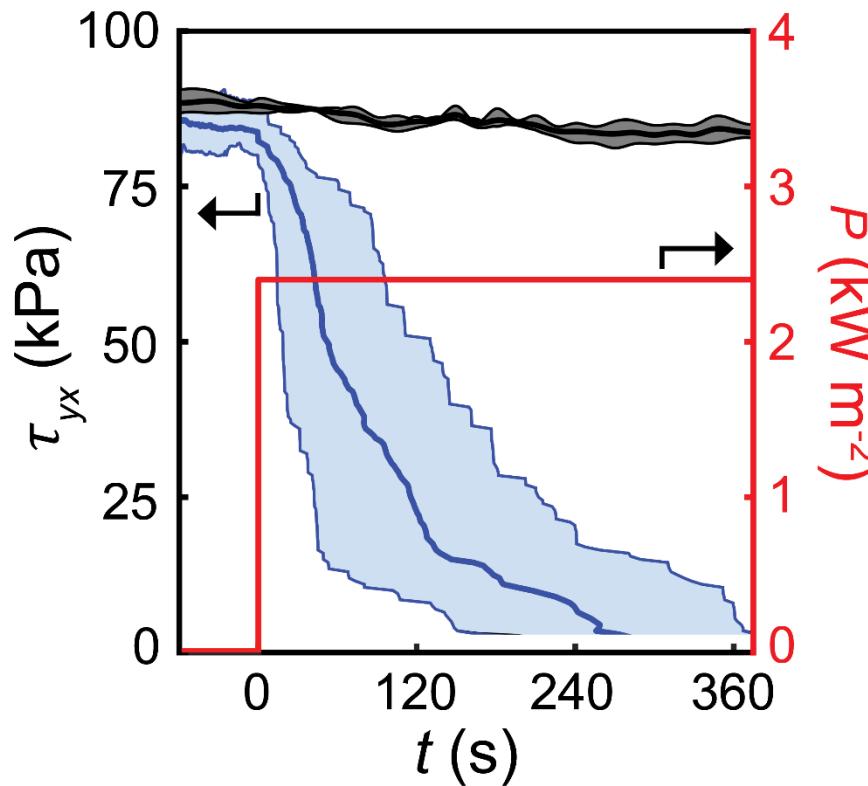


T_{yx} : shear stress



De-icing, anti-icing and defrosting

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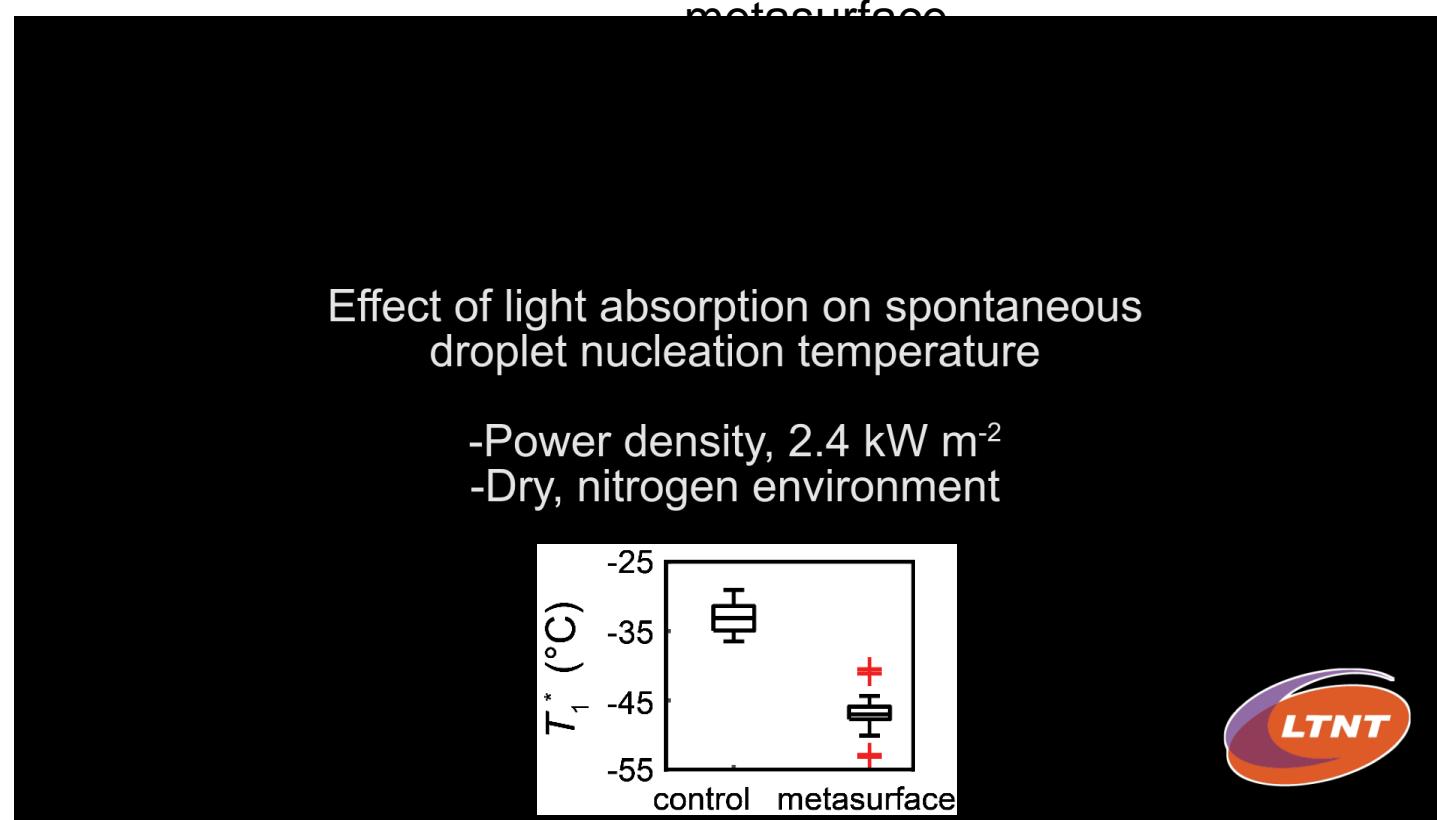


τ_{yx} : shear stress

t_d : de-icing time

De-icing, anti-icing and defrosting

□ Light-induced anti-icing experiments with a single water droplet

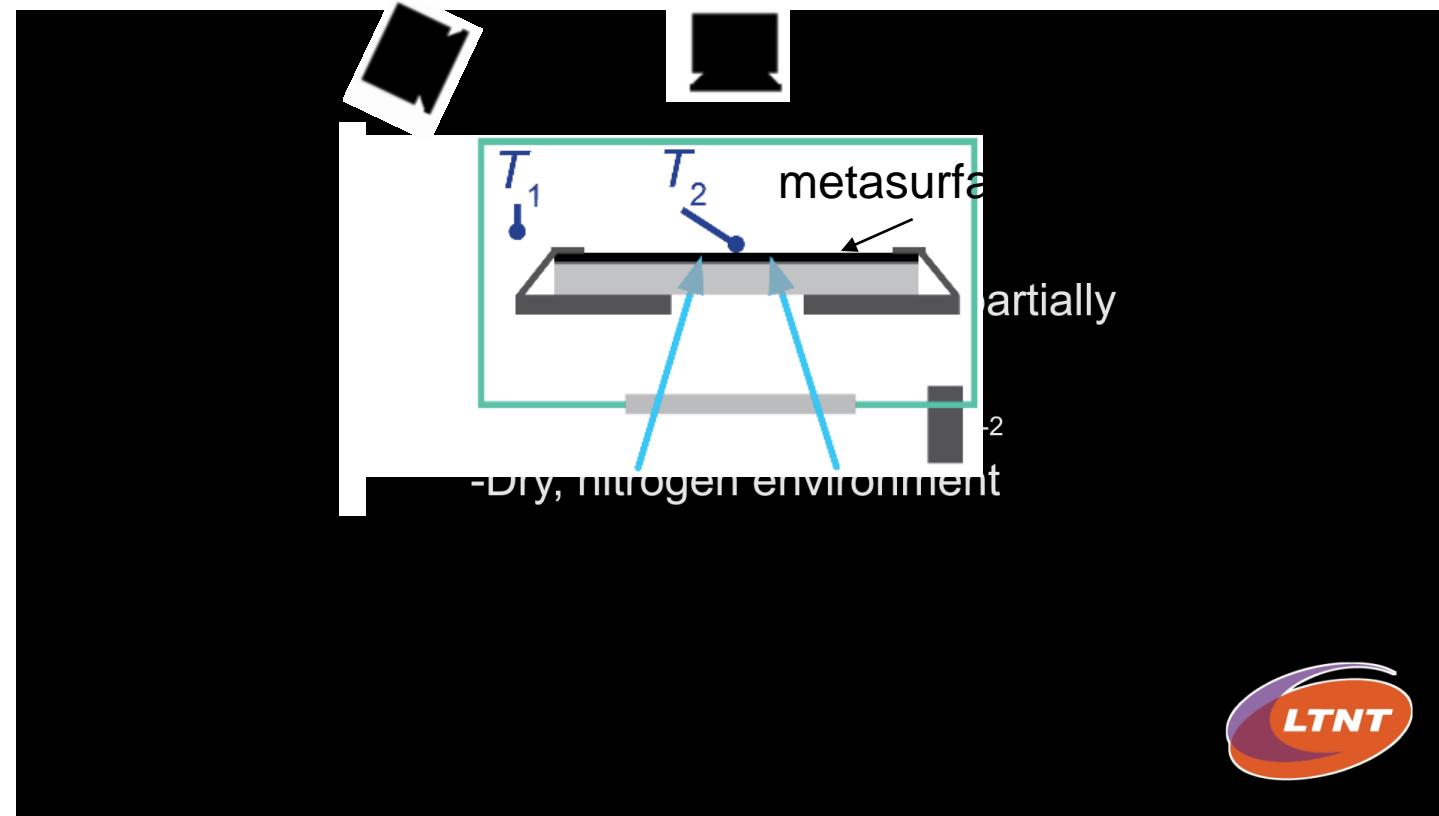


T_1 : gas temperature

T_2 : surface temperature

De-icing, anti-icing and defrosting

□ Light-induced defrosting experiments

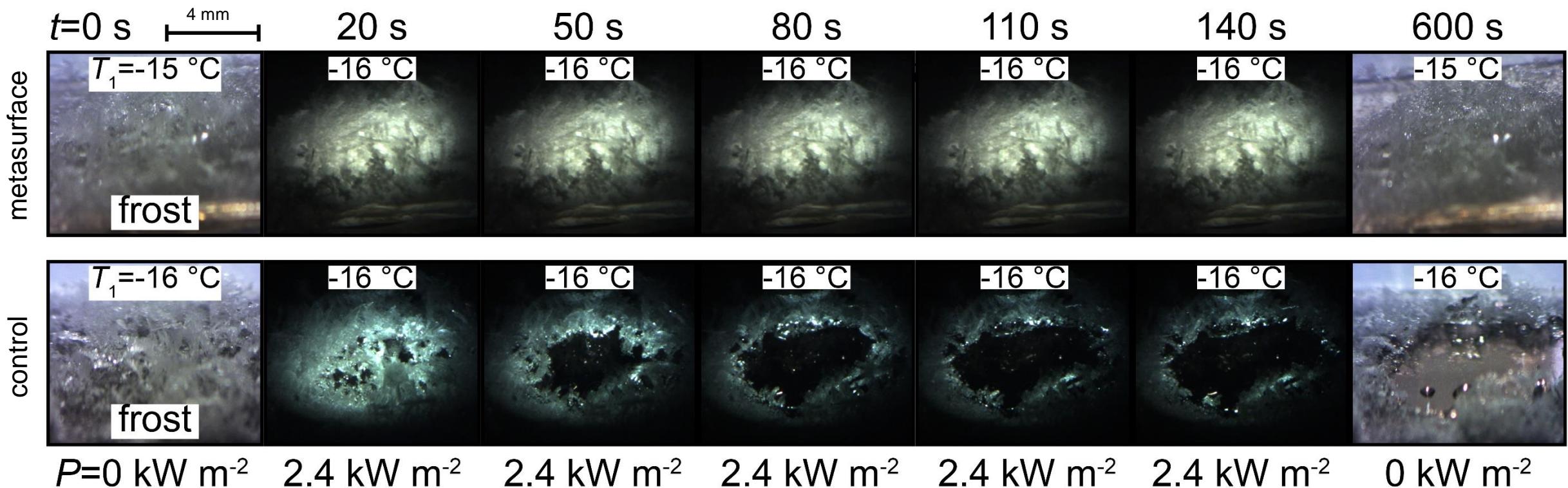


T_1 : gas temperature

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De-icing, anti-icing and defrosting

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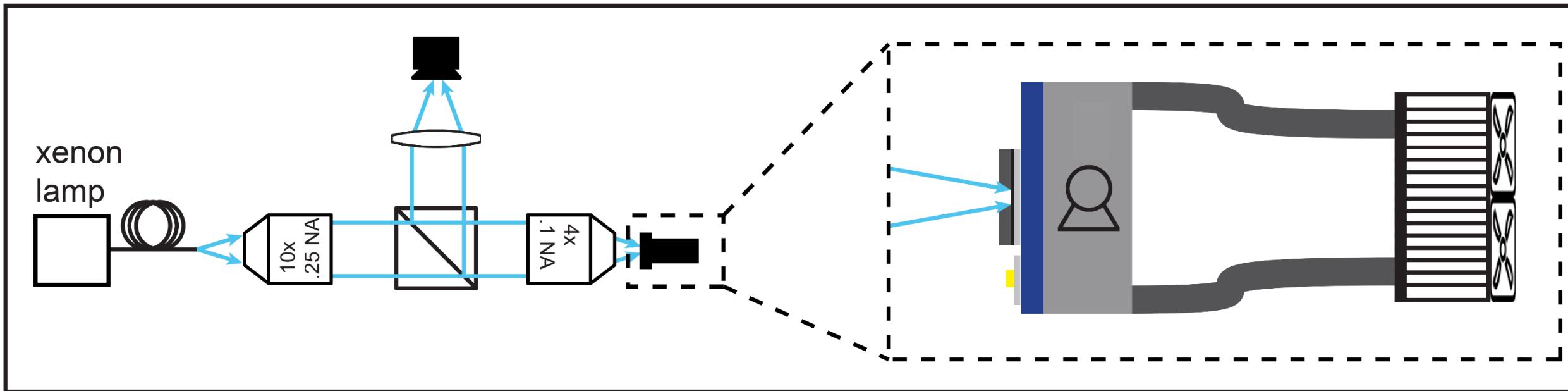
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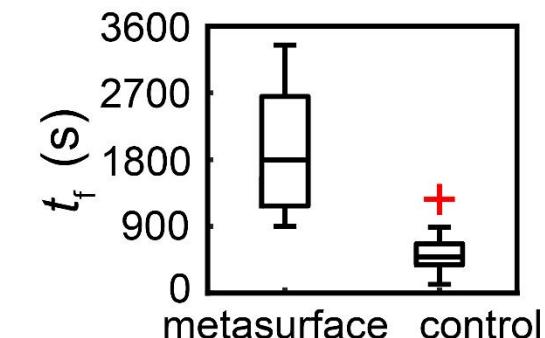
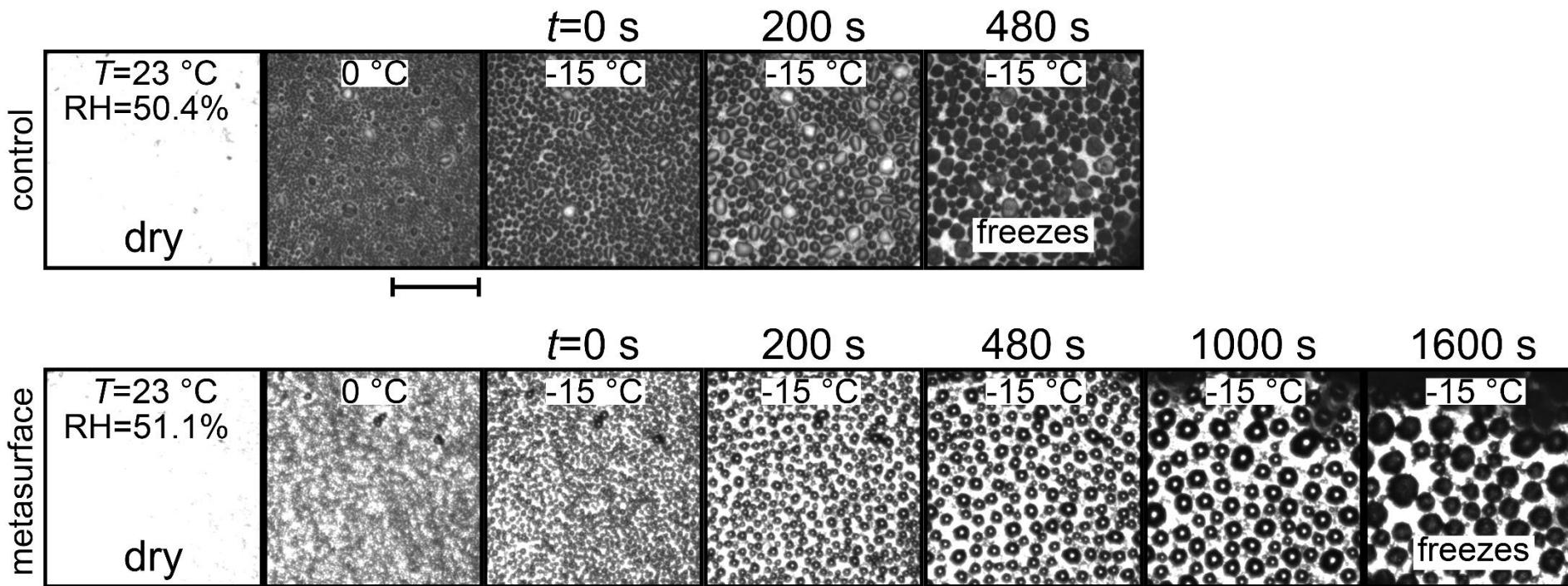
De-icing, anti-icing and defrosting

□ Light-induced anti-icing under one sun



De-icing, anti-icing and defrosting

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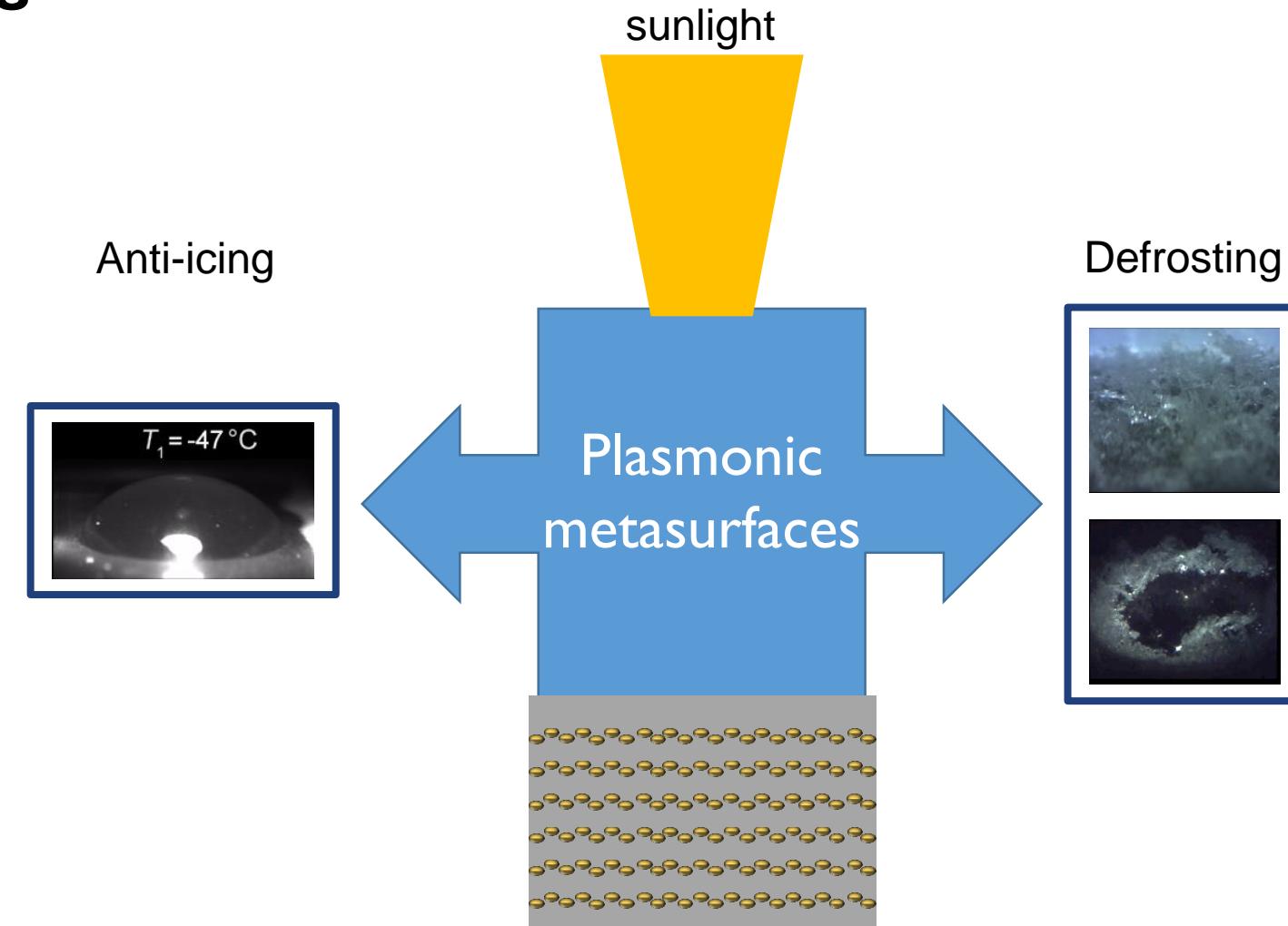
T : surface temperature

t_f : freezing time

Conclusions and outlook



Conclusions



Conclusions

- ❑ Ultra-thin, easy-to-fabricate, metamaterial coatings provide broadband light absorption.
- ❑ Rationally tunable absorption: samples can be transparent or black.
- ❑ The exhibited temperature increase is enough to melt ice and prevent its formation.
- ❑ Trade-off between level of absorption and performance.



jansen.com

Applications:

- ✓ Windows and windshields
- ✓ Sun roofs

...and more applications where a degree of optical transparency is required



solarbuldingtech.com

The future

Facts:

- ✓ **Superhydrophobicity** enhances the anti-icing performance.
- ✓ Superhydrophobicity: requires **micro-/nano-texture** and suitable surface **chemistry**.
- ✓ Minimize thermal conductivity of the substrate, or resistance to heat transfer.

The future

Facts:

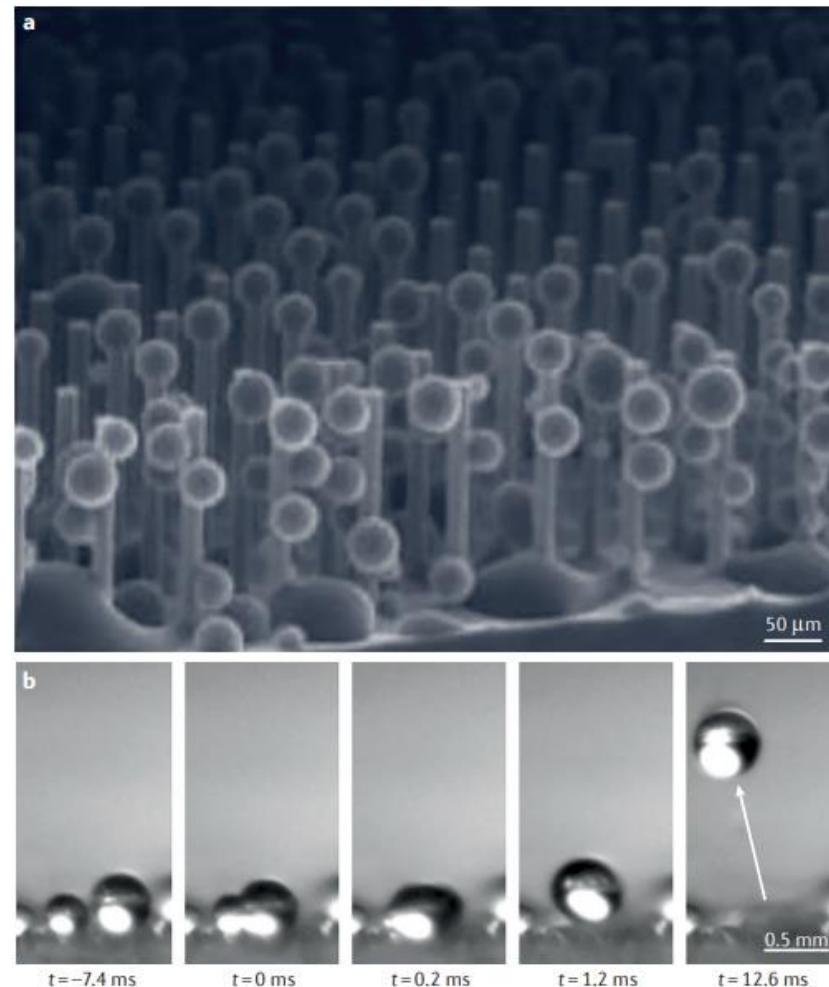
- ✓ **Superhydrophobicity** enhances the anti-icing performance.
- ✓ Superhydrophobicity: requires **micro-/nano-texture** and suitable surface **chemistry**.
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The future

- ✓ **Superhydrophobicity** enhances the anti-icing performance. [1],[2]
- ✓ Superhydrophobicity: requires **micro/nano-texture** and suitable **chemistry**.

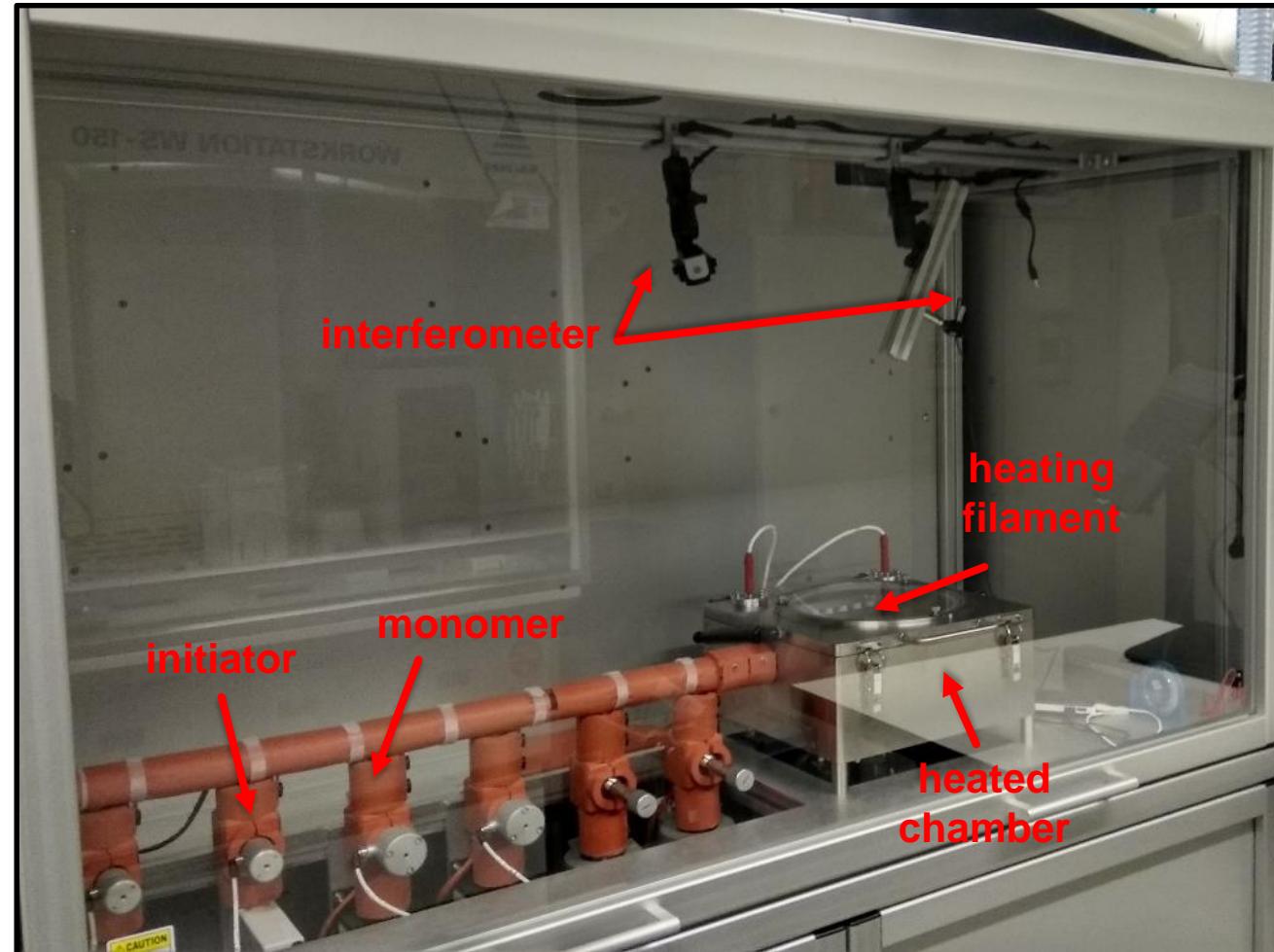
Micro/nano-texture: cleanroom or wet chemistry techniques

Chemistry: low surface energy coating: polymer



Initiated chemical vapor deposition (iCVD)

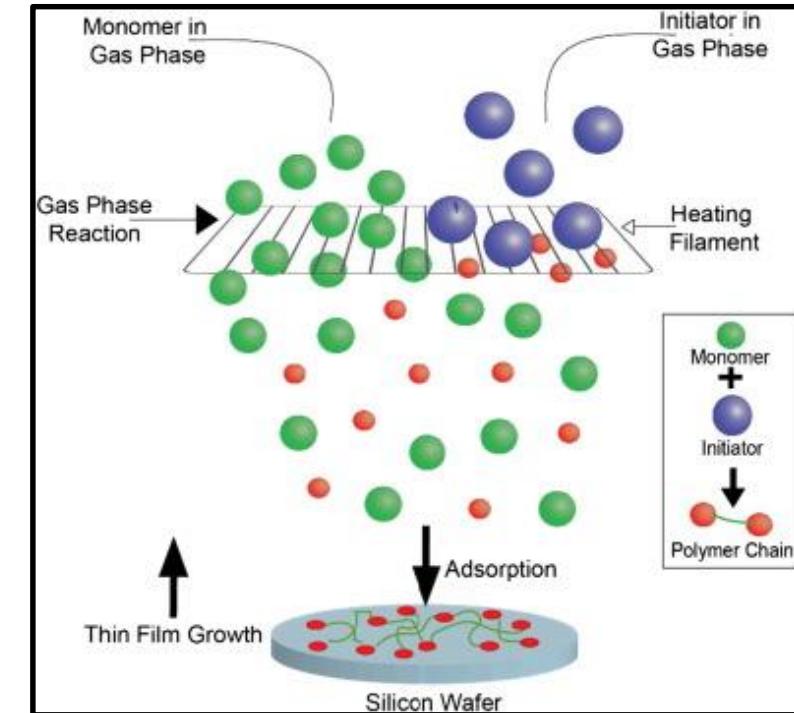
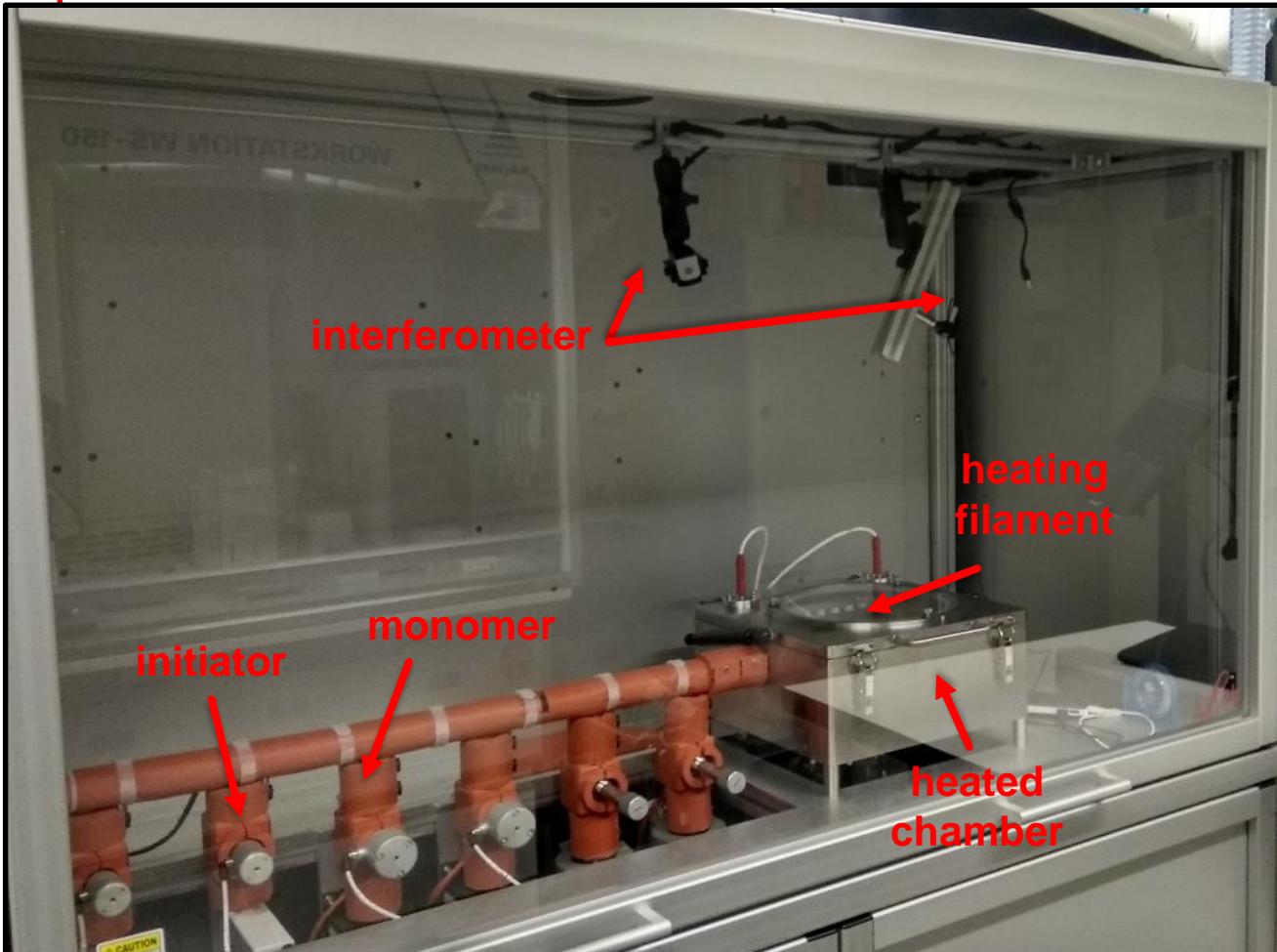
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Initiated chemical vapor deposition (iCVD)

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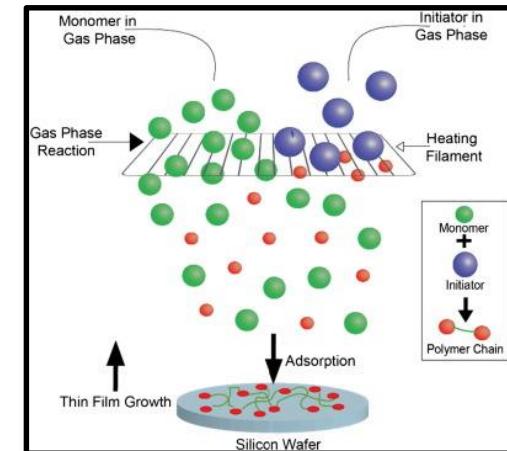
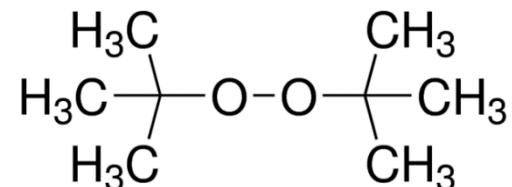
Initiator: **tert-Butyl peroxide (TBPO)**

Monomer: **1H,1H,2H,2H-Perfluorodecyl acrylate (PFDA)**

Initiated chemical vapor deposition (iCVD)

Chemistry: low surface energy coating: polymer

Initiator: *tert*-Butyl peroxide (**TBPO**)



- Used as initiator in free radical polymerization.

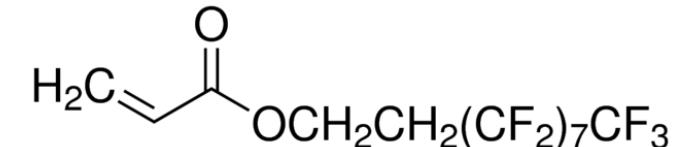


Monomer: 1*H*,1*H*,2*H*,2*H*-Perfluorodecyl acrylate (**PFDA**)

- Has one C=C bond that will be attacked by the free radical.
- Can lead to low surface energy hydrophobic polymer coatings.



18 mN/m



9.3 mN/m

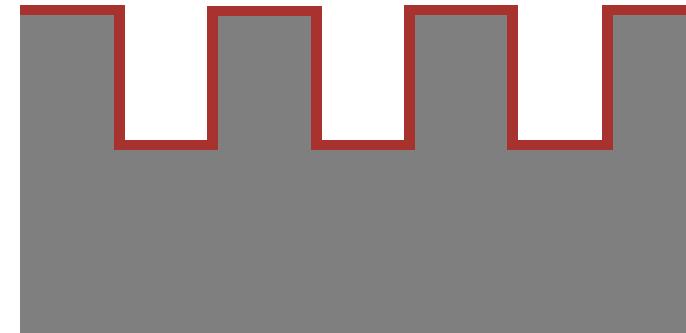
Initiated chemical vapor deposition (iCVD)

Chemistry:

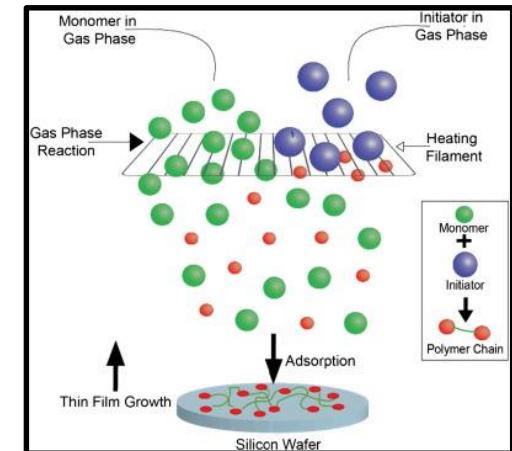
low surface energy coating: polymer



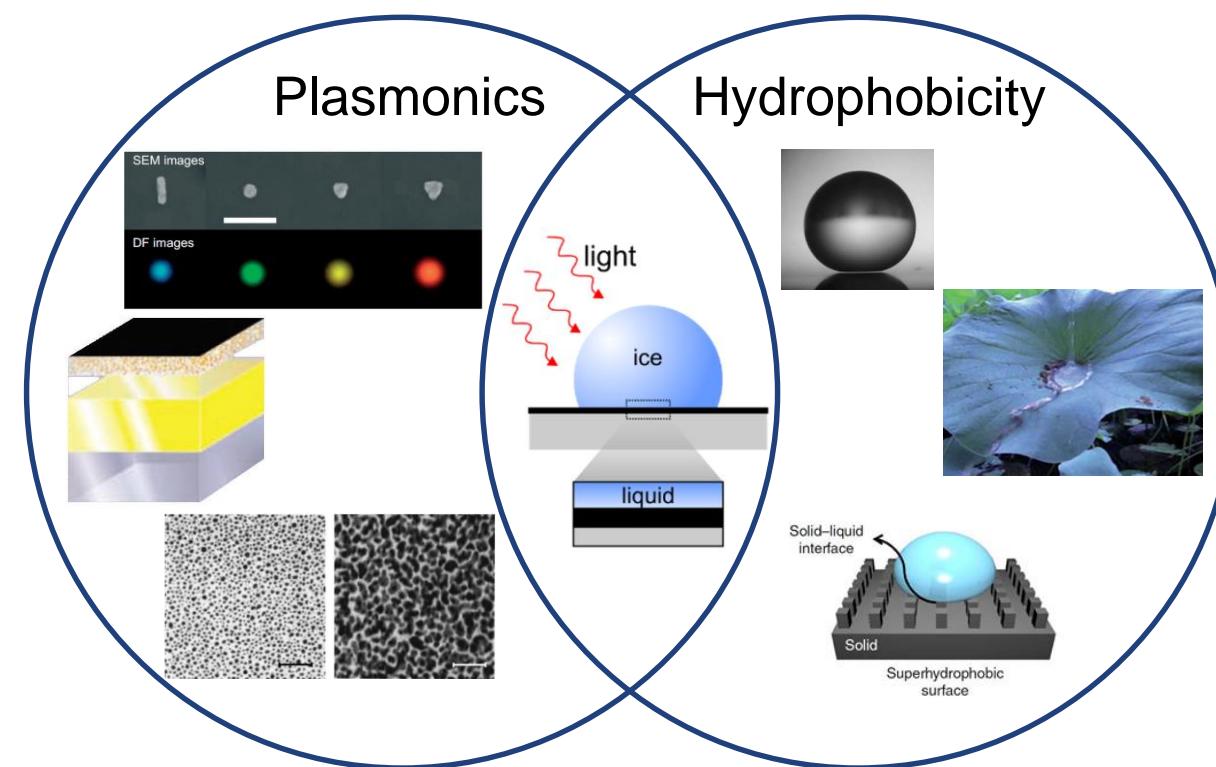
before deposition



after deposition



Hybrid icephobicity





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